

1 Energy conservation via hydrogen cycling in the methanogenic archaeon

2 *Methanosarcina barkeri*

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10 Running Head: Energy conservation via H₂ cycling in *M. barkeri*

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20 **Abstract**

21 Energy conservation via hydrogen cycling, which generates proton motive force
22 by intracellular H₂ production coupled to extracellular consumption, has been
23 controversial since it was first proposed in 1981. It was hypothesized that the
24 methanogenic archaeon *Methanosarcina barkeri* is capable of energy conservation via
25 H₂ cycling, based on genetic data that suggest H₂ is a preferred, but non-essential,
26 intermediate in the electron transport chain of this organism. Here, we characterize a
27 series of hydrogenase mutants to provide direct evidence of H₂ cycling. *M. barkeri*
28 produces H₂ during growth on methanol, a phenotype that is lost upon mutation of the
29 cytoplasmic hydrogenase encoded by *frhADGB*, although low levels of H₂, attributable
30 to the Ech hydrogenase, accumulate during stationary phase. In contrast, mutations that
31 conditionally inactivate the extracellular Vht hydrogenase are lethal when expression of
32 the *vhtGACD* operon is repressed. Under these conditions H₂ accumulates, with
33 concomitant cessation of methane production and subsequent cell lysis, suggesting that
34 the inability to recapture extracellular H₂ is responsible for the lethal phenotype.
35 Consistent with this interpretation, double mutants that lack both Vht and Frh are viable.
36 Thus, when intracellular hydrogen production is abrogated, loss of extracellular H₂
37 consumption is no longer lethal. The common occurrence of both intracellular and
38 extracellular hydrogenases in anaerobic microorganisms suggests that this unusual
39 mechanism of energy conservation may be widespread in nature.

40

41 **Importance**

42 Adenosine triphosphate (ATP) is required by all living organisms to facilitate
43 essential endergonic reactions required for growth and maintenance. Although
44 synthesis of ATP by substrate-level phosphorylation is widespread and significant, most
45 ATP is made via the enzyme ATP synthase, which is energized by transmembrane
46 chemiosmotic gradients. Therefore, establishing this gradient across the membrane is
47 of central importance to sustaining life. Experimental validation of H₂ cycling adds to a
48 short list of mechanisms for generating a transmembrane electrochemical gradient that
49 is likely to be widespread, especially among anaerobic microorganisms.

50

51 **Introduction**

52 An essential requirement for life is the ability to couple exergonic metabolism to
53 the endergonic synthesis of adenosine triphosphate (ATP). While some ATP is made by
54 direct phosphorylation of adenosine diphosphate using “high-energy” metabolites such
55 as phosphoenolpyruvate or 1,3-diphosphoglycerate, the vast majority is produced via
56 the enzyme ATP synthase using energy stored in a transmembrane proton (or sodium)
57 gradient. These electrochemical gradients are typically established during the process
58 of electron transport by membrane proteins that couple exergonic redox reactions to
59 generation of an ion-motive force by one of three general mechanisms: (i) vectorial
60 proton pumping, (ii) scalar movement of protons across the membrane, as in the Q-
61 cycle or Q-loop, or (iii) coupled reactions that consume protons within the cell and
62 produce protons on the outside (1, 2). Given the importance of this process, it is not
63 surprising that this central aspect of living systems has been the subject of intense
64 study (and at least three Nobel Prizes). Indeed, we now possess a detailed, molecular-

65 level understanding of chemiosmotic energy conservation as it applies to
66 photosynthesis and aerobic respiration in a wide variety of organisms including
67 eukaryotes, bacteria, and archaea. Nevertheless, unique and sometimes surprising
68 mechanisms for generation of chemiosmotic gradients continue to be found, including
69 sodium-pumping methyltransferases in methanogenic archaea (3), electrogenic
70 formate:oxalate antiporters in bacteria (4, 5), and light-driven, proton-pumping
71 rhodopsins (6).

72 A controversial, and as yet unproven, mechanism for creating transmembrane
73 proton gradients called H₂ cycling was proposed by Odom and Peck in 1981 to explain
74 ATP synthesis in sulfate-reducing bacteria (7). In this proposed energy-conserving
75 process, protons in the cytosol are reduced to molecular H₂ by enzymes known as
76 hydrogenases. The H₂ so produced then diffuses across the membrane where it is re-
77 oxidized by extracellular hydrogenases, releasing protons that contribute to a trans-
78 membrane proton gradient that can be used to make ATP. The electrons produced by
79 this reaction are returned to the cytoplasm via a membrane-bound electron transport
80 chain, completing the redox process.

81 Although H₂ cycling has been suggested to occur in a number of anaerobic
82 organisms (7-11), the hydrogen cycling hypothesis has not been widely accepted. A key
83 argument against the idea is based on the high diffusion rate of molecular hydrogen.
84 Thus, unless extracellular recapture is exceptionally efficient, hydrogen produced in the
85 cytoplasm would be easily lost, resulting in redox imbalance and presumably cell death.
86 Nevertheless, experimental demonstration of simultaneous production and consumption
87 of H₂ by *Desulfovibrio* supports the model (12), as does metabolic modeling (13).

88 However, other data are inconsistent with the idea, including the ability of hydrogenase
89 mutants to grow on lactate (14) and the inability of high external H₂ pressures to inhibit
90 substrate catabolism (15). Thus, the H₂ cycling model for energy conservation remains
91 unproven.

92 Based on series of genetic experiments, we proposed that the methanogenic
93 archaeon, *Methanosarcina barkeri*, employs H₂ cycling during growth on one-carbon (C-
94 1) substrates and acetate (16, 17). During growth on C-1 compounds such as methanol,
95 the putative cycling pathway would produce H₂ using the cytoplasmic F420-dependent
96 (Frh) and energy-converting ferredoxin-dependent (Ech) hydrogenases, while H₂
97 production during growth on acetate would be mediated solely by Ech. Both pathways
98 would converge on the methanophenazine-dependent hydrogenase (Vht), which is
99 thought to have an active site on the outer face of the cell membrane (18), to consume
100 extracellular H₂ and deliver electrons to the membrane-bound electron transport chain,
101 where they serve to reduce the coenzyme M-coenzyme B heterodisulfide (CoM-S-S-
102 CoB) produced during the production of methane (Fig 1). However, these genetic
103 studies remain incomplete because neither the role of Vht, nor the production and
104 consumption of hydrogen were examined. Here we explicitly test both, providing strong
105 experimental support for the role of H₂ cycling in energy conservation in *M. barkeri*.

