

1 Original Article

2 **Ozone mediates tumor-selective cell death caused by air plasma-activated medium**
3 **independently of NOx**

4 Manami Suzuki-Karasaki^{1,2,3}, Yushi Ochiai³, Shizuka Innami³, Hiroshi Okajima^{3,4}, Miki
5 Suzuki-Karasaki³, Hideki Nakayama², Yoshihiro Suzuki-Karasaki³#

6 ¹Graduate School of Medical Sciences, Kumamoto University, Kumamoto, Japan

7 ²Department of Oral and Maxillofacial Surgery, Faculty of Life Sciences, Kumamoto University,
8 Kumamoto, Japan.

9 ³Department of Research and Development, Plasma ChemiBio Laboratory, Nasushiobara, Tochigi,
10 Japan.

11 ⁴Communication&Control Systems Company, Tokyo Keiki Incorporation, Tokyo, Japan

12 #Corresponding author: Yoshihiro Suzuki-Karasaki, Ph.D., Plasma ChemiBio Laboratory, 398
13 Takaatsu, Nasushiobara 329-2813, Tochigi, Japan.

14 E-mail: suzuki.pcbi@gmail.com

15 Keywords: cancer; dissolved ozone; tumor-specificity; anticancer; monopolar perinuclear
16 mitochondrial clustering; plasma-activated medium

17 **Abbreviations.** 7-AAD, 7-amino-actinomycin D; APAM, air plasma-activated medium; CAP, cold
18 atmospheric plasma; CL, cardiolipin; Drp, dynamin-related protein; Fer-1, ferrostatin-1, Hoe,
19 Hoechst 33342; LPO, lipid peroxide; MitoSOX, MitoSOX Red CMXRos; MPMC, monopolar
20 perinuclear mitochondrial clustering; mROS, mitochondrial reactive oxygen species; MTR,
21 MitoTracker Red; nROS, nuclear reactive oxygen species; NAC, *N*-acetylcysteine NAO, 10-*N*-nonyl
22 acridine orange; NAC, *N*-acetylcysteine; NC, Nocodazole; OC, oral cancer; OS, osteosarcoma;
23 PNMC, perinuclear mitochondrial clustering; TTG, tubulin tracker green.

24 **Abstract**

25 Cold atmospheric plasma and plasma-treated liquids (PTLs) are emerging promising tools for
26 tumor-targeted cancer treatment, as they preferentially injure tumor cells more than
27 non-malignant cells. Oxidative stress is critical to the antitumor effect, but the oxidant mediating
28 the effect is debatable. Previously, we reported that air plasma-activated medium (APAM) has
29 tumor-selective cytotoxicity in vitro and in vivo. Moreover, an unusual mitochondrial positioning
30 named monopolar perinuclear mitochondrial clustering (MPMC) and nuclear damage proceeds
31 to cell death. We noticed that air plasma generation was accompanied by ozone (O₃) formation,
32 leading to suppose the possible role of O₃ in the effect of APAM. In this study, we produced an
33 O₃-dissolved medium (ODM) and comparatively analyzed its biological effect with APAM. Both
34 agents had comparable amounts of dissolved O₃ (dO₃), while APAM, but not ODM, contained
35 nitrite and nitrate. Like APAM, ODM could induce apoptosis, nonapoptotic cell death, tubulin
36 remodeling, MPMC, and nuclear shrinkage. Catalase mitigated all these events. The increases in

37 various intracellular and mitochondrial reactive oxygen species (ROS) and lipid peroxides
38 proceeded to cell death, and catalase also prevented them. Conversely, suppressing cellular H₂O₂
39 removal systems augmented mitochondrial ROS production and cell death. In contrast, like
40 APAM, ODM minimally increased ROS production and MPMC in non-malignant cells. These
41 results indicate that dO₃ is a critical mediator of the actions of APAM, including tumor-selective
42 induction of MPMC and cell death. Our findings suggest ODM could be a more
43 chemically-defined alternative to PTLs in cancer treatment.

44 **1. Introduction**

45 Like other intractable cancers, once developed, oral cancers (OCs) are highly refractory, recurrent,
46 metastatic, resistant to anticancer agents and irradiation, and hard to remove by surgical
47 resection [1,2]. Thus, improving prognosis through standard therapies remains challenging, and
48 innovative treatment is urgently required. Cell death induction is one of the powerful strategies
49 to eliminate cancers. Apoptosis is the representative mode of cancer cell death targeted by
50 conventional therapies, including most anticancer agents and radiation. However, intractable
51 cancers, including OCs, have congenital and acquired resistance. It is widely accepted that
52 besides apoptosis, various nonapoptotic cell death forms contribute to cancer cell death [3].
53 Consequently, drugs and tools targeting other forms of cell death could serve as successful
54 venues for OC treatment.

55 Cold atmospheric plasmas (CAPs) are partially ionized gasses containing ions, electrons, and free
56 radicals and have tumor-selective toxicity. Flushing CAPs to various solutions, such as culture
57 media, buffers, and infusion fluids, result in plasma-treated liquids (PTLs), which also
58 preferentially injure tumor cells while damaging non-malignant cells minimally [4–6]. CAPs and
59 PALs also reduced tumor growth in animal models with minimal adverse events. Thus, CAP-based
60 treatments have attracted much attention in cancer treatment. However, PTLs have different
61 chemical and biological properties, depending on the physical nature of the plasmas generated,
62 the solutions used, and various experimental parameters. In addition, the biological outcomes
63 may also vary significantly depending on the target cell systems. As a result, their actions,
64 including targeted cell death, are highly complicated; While most of them induce apoptosis,
65 some of them can trigger nonapoptotic cell death forms, including autophagy [7,8], necroptosis
66 [6], and ferroptosis [9] in various cancer cell models. Although numerous reports have shown the
67 vital role of reactive oxygen and nitrogen species (ROS/RNS) in mediating the antitumor effect,
68 the detailed mechanisms remain obscure. Moreover, PTLs have complicated components, and
69 complete chemical identification of the active substances is challenging. Indeed, many kinds of
70 chemical species, atomic oxygen (O), singlet oxygen (¹O₂), superoxide (O₂^{•-}), hydroxyl radicals
71 ([•]OH), hydrogen peroxide (H₂O₂), nitric oxide (NO[•]), and nitrogen oxide anions (NOx⁻), such as
72 nitric dioxide (NO₂[•]), nitrite (NO₂⁻), and nitrate (NO₃⁻) have been proposed to participate in the
73 effect. More complicatedly, the exact contents of these species vary considerably depending on
74 the CAP source, settings, and ambient conditions. In addition, these oxidants can function
75 cooperatively. Several studies have shown the synergistic effect of H₂O₂ and NO₂⁻ in cell death
76 and tumor selectivity [10, 11]. Consequently, the primary mediator(s) and their molecular targets
77 in the antitumor properties are poorly understood. Such chemical ambiguity heavily frustrates
78 distinct descriptions of the mechanism of action and clinical application.

79 O₃ is the oxygen allotrope consisting of three oxygen atoms. It is an acrid gas at ordinary
80 temperature and pressure. O₃ is an unstable molecule with unique physicochemical and

81 biological properties due to its resonance structures. It has the second-highest oxidizing power
82 behind fluorine. O₃ directly injures the cell membrane of bacteria through the oxidation of
83 phospholipids and lipoproteins and has potent antibacterial activity. O₃ also inactivates viruses,
84 fungi, yeast, and protozoa. Therefore, it has been widely used for sterilization and disinfection. O₃
85 has also been utilized in treating various diseases for over a century [12]. Moreover, numerous in
86 vitro and in vivo studies and a few clinical trials [13–15] have shown the anticancer effects of O₃.
87 O₃ has a direct antitumor effect in some cancers while having indirect effects, such as
88 immunomodulation, synergistic or adjuvant effects with various anticancer drugs and radiation
89 (cisplatin, 5-fluorouracil, etoposide, and gemcitabine) [16–18]. O₃ is administered to animals in
90 various ways, including topical gas, rectal or intraperitoneal insufflation, and intravenous or
91 intratumoral injection. O₃ has a substantial solubility in water (14 mmol/L at 20°C) and has also
92 been used with nanotechnology as ozonated or ozonized water. However, it must be aware that
93 pure O₃ production by conventional methods is challenging. The most widely used method to
94 generate O₃ is the silent discharge in the air. However, this method results in the excitation of
95 molecular nitrogen (N₂) as well as molecular oxygen (O₂), leading to the production of various
96 nitrogen oxides (NO_x). Accordingly, ozonated/ozonized fluids could contain a mixture of dO₃ and
97 multiple bystander nitrogen oxide anions (NO_x⁻). NO_x, such as nitric oxide (NO[•]) and nitric
98 dioxide (NO₂[•]), can produce O₃ and more complicated, harmful oxidants in the presence of light.
99 In addition, NO[•] has been recognized as a primary modulator of cell death and a potential target
100 for anticancer therapy [19, 20]. As a result, the co-existence of bystander NO_x could confuse the
101 precise evaluation of the biological activity of dO₃ in these ozonated fluids. Therefore, eliminating
102 NO_x is essential to define the effect of dO₃ itself.

103 Mitochondria are highly plastic, dynamic, and heterogenous organelles in connection with these
104 diverse functions. Their size, shape (macroscopic shape), and location vary in different tissues,
105 cells, and experimental conditions. An emerging view is that mitochondrial shape and location
106 changes are not passive events. Instead, they are active events coupled with cellular functions,
107 cell death, and survival [21–23]. Mitochondria can distribute broadly throughout the cytoplasm
108 (Pan-cytoplasmic), subplasmalemmal, or perinuclear sites. Pan-cytoplasmic distribution governs
109 Ca²⁺ transport from the endoplasmic reticulum (ER) via tethering with ER [24]. The
110 subplasmalemmal distribution controls Ca²⁺ channel activity by regulating mitochondrial
111 function as a Ca²⁺ reservoir [25]. The mitochondrial location around the perinuclear regions is
112 called perinuclear mitochondrial clustering (PNMC). This response is induced by stresses, such as
113 hypoxia and heat shock, and is associated with mitochondrial fragmentation [26, 27]. This type of
114 positioning is thought to be adaptive to stresses and cytoprotective. Microtubule polymerization
115 inhibitors, such as Nocodazole (NC) and Cholchicine, and the expression of KinAΔN355 mutant of
116 Kinesin prevent it, indicating that microtubule- and Dynein and Kinesin-dependent. Mitochondria
117 may be transported along the microtubule track. PNMC participates in cancer adaptation to
118 hypoxia through mitochondrial reactive oxygen species (ROS) production and Hypoxia-inducible
119 factor-1α, the master transcription factor of hypoxic signaling [26]. Recently, we reported another
120 type of mitochondrial distribution associated with cancer cell death [28]. In this case,
121 mitochondria become fragmented and gather one side of the perinuclear sites. Drastic changes
122 in the shape and location of tubulin accompany this phenomenon. Like mitochondria, tubulin
123 concentrates one side of the nuclei. NC and antioxidants prevent these changes, suggesting the
124 involvement of similar mitochondrial transport mechanisms in PNMC and ROS. We named this
125 unique mitochondrial positioning monopolar perinuclear mitochondrial clustering (MPMC) to
126 distinguish it from other known mitochondrial locations.

127 We noticed that air plasma generation was accompanied by O₃ production. This notion led us to
128 assume the possible role of the oxidant in the effect of APAM. To examine this hypothesis, we
129 developed a new dO₃-generating system to produce O₃ without exciting N₂ and dissolved it in a
130 culture medium by bubbling. The resulting O₃-dissolved medium (ODM) can keep dO₃ for a
131 month at -80°C. In this study, we comparatively analyzed the biological effects between APAM
132 and ODM. Results showed that without NO_x⁻, dO₃ could mimic various effects of APAM, including
133 tumor-selective induction of MPMC and oxidative cell death.

134 **2. Materials and methods**

135 **2.1. Materials**

136 Unless otherwise specified, all chemicals were purchased from Sigma-Aldrich (St. Louis, MO, USA).
137 The pan-caspase inhibitor Z-VAD-FMK was obtained from Merck Millipore (Darmstadt, Germany).
138 All insoluble reagent was dissolved in dimethyl sulfoxide (DMSO), and the stock solution diluted
139 in 10% Fetal Bovine Serum(FBS) contained Dulbecco's Modified Eagle Medium (DMEM; final
140 DMSO concentration, <0.1%) before use.

141 **2.2. Cell culture**

142 The human OC cell line HOC-313 was kindly provided by the Department of Oral and
143 Maxillofacial Surgery, Graduate School of Medical Science, Kanazawa University (Kanazawa,
144 Japan). Another human OC cell line SAS and glioblastoma (GBM) cell line U251MG were obtained
145 by the Japanese Collection of Research Bioresource (JCRB) Cell Bank of the National Institute of
146 Biomedical Innovation, Health, and Nutrition (Osaka, Japan). Human fetal osteoblast hFOB was a
147 kind gift from Dr. T. Ando (Yamanashi University). Human dermal fibroblasts (HDFs) were
148 obtained from Cell Applications (San Diego, CA, USA). The cells were maintained in 10% FBS
149 (Serena Europe GmbH, Brandenburg, Germany) containing DMEM (Merck, NM, USA)
150 supplemented with 100 U/mL penicillin and 100 µg/mL streptomycin (FBS/DMEM) at 37 °C in a
151 5% CO₂ incubator.