106

107 **Results and Discussion**

108 **Hydrogenases of *M. barkeri*.** Three distinct types of hydrogenases are encoded
109 by *M. barkeri* Fusaro (Fig. S1) (19). The F420-reducing hydrogenase (Frh) is a
110 cytoplasmic, 3-subunit (α , β , and γ) enzyme encoded by the *frhADGB* operon, which

111 also includes a maturation protease, FrhD (20). This enzyme couples the
112 oxidation/reduction of the deazaflavin cofactor F420 with production/consumption of H₂.
113 The membrane-bound Vht hydrogenase utilizes the quinone-like electron carrier,
114 methanophenazine, as a cofactor (21). Like Frh, Vht is a 3-subunit enzyme encoded by
115 a four-gene operon (*vhtGACD*) that includes a maturation protease, VhtD (19). *M.*
116 *barkeri* also encodes homologs of both the *frh* and *vht* operons (the *freAEGB* and
117 *vhxGAC* operons, respectively); however, multiple lines of evidence suggest these
118 genes are incapable of producing active hydrogenases (16, 22). Thus, the presence of
119 these genes has no bearing on the results presented herein. The final hydrogenase
120 encoded by *M. barkeri* is a membrane-bound, energy-converting hydrogenase (Ech),
121 which couples the oxidation/reduction of ferredoxin and H₂ to the
122 production/consumption of a proton motive force (23, 24). Thus, the enzyme can use
123 proton motive force to drive the endergonic reduction of ferredoxin by H₂, which is
124 required for CO₂ reduction during hydrogenotrophic methanogenesis and for
125 biosynthesis during growth by H₂-dependent reduction of C-1 compounds (methyl-
126 reducing methanogenesis). During both methylotrophic and aceticlastic
127 methanogenesis, Ech is believed to couple oxidation of reduced ferredoxin to
128 production of proton motive force and H₂. The hydrogen thus produced would need to
129 be recaptured by Vht in a putative H₂ cycling process that contributes to proton motive
130 force (Fig 1) (17).

131 **The cytoplasmic Frh hydrogenase is responsible for production of H₂**
132 **during growth on methanol.** A number of studies have shown that assorted
133 *Methanosarcina* strains produce H₂ during growth on methylotrophic and aceticlastic

134 substrates (9, 25-30); however, to our knowledge this has never been assessed in *M.*
135 *barkeri* strain Fusaro. To test this, we quantified the accumulation of CH₄ and H₂ during
136 growth on methanol medium (Fig 2). Consistent with the hydrogen-cycling hypothesis,
137 we observed significant H₂ production, which reached a maximum partial pressure of
138 ca. 20 Pa near the end of exponential growth. As expected, the culture also produced
139 substantial levels of methane. As previously observed (16), a mutant lacking Frh
140 (WWM115, Table S1) grew at a slower rate than its isogenic parent, and produced
141 somewhat smaller amounts of methane. Very little H₂ (< 4 Pa) was produced during
142 growth of the Δfrh mutant; however, after growth ceased, the H₂ concentration slowly
143 rose, reaching a maximum level of 7 Pa. Thus, Frh is responsible for most hydrogen
144 production during growth of *M. barkeri* Fusaro on methanol, although, some hydrogen is
145 still produced in the Δfrh mutant. As will be shown below, Ech is probably responsible
146 for the low levels of H₂ seen in the Δfrh mutant.

147 **Vht activity is required for viability of *M. barkeri*.** To investigate the role of Vht
148 during growth of *M. barkeri*, we attempted to delete the *vhtGACD* operon via
149 homologous gene replacement (31, 32). However, despite numerous attempts,
150 including selection on a variety of media, with and without supplementation of potential
151 biosynthetic intermediates, no mutant colonies were obtained. We also attempted to
152 delete the *vht* operon using the markerless deletion method of genetic exchange (33).
153 This method relies on construction of a merodiploid strain with both mutant and wild
154 type alleles. Upon segregation of the merodiploid, 50% of the recombinants are
155 expected to be mutants if there is no selective pressure against the mutant allele.
156 However, if the mutation causes a reduction in growth rate (with lethality being the most

157 extreme case), the probability of obtaining recombinants with the mutant allele is
158 severely reduced. We tested 101 haploid recombinants obtained from a
159 *vhtGACD*⁺/ Δ *vhtGACD* merodiploid; all carried the wild-type *vht* allele. Taken together,
160 these data suggest that the *vhtGACD* operon is critical for normal growth of *M. barkeri*.

161 To test whether Vht is essential, we constructed a mutant in which the *vht* operon
162 was placed under control of a tightly regulated, tetracycline-dependent promoter (34).
163 We then examined the viability of the mutant and its isogenic parent by spotting serial
164 dilutions on a variety of media, with and without tetracycline. As shown in Figure 3, the
165 *P_{tet}::vht* mutant is unable to grow in the absence of the inducer, but grew well when
166 tetracycline was added, whereas the isogenic parent grew with or without the addition of
167 tetracycline. These phenotypes were observed on a variety of media including (i)
168 methanol, (ii) methanol plus H₂, (iii) H₂/CO₂, and (iv) acetate, which were chosen
169 because they encompass growth conditions that require each of the four known
170 methanogenic pathways used by *M. barkeri* (Fig 4). It should be stressed that the
171 *P_{tet}::vht* mutant used in this experiment was pre-grown in the presence of inducer. Thus,
172 at the start of the experiment, all cells have active Vht. However, during cultivation in the
173 absence of tetracycline, pre-existing Vht is depleted by protein turnover and cell
174 division, thereby allowing characterization of the Vht-deficient phenotype. The absence
175 of growth of the diluted cultures in all media shows that Vht is essential for growth via
176 the methylotrophic (methanol), methyl-reducing (methanol plus H₂), hydrogenotrophic
177 (H₂/CO₂) and aceticlastic (acetate) methanogenic pathways.

178 **Depletion of Vht results in H₂ accumulation and cell lysis.** To help
179 understand why Vht is essential, we quantified production of H₂ and CH₄ in cultures of

180 the $P_{tet}::vht$ strain with and without tetracycline (Fig 2). When grown in methanol
181 medium in the presence of tetracycline, the accumulation of H₂ and CH₄ was essentially
182 identical to that of the isogenic parent. Cultures in which *vht* is not expressed (*i.e.*
183 without tetracycline) grew initially, but rapidly slowed, and reached an optical density
184 that was less than half of that obtained when *vht* was expressed. The optical density
185 subsequently dropped, suggesting cell death and lysis. Similarly, methane accumulation
186 in cultures not expressing *vht* was much slower than in induced cultures, and only
187 reached half of that seen under inducing conditions. In contrast, H₂ accumulation was
188 much higher in the absence of Vht, with final levels nearly six-fold higher than those
189 seen in cultures that express Vht. These data clearly show that Vht is required for
190 efficient recapture of H₂ produced by Frh and Ech. Moreover, they suggest that H₂ loss
191 is responsible for the lethal consequences of *vht* repression.