152 **2.3. O₃ and ODM preparations**

153 O₃ was generated in two different ways. Method A produced O₃ by irradiating an excimer light
154 (185 nm) into the air using a UV-C lamp (Fig. 1A). This method allows specific excitation of O₂ but
155 not molecular nitrogen (N₂), the cause of NO_x production. Method B made it by the silent
156 discharge of highly pure (99.9%) O₂ (Taiyo Nippon Sanso JFP Corporation, Kanagawa, Japan)
157 using the O₃ generator (Communication&Control Systems Company, Tokyo Keiki Incorporation,
158 Tokyo, Japan) equipped with a dielectric barrier discharge probe illustrated in Fig. 1B. This closed
159 system does not allow air contamination and the resulting excitation of N₂. The generated O₃ was
160 introduced into phenol red-free DMEM (Fuji Film Wako Chemicals, Osaka, Japan) by bubbling,
161 resulting in an O₃-dissolved medium (ODM). The typical experiment conditions are a 10–15 KV
162 voltage and a gas flow rate of 0.4 L/min. ODM was made by O₃ bubbling at the ratio of 1 min
163 bubbling/1 mL ODM. The original ODM was diluted to a final concentration of 6.3–50% with
164 phenol red-free DMEM (biochemical experiments) or FBS/DMEM (for cell experiments).

165 **2.4. APAM preparation**

166 Air plasma and APAM were prepared as reported previously [28]. Briefly, the plasma was
167 generated from the ambient air using a Piezobrush™ PZ2 model plasma jet (relyon, Germany)
168 equipped with a piezo element. The APAM (1 mL) was made by flushing the plasma to phenol
169 red-free DMEM for 1 min at a distance of 20 mm. The original APAM was diluted to a final
170 concentration of 6.3–50% with the medium (biochemical experiments) or FBS/DMEM (for cell
171 experiments).

172 **2.5. Quantitation of oxidants**

173 dO₃ concentration was measured by a polarographic O₃ meter (DOZ-1000PE, Custom
174 Corporation, Tokyo, Japan). The concentration was also measured by a Digital Pack test
175 (DPM2-O₃, range 0.25~5 mg/L, Kyoritsu Chemical-Check Lab. Corp., Kanagawa, Japan). The
176 concentrations of NO₂⁻ and NO₃⁻ were measured using a Kit (NO₂ / NO₃ Assay
177 Kit-CII(Colorimetric) Griess Reagent Kit, Dojindo, Kumamoto, Japan) according to the
178 manufacturer's protocols. The concentrations of NO₂⁻ and NO₃⁻ were calculated using a standard
179 curve made using the authentic samples from the kit.

180 **2.6. Cell viability assay**

181 Cell viability was measured by the WST-8 assay using a Cell Count Reagent SF (Nacalai Tesque,
182 Inc, Kyoto, Japan). Cells in FBS/DMEM were plated at 4 x 10³ cells/well in 96-well plates (Corning
183 Incorporated, NY, USA). They were incubated at 37 °C overnight, added with agents, and
184 incubated for 72h. In the resistance assay, agents were added after incubating cells for 72 h. This
185 prolonged preincubation allows cells to reach a higher density and become less sensitive to
186 agents. After treatment, cells were added 10 µL Cell Count Reagent SF and further incubated for
187 2h. Absorbances at 450 nm were measured by a Nivo 3F Multimode Plate Reader (PerkinElmer
188 Japan Company, Ltd., Yokohama, Japan).

189 **2.7. Determination of cell death**

190 Cell death was evaluated by fluorescence microscopy as previously described [29] with minor
191 modifications. Briefly, cells in FBS/DMEM were seeded at 1.5 x 10⁴ cells/well in a 35 mm
192 Poly-Lysine-coated glass bottom dish (D11531H, Matsunami Glass Ind. Corp., Osaka, Japan) and
193 treated with agents at 37 °C overnight in a 5% CO₂ incubator. After removing the medium by
194 aspiration, the cells were stained with 4 µM each of Calcein-AM and Ethidium bromide
195 homodimer-1 (EthD-1) for 30min to label live and dead cells, respectively, using a kit (LIVE/DEAD
196 Viability/Cytotoxicity kit) Cells were washed and immersed in FluoroBrite™ DMEM. Images were
197 obtained using BZ-X 710 digital Biological Microscopy (Keyence, Osaka, Japan).

198 **2.8. Imaging of mitochondria and tubulin**

199 The mitochondrial morphology and positioning and tubulin were analyzed as previously
200 described [28]. Briefly, cells FBS/DMEM were seeded at a 5 x 10⁴/mL density in a 35 mm
201 Poly-Lysine-coated glass bottom dish, as described above, and treated with agents at 37 °C for 2
202 or 18 h in a 5% CO₂ incubator. After removing the medium, the cells were stained with 20 nM
203 MitoTracker™ Red CMXRos (MTR) or MitoTracker™ Green FM (MTG, ThermoFisher). The nuclei
204 were counterstained with 1 mg/mL of Hoechst33342 for 1 h at 37 °C in a 5% CO₂ incubator.
205 Tubulin was stained with Tubulin Trackerr™ Green (TTG, ThermoFisher). After washing with

206 FluoroBrite™, the cells were immersed in FluoroBrite™. Images were obtained using a BZ X-710
207 Digital Biological Microscope (Keyence, Osaka, Japan) equipped with a 100 ×, 1.40 n.a.
208 UPlanSApo Super-Apochromat, coverslip-corrected oil objective (Olympus, Tokyo, Japan) and
209 analyzed using BZ-H3A application software (Keyence). For each experimental group, the
210 mitochondria in 2 or 3 different pictures were counted for three different distribution patterns,
211 pan-cytoplasmic (Type I), PNMC (Type II), and MPMC (Type III).

212 **2.9. Measurement of intracellular ROS generation**

213 As described above, cells were seeded in a 35 mm Poly-Lysine-coated glass bottom dish and
214 treated with agents at 37 °C for 2 h in a 5% CO₂ incubator. After removing the medium, the cells
215 were stained with 1 μM OxiORANGE™ (OXO), HYDROP™ (Hydrop, Goryo Chemicals, Sapporo,
216 Japan), or 5 μM MitoSOX™ Red (MitoSOX, ThermoFisher) to detect ·OH, H₂O₂, and O₂^{·-},
217 respectively. Images were obtained using BZ-X 710 digital Biological Microscopy fluorescence
218 microscopy and analyzed using BZ-H3A application software as described above.

219 **2.10. Statistical analysis**

220 Data are presented as mean ± standard deviation (SD) and analyzed by a one-way analysis of
221 variance followed by Tuckey's post hoc test using statistical software with Excel 2019 for windows
222 (SSRI, Tokyo, Japan). P < 0.05 was considered statistically significant.

223 **3. Results**

224 **3.1. ODM and APAM have dO₃**

225 Fig. 1A and B show the schematic diagrams of the generating systems for O₃ and dO₃. O₃ was
226 generated from the air using UV-C excimer light irradiation (Fig. 1A) or from highly pure O₂ using
227 dielectric barrier discharge (Fig. 1B). The O₃ was introduced into a culture medium by bubbling,
228 resulting in ODM. Next, we quantified the oxidants in ODM made by method B. As expected, the
229 ODM contained substantial amounts of dO₃ (Fig. 1C). The quantity was increased over
230 introducing time for at least 30 min. APAM had comparable levels of dO₃. On the other hand,
231 APAM contained about 100 μM of NO₂⁻ and NO₃⁻ while they were under detection limits in ODM
232 (Fig. 1D). ODM made by method A also had dO₃ but not NO₂⁻ and NO₃⁻ (not shown).

233 **3.2. ODM can kill different tumor cells**

234 Next, we examined whether ODM affected tumor cell growth. Cells were treated with varying
235 concentrations (12.5– 50% solution) of ODM for 72 h and analyzed for cell growth by WST-8
236 assay. The treatment dose-dependently reduced the viability of OC cell lines, such as SAS and
237 HOC-313 (Fig. 2A, B). ODM had a similar effect in several OS and GBM cell lines, including HOS
238 (Fig. 2C, D). Cellular sensitivity to ODM varied considerably depending on the cellular growing
239 status. In sensitive cells, ODM (≤12.5%) was enough to reduce their growth potently (≥50%
240 reduction), while in insensitive cells, at higher concentrations (≥25%), it had a moderate effect
241 (<50% reduction). Robust cell and nuclear morphological changes accompanied the growth
242 suppression. Most untreated cells were adherent spindle cells. Treatment with ODM (50%) for 2 h
243 resulted in blebbed, less adhesive round cells, most of which maintained membrane integrity (Fig.
244 2E, F). At the same time, smooth surface round nuclei became smaller and deformed. These

245 results show that ODM can injure different cancer cell types. We further analyzed the effect of
246 ODM using OC cells as a model.

247 **3.3. ODM primarily induces oxidative cell death in an H₂O₂-dependent manner**

248 APAM can induce apoptosis and nonapoptotic cell death in OC cells depending on the cell line
249 and concentration [28]. Therefore, we examined whether ODM had similar effects. When ODM
250 reduced cell viability potently, the broad-spectrum caspase inhibitor Z-VAD-FMK failed to
251 suppress the effect in SAS cells (Fig. 3A). The necroptosis inhibitor Nec-1 did not affect the effect
252 either. We obtained similar results in HOC-313 cells (Fig. 3B). In contrast, Z-VAD-FMK and Nec-1
253 protected the cells from TRAIL cytotoxicity. The ferroptosis inhibitor Ferrostatin-1 (Fer-1) also
254 failed to affect ODM cytotoxicity (Fig. 3A, B) while preventing Erastin cytotoxicity. On the other
255 hand, when the effect of ODM was moderate, Z-VAD-FMK significantly suppressed it (Fig. 3C),
256 indicating that apoptosis plays a role in specific cellular conditions. Regardless of the degree of
257 the effect, the H₂O₂-degrading enzyme catalase protected both cells from ODM, while MnTBaP,
258 the scavenger of O₂^{•-} did not (Fig. 3C–E). Catalase also prevented APAM cytotoxicity (Fig. 3F).
259 These results show that similar to APAM, ODM induces oxidative cell death in an
260 H₂O₂-dependent manner.

261 **3.4. ODM increases mitochondrial oxidative stress in an H₂O₂-dependent manner**

262 APAM increases intracellular ROS, including those within mitochondria in OC cells [28]. Therefore,
263 we determined whether ODM also could affect cellular ROS levels. First, we analyzed the
264 intracellular O₂^{•-} using the oxidant-specific probe MitoSOX Red (MitoSOX) in live cells. O₂^{•-} was
265 significantly increased in ODM-treated cells compared with untreated cells (Fig. 4A). The increase
266 was observed before significant cellular and nuclear morphological changes. Moreover, catalase
267 abolished the effect, indicating it is H₂O₂-dependent. Treatment with the irreversible catalase
268 inhibitor 3-AT increased O₂^{•-} (Fig. 4A), supporting the role of H₂O₂. Figure 4B shows the
269 quantitative analysis of the increase and supports the view. In addition, the treatment increased
270 [•]OH, detected by OXO, and catalase suppressed the effect, too (Fig. 4C). As MitoSOX and OXO
271 can detect O₂^{•-} and [•]OH, respectively, within mitochondria, the above results suggested the
272 increases in mitochondrial ROS (mROS). To further test this view, we analyzed the cellular location
273 of [•]OH. In untreated cells, the oxidant has broadly distributed the cytoplasm. After ODM
274 treatment, the oxidant was seen at one side of the perinuclear sites. As a result, it colocalized with
275 mitochondria, and catalase abolished the changes (Fig. 4D). These results show that ODM
276 increases mitochondrial oxidative stress in an H₂O₂-dependent manner.

277 **3.5. Reduced H₂O₂ removal augments ODM-induced H₂O₂ increase and cell death in 278 tolerant cells**

279 We noticed that the effect of ODM decreased as cell density increased. As a result, HOC-313 cells
280 grown at a high density became highly resistant to treatment with ODM (50%) (Fig. 5A). As the
281 above results suggested that H₂O₂ was critical in oxidative cell death, we assumed the resistance
282 might be related to H₂O₂-scavenging activity. To test this hypothesis, we examined the impact of
283 agents affecting the activity on ODM cytotoxicity. The glutathione (GSH) synthase inhibitor BSO
284 has been shown to reduce GSH synthesis and GSH-dependent antioxidant systems, including
285 GSH peroxidase (GPX), the primary H₂O₂-scavenging enzyme [30, 31]. On the other hand, 3-AT,
286 an irreversible catalase inhibitor, can block H₂O₂ removal by catalase. BSO had moderate

287 cytotoxicity and markedly augmented growth inhibition and cell death caused by ODM (Fig. 5A,
288 B). BSO also synergistically increased intracellular H₂O₂ with ODM (Fig. 5C). 3-AT also augmented
289 the growth inhibitory effect to a lesser extent. These results suggest that the resistance to ODM
290 may be due to increased H₂O₂ removal.