192 **Vht is not essential in Δfrh mutants.** If the inability to recapture H₂ is
193 responsible for the essentiality of Vht, then it should be possible to delete the *vht* operon
194 in strains that do not produce hydrogen. As described above, Frh is responsible for the
195 majority of H₂ production during growth. Thus, we attempted to introduce a Δvht allele
196 into the Δfrh host. In contrast to our prior unsuccessful attempts to create a Δvht single
197 mutant, the $\Delta vht/\Delta frh$ double mutant was isolated in the first attempt. Therefore, Vht is
198 not required when Frh is absent. Like the Δfrh single mutant, the $\Delta vht/\Delta frh$ double
199 mutant grows slowly on methanol and produces lower levels of methane (Fig 2).
200 Significantly, the double mutant does not produce the excessive level of H₂ seen in the
201 uninduced $P_{tet}::vht$ strain, instead accumulating H₂ at levels similar to those of the
202 parental strain (*ca.* 20 Pa). Because Ech is the only active hydrogenase remaining in

203 the $\Delta vht/\Delta frh$ mutant, it must be responsible for H₂ production in this strain. This begs
204 the question of why H₂ accumulation stops at 20 Pa in the double mutant, while the
205 uninduced P_{tet::vht} strain produces much higher levels. We suggest that the coupling of
206 Ech activity to generation of proton motive force thermodynamically restrains excessive
207 H₂ production, even in the absence of H₂ uptake by Vht. This would also explain the
208 viability of the $\Delta vht/\Delta frh$ double mutant. This situation is in stark contrast to that seen in
209 the *vht*-depleted strain, where the F420-dependent Frh is responsible for most of the H₂
210 production (see above). Accordingly, at the low H₂ partial pressures observed in our
211 experiments, reduction of protons with F420 is strongly exergonic, allowing excessive
212 hydrogen accumulation. This is also consistent with the observation that the redox state
213 of F420 is in rapid equilibrium with H₂ (35). Interestingly, the lower amount of H₂
214 accumulation in the Δfrh mutant, relative to that seen in the $\Delta vht/\Delta frh$ mutant, shows that
215 Vht also consumes H₂ produced by Ech. This supports previous studies indicating
216 potential energy conservation via Ech/Vht H₂ cycling during acetate metabolism (17,
217 23).

218 ***M. barkeri* has a bifurcated electron transport chain with H₂-dependent and**
219 **-independent branches.** We previously showed that *M. barkeri* has a branched
220 electron transport chain, with Frh- and F420 dehydrogenase (Fpo)-dependent branches
221 (16). The data reported here extend our understanding of the Frh-dependent branch,
222 and are fully consistent with the model depicted in Fig 1. Thus, during growth on
223 methylotrophic substrates such as methanol, reduced F420 is preferentially oxidized via
224 an energy conserving, H₂ cycling electron transport chain that requires Frh. However, in
225 the absence of Frh, reduced F420 is channeled into the Fpo-dependent electron

226 transport chain, which supports growth at a significantly slower rate (Figs 2 & 4). This
227 alternate pathway accounts for the viability of the Δfrh mutant, which is lost when both
228 *frh* and *fpo* are deleted (16). Similar, but less severe phenotypes have been observed in
229 *fpo* and *frh* mutants of *Methanosarcina mazei*, thus it seems likely that H₂ cycling also
230 occurs in this closely-related species (36). However, many *Methanosarcina* species,
231 especially those that inhabit marine environments, are devoid of hydrogenase activity,
232 despite the presence of hydrogenase encoding genes. We, and others, have interpreted
233 this to be an adaptation to the marine environment, where H₂-utilizing sulfate reducers
234 are likely to disrupt H₂ cycling due to the superior thermodynamics of H₂ oxidation
235 coupled to sulfate reduction (19, 37).

236 A similar branched electron transport chain may also explain the contradictory
237 evidence regarding H₂ cycling in *Desulfovibrio* species. Thus, the viability of
238 *Desulfovibrio* hydrogenase mutants and the inability of excess H₂ to suppress substrate
239 catabolism can both be explained by the presence of alternative electron transport
240 mechanisms. Indeed, metabolic modeling of *D. vulgaris* strongly supports this
241 interpretation (13). Thus, it is critical that experiments designed to test the H₂ cycling
242 mechanism be interpreted within a framework that includes the possibility of branched
243 electron transport chains. With this in mind, it seems likely that many anaerobic
244 organisms might use H₂ cycling for energy conservation. Indeed, since it was originally
245 proposed, H₂ cycling has been suggested to occur in the acetogen *Acetobacterium*
246 *woodii* (10) and in the Fe (III) respiring *Geobacter sulfurreducens* (8).

247 **Why are Vht mutants inviable during growth on methanol/H₂ or H₂/CO₂?**

248 Although the data presented here strongly support the H₂ cycling model, they raise

249 additional questions regarding H₂-dependent methanogenesis that are not easily
250 explained. In particular, it is not readily apparent why the uninduced P_{tet}::*vht* mutants are
251 inviable during hydrogenotrophic or methyl-reducing growth. As shown in Figure 4, it
252 should be possible to channel electrons from H₂ oxidation into the electron transport
253 chain via Frh and Fpo. Indeed, Thauer *et al* have proposed that this alternate pathway
254 is functional in *Methanosarcina* (38). Nevertheless, the P_{tet}::*vht* mutant does not grow
255 under repressing conditions on either H₂/CO₂ or methanol plus H₂. It should be
256 stressed, that we use high concentrations of hydrogen during growth on these
257 substrates. Thus, it is expected that reduction of F420 via Frh should be exergonic in
258 our experiments, which would favor this pathway. (This is in contrast to the
259 methylotrophic or acetoclastic growth conditions described above, under which the
260 reverse reaction (*i.e.* hydrogen production) is favored.) Thus, a thermodynamic
261 argument cannot easily explain the results. Further, based on available evidence (16,
262 39, 40), energy conservation via the Vht-dependent pathway should be identical to that
263 of the alternate Frh/Fpo-dependent pathway. Thus, an energy conservation argument
264 also cannot explain the phenomenon. One might argue that faster kinetics of the Vht-
265 dependent pathway could be responsible, but, in our opinion, the growth (albeit slower
266 than wild-type) of the Δfrh and $\Delta vht/\Delta frh$ mutants during methylotrophic growth, which
267 depends on Fpo, argues against this explanation. Therefore, as yet unknown regulatory
268 and/or biochemical constraints on hydrogen metabolism in *Methanosarcina* await
269 discovery.