291 **3.6. ODM induces MPMC and microtubule remodeling in an H₂O₂-dependent manner**

292 In untreated cells, most mitochondria belong to Type I or II. Following ODM treatment, they
293 became Type III (Fig. 6A). These morphological changes occurred as rapidly as within 2 h after
294 ODM treatment. Like ODM cytotoxicity, catalase prevented the effect. Quantitative analyses of
295 Type I, II, and III mitochondria ratios confirmed these observations (Fig. 6B). Similar
296 morphological changes were observed in SAS, HOS, and U251MG (Supplementary Fig. S1A–C).
297 The above results suggested the onset of mitochondrial movement before the distribution
298 change. Since mitochondria can move along the microtubule track, we analyzed the impact of
299 ODM on tubulin dynamics. In untreated cells, tubulin had a network structure distributing
300 broadly on both sides of the nuclei in the cytoplasm (Pan-cytoplasmic) (Fig. 6C). On the other
301 hand, after ODM treatment, it condensed and assembled one side of the nuclei (Perinuclear).
302 Catalase also prevented the effect (Fig. 6C). Similarly, the microtubule inhibitor Nocodazole (NC)
303 prevented MPMC and the distribution shift (Supplementary Fig. S1D). These results show that
304 ODM induces MPMC and microtubule remodeling in an H₂O₂-dependent manner.

305 **3.7. ODM causes minimal MPMC, ROS production, and cell death in non-malignant cells**

306 Next, we examined whether ODM had cytotoxicity in non-malignant cells. ODM (<25%) had
307 minimal effect on the viability of HaCaT and hFOB, the non-transformed counterparts of OC and
308 OS, respectively (Fig. 7A, B). Similar results were obtained in HDFs (not shown). At a higher
309 concentration (50%), ODM decreased cell viability moderately (around 50%). Moreover, minimal
310 cell and nuclear morphological changes were observed after ODM treatment in HaCaT and HDFs
311 (Fig. 7C, D). ODM had a minimal effect on mitochondrial morphology in HaCaT while increasing
312 mitochondrial fragmentation but evoked MPMC minimally in HDFs (Fig. 7C, D). In addition, ODM
313 minimally increased intracellular H₂O₂ and [•]OH in HaCaT cells (Supplementary Fig. S2). These
314 results indicate that ODM causes minimal MPMC, ROS production, and cell death in
315 non-malignant cells.

316 **4. Discussion**

317 The present study aimed to examine the antitumor activity of dO₃ and verify its role in mediating
318 the action of APAM. As expected, ODM had considerable amounts of O₃ (Fig. 1) and potent
319 cytotoxicity in various cancer cells, including OC, OS, and GBM (Fig. 2). Moreover, it primarily
320 triggered oxidative cell death in OC cells. Meanwhile, apoptosis explicitly participated in the
321 moderate effect (Fig. 3). Thus, apoptosis induction may occur under limited conditions and may
322 be insufficient for killing OC cells effectively. Oxidative cell death may be essential for efficient OC
323 killing. In line with the role of oxidative stress, ODM increased mitochondrial ROS, such as O₂^{•-}
324 and [•]OH (Fig. 4). catalase prevented the increases while reduced H₂O₂-removal upregulated them
325 (Fig. 4 and 5). These findings indicate that H₂O₂ mediates oxidative stress. These results are
326 similar to those obtained with APAM previously described [28]. Moreover, APAM had dO₃ as
327 much as ODM (Fig. 1). APAM has potent antitumor activity against different cancer cell lines.
328 More importantly, it acts on cancer cells preferentially [28]. Notably, our data indicate that ODM

329 has similar tumor-selective action. ODM showed little cytotoxicity against non-malignant cells
330 and minimally increased MPMC and ROS production (Fig. 7). Our results indicate that dO₃ has
331 tumor-selective cytotoxicity and support the view that dO₃ mediates the action of APAM. In
332 addition, ODM had NO₂⁻ and NO₃⁻ below the detection limit (Fig. 1), indicating that the effect is
333 independent of NO_x.

334 It is known that dO₃ can directly generate H₂O₂ in the reaction with H₂O. Our previous study
335 demonstrated that H₂O₂ increased mitochondrial O₂^{•-} via the reduction of the electron transport
336 chain in different cancer cells [32]. In addition, H₂O₂ also can generate [•]OH in reacting with iron
337 (II) or copper (I). Collectively, H₂O₂ produced from dO₃ can trigger mitochondrial oxidative stress.
338 Another finding that GSH synthesis inhibition sensitized tolerant cells to ODM (Fig. 5) supports
339 this view. These observations suggest that the GSH-dependent H₂O₂-removal mechanism,
340 possibly GPXs, is vital in removing H₂O₂ after dO₃ treatment. The inhibition of catalase activity by
341 3-AT increased mitochondrial ROS but minimally affected cell survival (Fig. 4 and 5), suggesting
342 the requirement of another event to cause cell death in tolerant cells. Ferroptosis has emerged as
343 a critical oxidative cell death mode in cancer cells. It is regulated necrosis resulting from LPO
344 accumulation where GPX and iron are crucial (for a recent review, see [33]). ODM could increase
345 [•]OH, the initiator of lipid peroxidation (Fig. 4). In addition, preliminary results showed that iron
346 chelation partially prevented the effect, while iron (II) addition augmented it. Therefore, [•]OH
347 generation from H₂O₂ and iron (II) via the Fenton reaction might play a role in the production.
348 This view favors the notion that besides H₂O₂, other factors are essential for sufficient cell death.
349 Our data indicate similarities and differences between cell death caused by ODM and ferroptosis.
350 The possible involvement of GPX4 resembles ferroptosis. On the other hand, the failure of Fer-1
351 in protecting cells from ODM (Fig. 3) suggests the role of different mechanisms and lipid radicals.
352 Further characterization, including the role of several master ferroptosis regulators and genes, is
353 underway.