270

271 **Materials and Methods**

272 **Strains, media, and growth conditions.** The construction and genotypes of all
273 *Methanosarcina* strains are presented in Table S1. *Methanosarcina* strains were grown
274 as single cells (41) at 37 °C in high salt (HS) broth medium (42) or on agar-solidified
275 medium as described (43). Growth substrates provided were methanol (125 mM in
276 broth medium and 50 mM in agar-solidified medium) or sodium acetate (120 mM) under
277 a headspace of either N₂/CO₂ (80/20%) at 50 kPa over ambient pressure or H₂/CO₂
278 (80/20%) at 300 kPa over ambient pressure. Cultures were supplemented as indicated
279 with 0.1% yeast extract, 0.1% casamino acids, 10 mM sodium acetate or 10 mM
280 pyruvate. Puromycin (CalBioChem, San Diego, CA) was added at 2 µg/ml for selection
281 of the puromycin transacetylase (*pac*) gene (33). 8-aza-2,6-diaminopurine (8-ADP)
282 (Sigma, St Louis, MO) was added at 20 µg/ml for selection against the presence of *hpt*
283 (33). Tetracycline was added at 100 µg/ml to induce the tetracycline-regulated
284 *PmcrB(tetO3)* promoter (34). Standard conditions were used for growth of *Escherichia*
285 *coli* strains (44) DH5α/λ-*pir* (45) and DH10B (Stratagene, La Jolla, CA), which were
286 used as hosts for plasmid constructions.

287 **DNA methods and plasmid construction.** Standard methods were used for
288 plasmid DNA isolation and manipulation using *E. coli* hosts (46). Liposome mediated
289 transformation was used for *Methanosarcina* as described (47). Genomic DNA isolation
290 and DNA hybridization were as described (32, 42, 43). DNA sequences were
291 determined from double-stranded templates by the W.M. Keck Center for Comparative
292 and Functional Genomics, University of Illinois. Plasmid constructions are described in
293 the supporting information (Tables S2 & S3).

294 **Construction of the Δfrh and $\Delta vht/\Delta frh$ mutants.** The markerless genetic
295 exchange method (33) using plasmid pGK4 was employed to delete *frhADGB* (Δfrh) in
296 the Δhpt background of *M. barkeri* Fusaro (Tables S1, S2, & S3) using methanol/H₂/CO₂
297 as the growth substrate. The $\Delta vht/\Delta frh$ mutant was constructed by deleting *vhtGACD* in
298 the Δfrh markerless mutant by the homologous recombination-mediated gene
299 replacement method (32). To do this, the 5.6 kb XhoI/NotI fragment of pGK82B was
300 used to transform the Δfrh mutant to puromycin resistance on methanol medium. The
301 mutants were confirmed by PCR and DNA hybridization (data not shown).

302 **Construction of the tetracycline-regulated *vht* mutant ($P_{tet}::vht$).** The
303 tetracycline-regulated *PmcrB(tetO3)* promoter was employed to drive conditional
304 expression of the *vht* operon in *M. barkeri* WWM157 (34). This strain was constructed
305 by transforming WWM154 to puromycin resistance using the 7 kb NcoI/SpeI fragment of
306 pGK61A (Tables S1, S2, & S3). The transformants were selected on methanol plus
307 H₂/CO₂ medium in the presence of puromycin and tetracycline. The $P_{tet}::vht$ strain was
308 confirmed by DNA hybridization (data not shown). To ensure that the native *vht*
309 promoter ($Pvht$) did not interfere with expression from *PmcrB(tetO3)*, 382 bp upstream
310 of *vhtG* were deleted in $P_{tet}::vht$. This left 1038 bp intact for the expression of the *hyp*
311 operon, which is upstream of the *vht* operon and expressed in the opposite direction.

312 **Determination of Vht essentiality during growth on all substrate-types.**
313 Growth of WWM157 ($P_{tet}::vht$) and WWM154 (isogenic parent) on methanol,
314 methanol/H₂/CO₂, H₂/CO₂, and acetate were analyzed by the spot-plate method (48).
315 Cultures were first adapted for at least 15 generations to the substrate of interest;
316 tetracycline was added to each medium for growth of WWM157. Upon reaching

317 stationary phase, 10 ml of culture was washed three times and re-suspended in 5 ml HS
318 medium that lacked growth substrate. Subsequently, 10 μ l of 10-fold serial dilutions was
319 spotted onto the following: 3 layers of GB004 paper (Whatman, NJ), 2 layers of GB002
320 paper (Schleicher & Schuell BioScience, NH), 1 layer of 3MM paper (Whatman, NJ),
321 and a 0.22 mM nylon membrane (GE Water and Process Technologies, PA) soaked in
322 43 ml of HS-medium containing the substrate of interest with and without Tc. Plates
323 were sealed and incubated at 37 °C for at least two weeks in an intrachamber anoxic
324 incubator (49). Growth on acetate and methanol was tested under an atmosphere of
325 N₂/CO₂/H₂S (80/19.9/0.1 ratio), while growth on methanol/H₂/CO₂ or H₂/CO₂ was tested
326 under an atmosphere of H₂/CO₂/H₂S (80/19.9/0.1 ratio).

327 **Measurement of H₂, CH₄ and OD₆₀₀ during growth on methanol. *M. barkeri***
328 WWM85 (isogenic parent), WWM157 (*P_{tet::vht}*; grown in presence of Tc), WWM115
329 (Δ *frh*) and WWM351 (Δ *vht*/ Δ *frh*) were grown on methanol until mid-exponential phase
330 (OD₆₀₀ c. 0.5) and then 1 ml (WWM85 and WWM157) or 5ml (WWM115 and WWM351)
331 were inoculated into 100 ml HS-methanol in a 500 ml serum bottle. For WWM157, the
332 culture was washed once prior to inoculation with or without tetracycline. To measure H₂
333 and CH₄, ca. 1 ml or 2 ml headspace sample was withdrawn aseptically from the culture
334 at various time points with a syringe that had been flushed with sterile, anaerobic N₂.
335 The gas sample was then diluted into 70 ml helium. A gas-tight syringe flushed with
336 helium was subsequently used to withdraw 3 ml of the diluted sample, which was then
337 injected into an SRI gas chromatograph, equipped with a reduction gas detector (RGD)
338 and a thermal conductivity detector (TCD) at 52°C. The RGD column was a three-foot
339 long 13X molecule sieve, whereas the TCD column was a six-foot HayeSep D. RGD

340 was used to detect H₂ by peak height and TCD for CH₄ by peak area. Helium was used
341 as the carrier gas. OD₆₀₀ was also measured during the growth curve.

342

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- 495
- 496

497 **Figure Legends**

498 **Figure 1. The putative H₂ cycling electron transport chain of *M. barkeri*.** Growth on
499 C-1 substrates generates reduced cofactor F420 (F420_{red}), which is a hydride carrying
500 cofactor analogous to NADH, and the reduced form of the small electron-carrying
501 protein ferredoxin (Fd_{red}). During acetoclastic methanogenesis only Fd_{red} is produced.
502 These reduced electron carriers are re-oxidized in the cytoplasm by the Frh and Ech
503 hydrogenases, respectively, with concomitant consumption of protons to produce
504 molecular H₂. H₂ subsequently diffuses out of the cell where it is re-oxidized by the Vht
505 hydrogenase, which has an active site located on the outer face of the cell membrane.
506 This reaction releases protons on the outside of the cell and produces reduced
507 methanophenazine (MPH₂), a membrane-bound electron carrier analogous to
508 ubiquinone. MPH₂ subsequently delivers electrons to the enzyme heterodisulfide
509 reductase (Hdr), which serves as the terminal step in the *Methanosarcina* electron
510 transport chain. This final reaction regenerates coenzyme B (CoB-SH) and coenzyme M
511 (CoM-SH) from the mixed disulfide (CoM-S-S-CoB), which is produced from the free
512 thiol cofactors during methanogenic metabolism. Electron (e⁻) flow and scalar protons
513 (H⁺) are shown in red. It should be noted that *M. barkeri* can also re-oxidize F420_{red}
514 using the membrane-bound, proton-pumping F420-dehydrogenase (Fpo). Thus, the cell
515 has a branched electron transport chain and, therefore, is not dependent on H₂ cycling
516 during growth on methylotrophic substrates (16); however, both pathways for electron
517 transport from F420 have identical levels of energy conservation: namely 4 H⁺/2e⁻. It
518 should also be noted that the Ech hydrogenase acts as a proton pump in addition to its
519 role in H₂ cycling, thus electron transport during acetoclastic methanogenesis conserves

520 $6\text{H}^+/2\text{e}^-$. Individual subunits of the various enzymes are indicated by letters (e.g. A, B,
521 C...).

522

523 **Figure 2. Hydrogen and methane production during methylotrophic growth.** The
524 partial pressures of H_2 (*panel A*) and methane (*panel B*) were monitored during the
525 course of growth (as indicated by optical density, *panel C*) in methanol medium for
526 various *M. barkeri* strains. Strains used were: *M. barkeri* isogenic parental strain (brown
527 circles, WWM85), tetracycline-regulated *vht* mutant (WWM157) with tetracycline (dark
528 blue squares) and without tetracycline (light blue squares), *frh* deletion mutant (red
529 triangles, WWM115), and *frh/vht* double deletion mutant (green diamonds, WWM351).
530 Measurements were performed in triplicates as described in the methods section.
531 Complete strain genotypes can be found in Table S1.

532

533 **Figure 3. Essentiality of the Vht hydrogenase in *M. barkeri*.** Cultures of the $\text{P}_{tet}::vht$
534 mutant (WWM157) and its isogenic parent (WWM154) were adapted to four different
535 substrates of interest (and in the presence of tetracycline for $\text{P}_{tet}::vht$), then washed,
536 serially diluted, and incubated with each substrate with and without tetracycline (Tet).
537 The media used indicate the ability to grow via each of the four known methanogenic
538 pathways: (i) methylotrophic (methanol), (ii) methyl-reduction (methanol/ H_2/CO_2), (iii)
539 hydrogenotrophic (H_2/CO_2), and (iv) acetoclastic (acetate).

540

541 **Figure 4. The role of H_2 cycling in the four methanogenic pathways of *M. barkeri*.**
542 *M. barkeri* utilizes four distinct methanogenic pathways to allow growth on a variety of

543 substrates. In the hydrogenotrophic pathway (shown in red), CO₂ is reduced to methane
544 using electrons derived from H₂, while in the methyl-reducing pathway (shown in
545 orange), H₂ is used to reduce C-1 compounds, such as methanol, directly to CH₄.
546 During methylotrophic methanogenesis (shown in green), C-1 compounds are
547 disproportionated to CO₂ and methane, with one molecule of the C-1 compound
548 oxidized to provide electrons for reduction of three additional molecules to methane.
549 Finally, in the acetoclastic pathway (shown in blue), acetate is split into a methyl group
550 and an enzyme-bound carbonyl moiety. The latter is oxidized to CO₂ to provide
551 electrons required for reduction of methyl group to methane. The steps catalyzed by
552 Fpo, Frh, Vht, Ech and Hdr proteins are indicated. Steps involving H₂ cycling are shown
553 as labeled, hyphenated arrows. An alternate, H₂-independent electron transport
554 pathway is shown in brown. Experimental data support the function of this alternate
555 pathway during methylotrophic methanogenesis, but not in hydrogenotrophic or methyl-
556 reducing methanogenesis (as indicated by the hyphenated brown box). Abbreviations;
557 Fpo, F420 dehydrogenase; Frh, F420-reducing hydrogenase; Vht, methanophenazine-
558 dependent hydrogenase; Ech, energy-converting ferredoxin-dependent hydrogenase;
559 Hdr, heterodisulfide reductase; CoM, coenzyme M; CoB, coenzyme B; CoB-CoM, mixed
560 disulfide of CoB and CoM; MP/MPH₂, oxidized and reduced methanophenazine;
561 F420_{ox}/F420_{red}, oxidized and reduced cofactor F420; Fd_{ox}/Fd_{red}, oxidized and reduced
562 ferredoxin; CHO-MF, formyl-methanofuran; H₄SPT, tetrahydrosarcinapterin; CHO-
563 H₄SPT, formyl-H₄SPT; CH≡H₄SPT, methenyl-H₄SPT; CH₂=H₄SPT, methylene-H₄SPT;
564 CH₃-H₄SPT, methyl-H₄SPT; CH₃-CoM, methyl-coenzyme M; CoA, coenzyme A;

565 CH₃CO-CoA, acetyl-coenzyme A; [CO], enzyme-bound carbonyl moiety; ATP,
566 adenosine triphosphate.

567

568 **Figure S1. Hydrogenase operons in *Methanosarcina barkeri*.** Three distinct types of
569 hydrogenase are encoded by *M. barkeri* Fusaro. The *frh* and *fre* operons encode
570 putative F420-reducing hydrogenases, while the *vht* and *vhx* operons encode putative
571 methanophenazine-reducing hydrogenases. Genetic and biochemical data show that
572 neither the *fre* nor the *vhx* operon is capable of producing an active hydrogenase under
573 any growth condition yet examined (T.D. Mand, G. Kulkarni, and W.W. Metcalf, 2018,
574 bioRxiv doi: <https://doi.org/10.1101/334656>). The *ech* operon encodes a ferredoxin-
575 dependent energy-conserving hydrogenase. The locus tags are shown below each
576 gene, with the prefix “Mbar_” omitted to save space (indicated by an asterisk) in some
577 cases.