354 Our results showed that similar to APAM [28], ODM could cause MPMC and tubulin remodeling
355 (Fig. 6). The findings provide evidence for the role of dO₃ in the action of APAM. Notably, catalase
356 blocked these two events, indicating the critical role of H₂O₂ in regulating MPMC and tubulin
357 remodeling. MPMC may involve multiple sequential cellular events, mitochondrial fragmentation,
358 movement, and assembly. In line with the role of H₂O₂, we previously demonstrated that H₂O₂
359 could cause mitochondrial fragmentation in MM cells via increased mitochondrial O₂^{•-} [34] (Saito
360 et al., 2016). This event was associated with increased phosphorylation of Drp1 at Ser 616, the
361 driving signal of mitochondrial fission. Similar mechanisms might be involved in the effect of
362 ODM. Tubulin polymerization is critical in altered mitochondrial distribution from Type I to Type II
363 in response to stresses, such as hypoxia and heat shock [26, 27]. This event is required for the
364 mitochondrial movement along the microtubule track by motor proteins such as Kinesin and
365 Dynein. Our data indicate that catalase (Fig. 6) and NC (Supplementary Fig. S1D) can prevent
366 tubulin remodeling and MPMC caused by ODM, suggesting the involvement of H₂O₂ and tubulin
367 polymerization in the mitochondrial movement. Besides mitochondrial oxidative stress, plasma
368 membrane depolarization (PMD) is essential for mitochondrial assembly [35]. Notably, dO₃ can
369 affect biological systems in several different forms. It can react with target biomaterials directly
370 and specifically as O₃ itself. It can also attack poly unsaturated fatty acids (PUFAs) in the cell
371 membrane, resulting in lipid oxides. The production follows PUFA peroxidation and the
372 production of other toxic radicals and substances, such as LO[•], LOO[•], lipo hydroperoxides (LOOH),
373 4-hydroxy-2,3 trans-noneal (HNE), and malonyl dialdehyde (MDA) [13]. These radicals and

374 aldehydes are toxic and could contribute to ODM cytotoxicity. Notably, linoleic acid
375 hydroperoxide can evoke the depolarization of plasma membrane and mitochondrial membrane
376 potentials [36]. Therefore, LPOs like lipid hydroperoxides might trigger PMD and mitochondrial
377 assembly. Further studies to explore this possibility are ongoing.

378 Several prior studies have implicated the role of O₃ in the antitumor effect of PTLs. Mokhtari and
379 colleagues [37] have shown the production of O₃ in plasma-activated media and the correlation
380 between the amount and antitumor activity. However, this report lacks evidence for the oxidant's
381 role in the media's action. Lunov et al. [38, 39] have demonstrated that O₃ is an abundant
382 component of air plasma and that O₃ gas flush induces necrosis. Their findings are similar to the
383 present findings, while there are some discrepancies between our results and theirs. The authors
384 reported the highest toxicity of O₃ gas for non-malignant cells. In contrast, dO₃ had little toxicity
385 in the present study (Fig. 7). The report also demonstrated that O₃ gas caused mitochondrial
386 permeability transition. However, our preliminary results showed that dO₃ had no such effect. A
387 possible explanation for the discrepancies may be the different biological effects between O₃ gas
388 and dO₃. As described above, dO₃ can exhibit its physical impacts by acting in various forms
389 other than O₃.

390 In summary, this study shows that dO₃ has tumor-selective cytotoxicity primarily via oxidative cell
391 death independently of NO_x. Our results suggest that mitochondrial ROS and the resulting
392 tubulin remodeling and MPMC play a vital role in the action. Also, our data support the view that
393 dO₃ is a critical mediator of the action of APAM. Thus, ODM could be a more chemically-defined
394 alternative to PTLs in cancer treatment.

395 **Figure legends**

396 **Figure 1. ODM has dO₃ but not NO₂⁻/NO₃⁻.** (A, B) Schematic O₃-generator and bubbling system
397 diagrams. In method A, O₃ was produced by irradiating an excimer light (185 nm) into the air
398 using a UV-C lamp (A). In method B, O₃ was generated by exciting high-purity (99.9%) molecular
399 oxygen (O₂) with dielectric barrier discharge (B). The resulting O₃ was then introduced into a
400 DMEM by bubbling. (C) Quantitation of dissolved O₃ (dO₃) in ODM and APAM. The amount of
401 dO₃ in ODM and APAM (12.5–50% solution) was measured as described in Materials. Data are the
402 mean ± SD (n=4). (D) The concentrations of NO₂⁻ and NO₃⁻ (C) in ODM (6.3–50% solution) were
403 measured using the Griess method. APAM was used as a positive control. NO₃⁻ concentration was
404 calculated following the formula; [NO₃⁻] = [(NO₂⁻ + NO₃⁻)] - [NO₂⁻]. Data are the mean ± SD
405 (n=4).

406 **Figure 2. ODM can injure different cancer cell lines.** (A–D) SAS (A), HOC-313 (B), HOS (C), and
407 U251MG (D) cells in FBS/DMEM were plated at 4 × 10³ cells/well and incubated overnight. The
408 cells were treated with ODM (12.5–50% solution) for 72 h and analyzed for growth using a WST-8
409 cell growth assay. Data are the mean ± SD (n =8). Data were analyzed by one-way analysis of
410 variance followed by Tukey's post hoc test. **P < 0.01; ***P < 0.001; NS, not significant vs. control
411 treated with vehicle. (E, F) SAS (E) and HOS (F) cells (1.5×10⁴ cells/well) cells/well were plated in a
412 35 mm glass bottom dish and incubated overnight, and then treated with ODM (50%) for 2 h.
413 After removing the medium by aspiration, the cells were stained with Hoechst33342 (Hoe).
414 Images were obtained from BZ-X 710 Digital Biological Microscope equipped with a 100x
415 objective and analyzed using BZ-H3A application software. PC, phase contrast. Bar = 10 μm.

416 **Figure 3. ODM induces oxidative cell death in an H₂O₂-dependent manner.** (A, B) Effect of
417 cell death inhibitors on ODM cytotoxicity. SAS (A) and HOC-313 (B) cells were processed as
418 described in the legend of Figure 2. The cells were incubated with 10 μM Z-VAD-FMK,
419 Ferostatin-1 (Fer-1), or 30 μM Necrostatin-1 (Nec-1) for 1 h and then treated with ODM (50%) for
420 72 h. The cell growth was measured as described in the legend of Figure 2. (C) HOC-313 cells
421 were incubated with 10 μM Z-VAD-FMK or 10 U/ml catalase for 1 h and then treated with ODM
422 (50%) for 72 h. The cell growth was measured as described in the legend of Figure 2. Data are the
423 mean ± SD (n = 8). Data were analyzed by one-way analysis of variance followed by Tukey's post
424 hoc test. ****P* < 0.001 vs. control treated with vehicle. ### *P* < 0.001. (D, E) SAS (D) and HOC-313
425 (E) cells were incubated with 10 U/ml catalase or 30 μM MnTBaP for 1 h and then treated with
426 ODM (50%) for 72 h. (F) SAS cells were incubated with 10 U/ml catalase for 1 h and then treated
427 with APAM (25, 50%) for 72 h. The cell growth was measured as described in the legend of Figure
428 2. Data are the mean ± SD (n = 8). Data were analyzed by one-way analysis of variance followed
429 by Tukey's post hoc test. ****P* < 0.001 vs. control treated with vehicle. ### *P* < 0.001.

430 **Figure 4. ODM increases different intracellular ROS in an H₂O₂-dependent manner.** (A)
431 HOC-313 cells were preincubated with 3-amino-1,2,4-triazole (3-AT, 1 mM) and catalase for 1h
432 and then treated with ODM (50%) for 2h. Mitochondrial reactive oxygen species (ROS) and nuclei
433 were stained using specific probes MitoSOX and Hoechst33342, respectively. All images were
434 obtained from BZ-X 710 Digital Biological Microscope with a uniform exposure time (1/2.5S). (B)
435 The luminescence of all color images was measured using a public-domain Java NIH ImageJ
436 program. Data are the mean ± SD (n=9). Data were analyzed by own-way analysis of variance
437 followed by Tuckey's post hoc test. ####*P* < 0.001. (C) The cells were pretreated with catalase for
438 1h and then incubated with ODM for 2h. Intracellular hydroxyl radicals were stained with OXO.
439 The nuclei were counterstained with Hoechst33342. The luminescence of all color images was
440 analyzed as described above. Data are the mean ± SD (n=8). Data were analyzed by own-way
441 analysis of variance followed by Tuckey's post hoc test.****P* < 0.001 vs. control. ####*P* < 0.001. Bar
442 = 300 μm. (D) HOC-313 cells were pretreated with catalase for 1h and then incubated with ODM
443 for 2h. Mitochondria, intracellular hydroxyl radicals, and nuclei were stained with MTG, OXO, and
444 Hoechst33342. Bar = 20 μm.

445 **Figure 5. BSO augments ODM-induced H₂O₂ increase and cell death in insensitive cells.** (A)
446 HOC-313 cells were pretreated with the DL-Buthionine-(S, R)-sulfoximine (BSO, 100 μM) or 3-AT
447 (1 mM) alone or in combination for 1h. then they were treated with the ODM (50%) for 72h. Cells
448 were analyzed for viability as described in the legend of Figure 3. Data were analyzed by
449 own-way analysis of variance followed by Tuckey's post hoc test. Data are the mean ± SD (n=7 or
450 8). ***P* < 0.01; NS, not significant vs. control; ####*P* < 0.001. (B) The cells were pretreated with BSO
451 for 1h and then treated with ODM (50%). They were incubated for 18h and analyzed for cell
452 death using Calcein-AM (Calcein, 2 μM) and ethidium bromide-1 (EthD-1, 2 μM). Live cells were
453 stained green with Calcein, whereas dead/dying cells were stained red with EthD-1. Images were
454 obtained from BZ-X 710 Digital Biological Microscope with a 4x objective. Bar = 300 μm. (C) The
455 cells were treated with the ODM (50%) or BSO alone or in combination for 18h and were stained
456 with HYDROP™ (Hydrop,1μM). Images were obtained from BZ-X 710 Digital Biological
457 Microscope with a 40x objective and analyzed as previously described in the legend of Figure 4.
458 Bar = 20 μm.

459 **Figure 6. ODM induces MPMC and tubulin remodeling in an H₂O₂-dependent manner.** (A)
460 After treating HOC-313 cells with medium or ODM (50 %) for 2h, mitochondria and nuclei were

461 stained with 20 nM MTR and Hoechst33342, respectively. Images were obtained from BZ-X 710
462 Digital Biological Microscope with a 100x objective and analyzed using BZ-H3A application
463 software. Bar =10 μ m. (B) Mitochondria exhibiting three different subcellular distributions,
464 Pan-cytoplasmic (Type I), PNMC (Type II), and MPMC (Type III), were counted in two or three
465 pictures. Data are the mean \pm SD (n= \sim 20). (C) The cells were treated with ODM (50%) for 2h and
466 stained with Tubulin Tracker™ Green (TTG, 100 nM) for 30 min. Bar =10 μ m. (D) Tubulin exhibiting
467 Pan-cytoplasmic or Perinuclear distribution was counted in two or three pictures. Data are the
468 mean \pm SD (n= \sim 20).

469 **Figure 7. ODM has low cytotoxicity in non-malignant cells.** (A, B) HaCaT (A) and hFOB (C)
470 cells were treated with ODM at the indicated concentration for 72h and were measured for
471 viability by a WST-8 assay. Data were analyzed by own-way analysis of variance followed by
472 Tuckey's post hoc test. ****P** < 0.01; NS, not significant vs. control. (C, D) HaCaT (C) and HDFs (D)
473 were treated with ODM (50%) for 2h. After stimulation, cells were stained with MTR and
474 Hoechst33342. Images were obtained and analyzed as described in the legend of Figure 6. Bar
475 =10 μ m. Mitochondria exhibiting three different subcellular distributions (Type I, Type II, Type III)
476 were counted in two or three pictures. Data are the mean \pm SD (n= \sim 20).

477 **Supplementary Figure legends**

478 **Figure S1. ODM induces MPMC and tubulin remodeling in different tumor cells via**
479 **microtubules.** (A–C) SAS (A), HOS (B), and U251MG cells (C) were treated with medium or ODM
480 (50 %) for 2h, and mitochondria and nuclei were stained with 20 nM MTR and Hoechst33342,
481 respectively. Images were obtained from BZ-X 710 Digital Biological Microscope with a 100x
482 objective and analyzed. Bar =10 μ m. (D) HOC-313 cells were treated with ODM (50%) for 2h and
483 stained with MTR, TTG, and Hoechst33342. Images were taken and analyzed as described in the
484 legend of Figure 6. Bar =10 μ m.

485 **Figure S2. ODM minimally increases intracellular ROS in non-malignant cells.** HaCaT cells
486 were treated with ODM (50%) for 2h. After stimulation, H₂O₂ and hydroxyl radicals were stained
487 with HYDROP™ (Hydrop,1 μ M) and OxiORANGE (OXO, 1 μ M), respectively. The nuclei were
488 stained with Hoechst33342. Images were obtained from BZ-X 710 Digital Biological Microscope
489 with a 40x objective and analyzed as described above. Bar = 20 μ m.