578

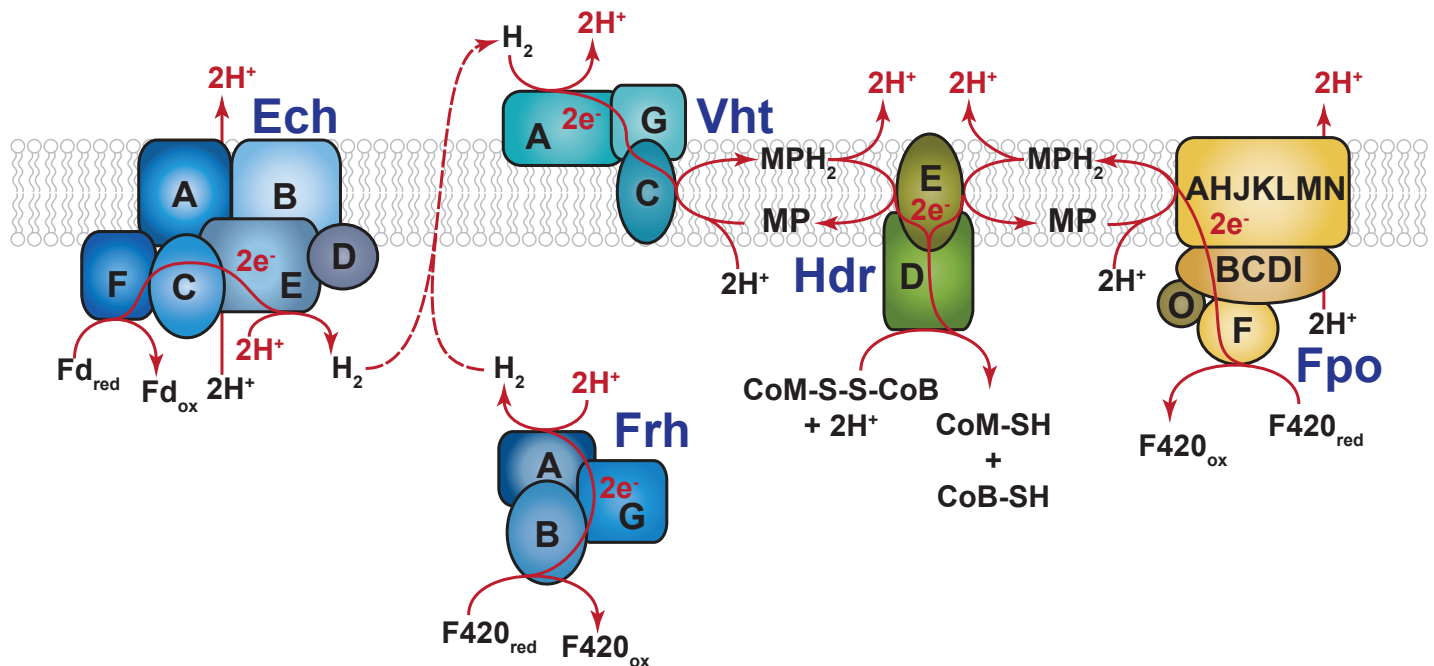


Figure 1. The putative H_2 cycling electron transport chain of *M. barkeri*. Growth on C-1 substrates generates reduced cofactor F420 ($F420_{red}$), which is a hydride carrying cofactor analogous to NADH; whereas growth on acetate generates the reduced form of the small electron-carrying protein ferredoxin (Fd_{red}). These reduced electron carriers are re-oxidized in the cytoplasm by the Frh and Ech hydrogenases, respectively, with concomitant consumption of protons to produce molecular H_2 . H_2 subsequently diffuses out of the cell where it is re-oxidized by the Vht hydrogenase, which has an active site located on the outer face of the cell membrane. This reaction releases protons on the outside of the cell and produces reduced methanophenazine (MPH_2), a membrane-bound electron carrier analogous to ubiquinone. MPH_2 subsequently delivers electrons to the enzyme heterodisulfide reductase (Hdr), which serves as the terminal step in the *Methanosarcina* electron transport chain. This final reaction regenerates coenzyme B (CoB-SH) and coenzyme M (CoM-SH) from the mixed disulfide (CoM-S-S-CoB), which is produced from the free thiol cofactors during methanogenic metabolism. Electron (e^-) flow and scalar protons (H^+) are shown in red. It should be noted that *M. barkeri* can also re-oxidize $F420_{red}$ using the membrane-bound, proton-pumping F420-dehydrogenase (Fpo). Thus, the cell has a branched electron transport chain and, therefore, is not dependent on H_2 cycling during growth on methylotrophic substrates (16); however, both pathways for electron transport from F420 have identical levels of energy conservation: namely $4 H^+/2e^-$. It should also be noted that the Ech hydrogenase acts as a proton pump in addition to its role in H_2 cycling, thus electron transport during acetoclastic methanogenesis conserves $6H^+/2e^-$. Individual subunits of the various enzymes are indicated by letters (e.g. A, B, C...).

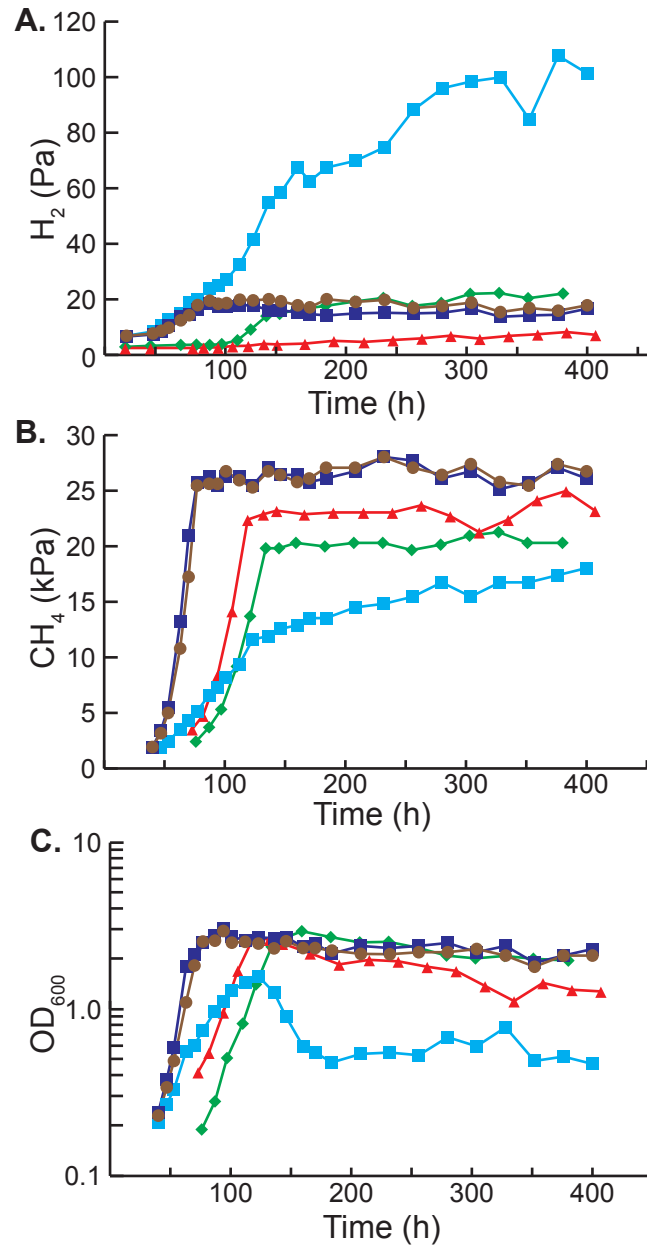


Figure 2. Hydrogen and methane production during methylotrophic growth. The partial pressures of H₂ (panel A) and methane (panel B) were monitored during the course of growth (as indicated by optical density, panel C) in methanol medium for various *M. barkeri* strains. Strains used were: *M. barkeri* isogenic parental strain (brown circles, WWM85), tetracycline-regulated *vht* mutant (WWM157) with tetracycline (dark blue squares) and without tetracycline (light blue squares), *frh* deletion mutant (red triangles, WWM115), and *frh/vht* double deletion mutant (green diamonds, WWM351). Measurements were performed in triplicates as described in the methods section. Complete strain genotypes can be found in Table S1.

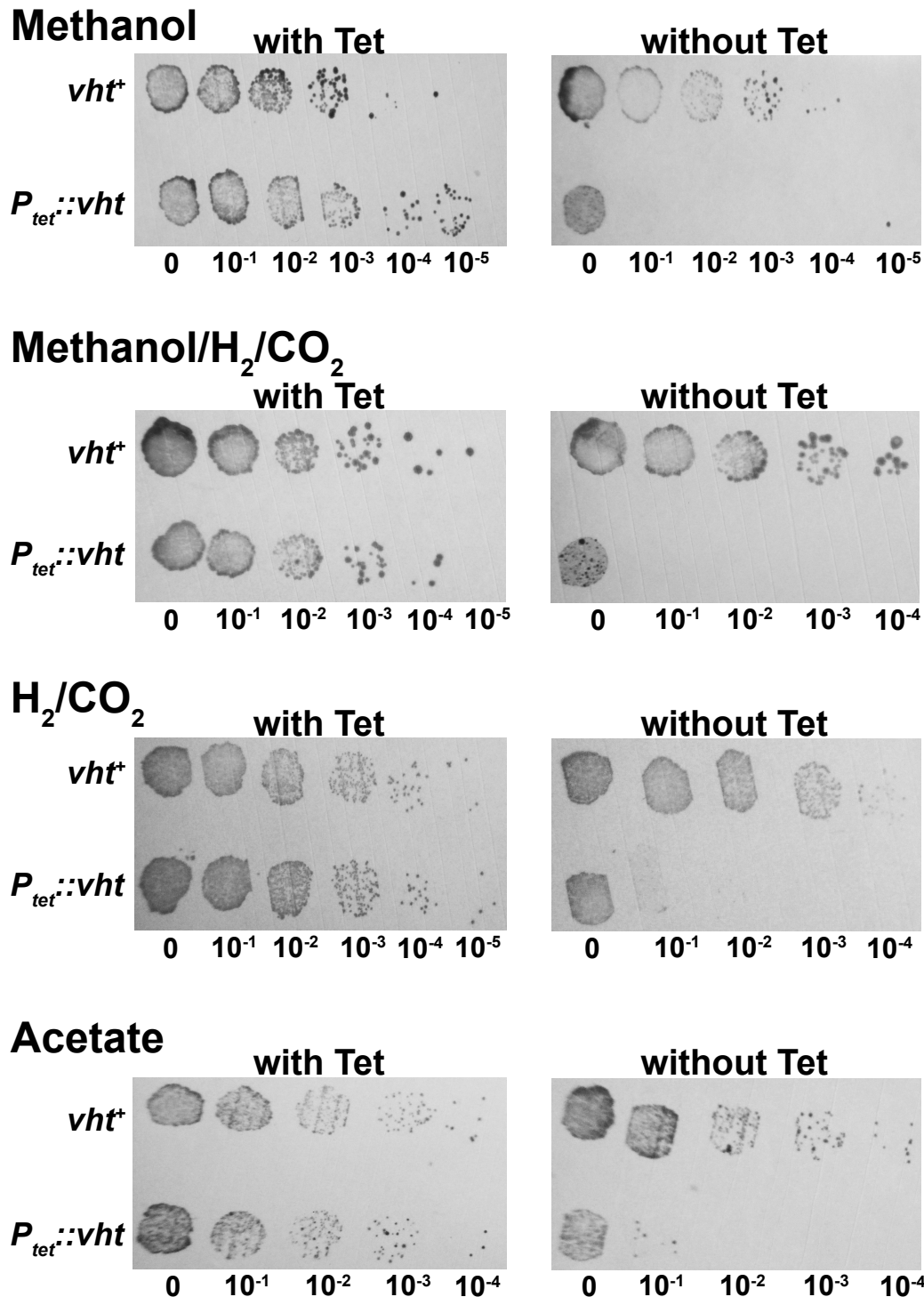


Figure 3. Essentiality of the Vht hydrogenase in *M. barkeri*. Cultures of the *P_{tet}::vht* mutant (WWM157) and its isogenic parent (WWM154) were adapted to four different substrates of interest (and in the presence of tetracycline for *P_{tet}::vht*), then washed, serially diluted, and incubated with each substrate with and without tetracycline (Tc). The media used indicate the ability to grow via each of the four known methanogenic pathways: (i) methylotrophic (methanol), (ii) methyl-reduction (methanol/H₂/CO₂), (iii) hydrogenotrophic (H₂/CO₂), and (iv) acetate.

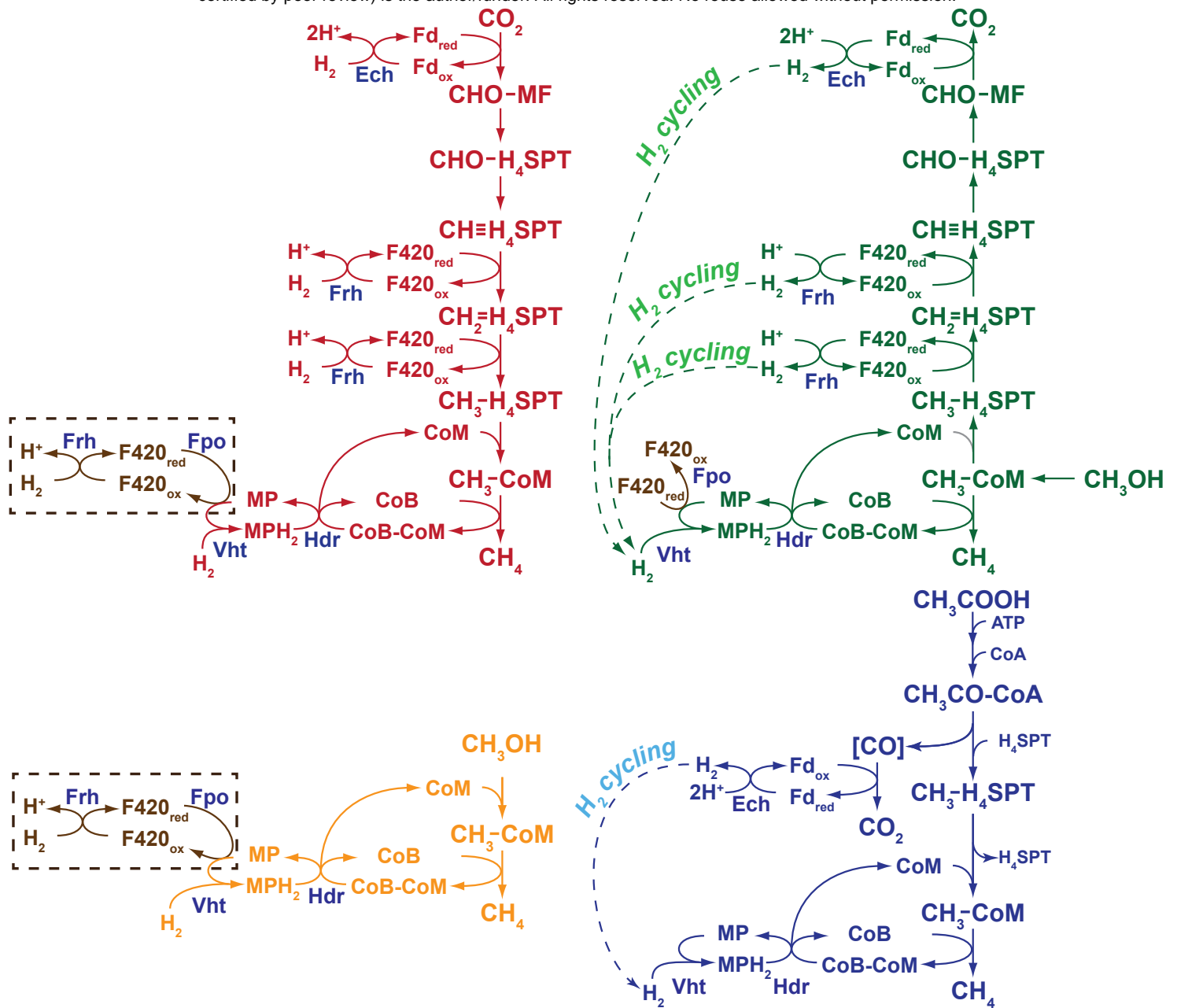


Figure 4. The role of H₂ cycling in the four methanogenic pathways of *M. barkeri*. *M. barkeri* utilizes four distinct methanogenic pathways to allow growth on a variety of substrates. In the hydrogenotrophic pathway (shown in red), CO₂ is reduced to methane using electrons derived from H₂, while in the methyl-reducing pathway (shown in orange), H₂ is used to reduce C-1 compounds, such as methanol, directly to CH₄. During methyl-trophic methanogenesis (shown in green), C-1 compounds are disproportionated to CO₂ and methane, with one molecule of the C-1 compound oxidized to provide electrons for reduction of three additional molecules to methane. Finally, in the aceticlastic pathway (shown in blue), acetate is split into a methyl group and an enzyme-bound carbonyl moiety. The latter is oxidized to CO₂ to provide electrons required for reduction of methyl group to methane. The steps catalyzed by Fpo, Frh, Vht, Ech and Hdr proteins are indicated. Steps involving H₂ cycling are shown as labeled, hyphenated arrows. An alternate, H₂-independent electron transport pathway is shown in brown. Experimental data support the function of this alternate pathway during methyl-trophic methanogenesis, but not in hydrogenotrophic or methyl-reducing methanogenesis (as indicated by the hyphenated brown box). Abbreviations; Fpo, F420 dehydrogenase; Frh, F420-reducing hydrogenase; Vht, methanophenazine-dependent hydrogenase; Ech, energy-converting ferredoxin-dependent hydrogenase; Hdr, heterodisulfide reductase; CoM, coenzyme M; CoB, coenzyme B; CoB-CoM, mixed disulfide of CoB and CoM; MP/MPH₂, oxidized and reduced methanophenazine; F420_{ox}/F420_{red}, oxidized and reduced cofactor F420; Fd_{ox}/Fd_{red}, oxidized and reduced ferredoxin; CHO-MF, formyl-methanofuran; H₄SPT, tetrahydrosarcinapterin; CHO-H₄SPT, formyl-H₄SPT; CH≡H₄SPT, methenyl-H₄SPT; CH₂=H₄SPT, methylene-H₄SPT; CH₃-H₄SPT, methyl-H₄SPT; CH₃-CoM, methyl-coenzyme M; CoA, coenzyme A; CH₃CO-CoA, acetyl-coenzyme A; [CO], enzyme-bound carbonyl moiety; ATP, adenosine triphosphate.