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495 Suzuki-Karasaki), S.I. H.O. investigation, M. S.-K. (Manami Suzuki-Karasaki), M. S.-K. (Miki
496 Suzuki-Karasaki), Y.O., S.I., H.O. Data acquisition and analysis, M. S.-K. (Manami Suzuki-Karasaki),
497 M. S.-K. (Miki Suzuki-Karasaki), Y.O., S.I., H.O. Funding acquisition, Y. S.-K., M. S.-K. (Manami
498 Suzuki-Karasaki). Methodology, S.I., H.O. Visualization, Y. S.-K., M. S.-K. (Manami Suzuki-Karasaki).
499 Project administration, Y. S.-K., S.I., H.O. Resources, S.I., H.O., H.N. writing–original draft
500 preparation, Y. S.-K., M. S.-K. (Manami Suzuki-Karasaki). writing–review and editing, Y. S.-K., M.
501 S.-K. (Manami Suzuki-Karasaki), H.N. supervision, Y. S.-K., H.N.

502 **Declaration of competing interest**

503 Manami Suzuki-Karasaki, Miki Suzuki-Karasaki, and Dr. Yoshihiro Suzuki-Karasaki are employees
504 of the Non-Profit Research Institute Plasma ChemiBio Laboratory. Other authors have no conflicts
505 of interest. The funders had no role in the study's design, in the collection, analyses, or
506 interpretation of data, in the writing of the manuscript, or in the decision to publish the results.

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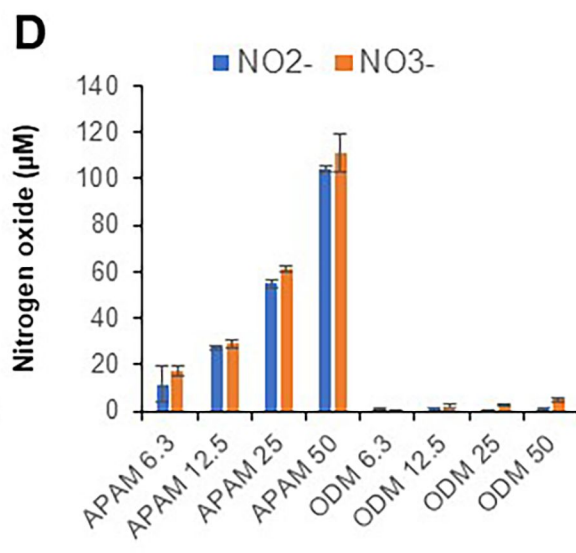
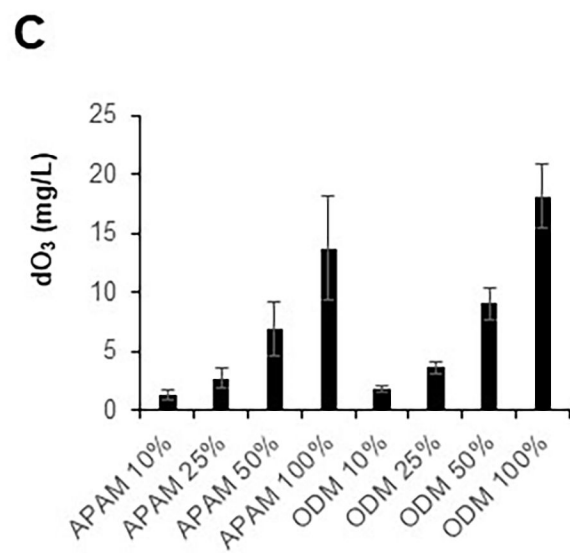
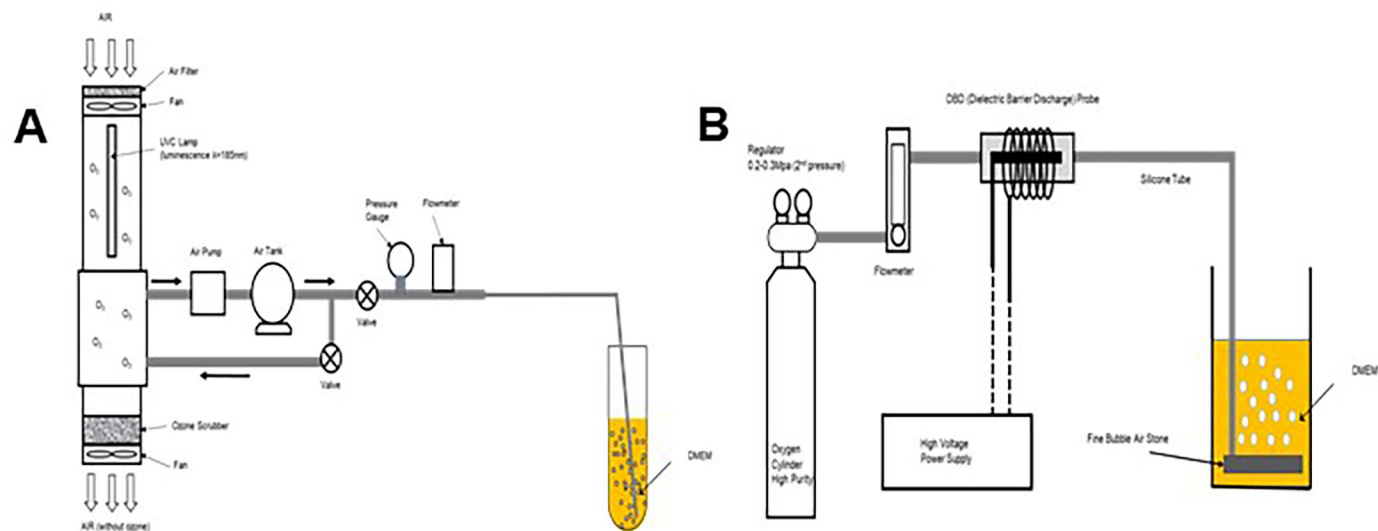
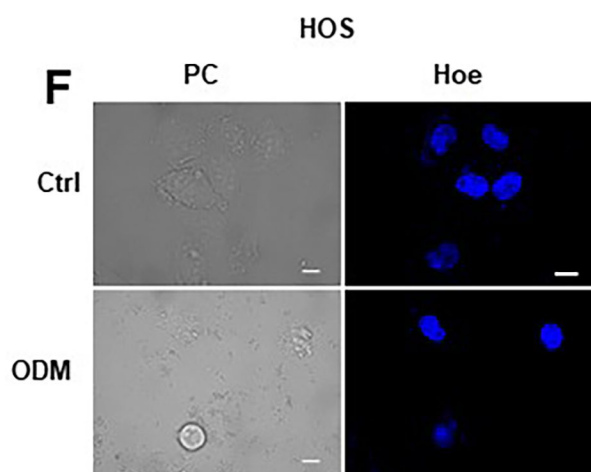
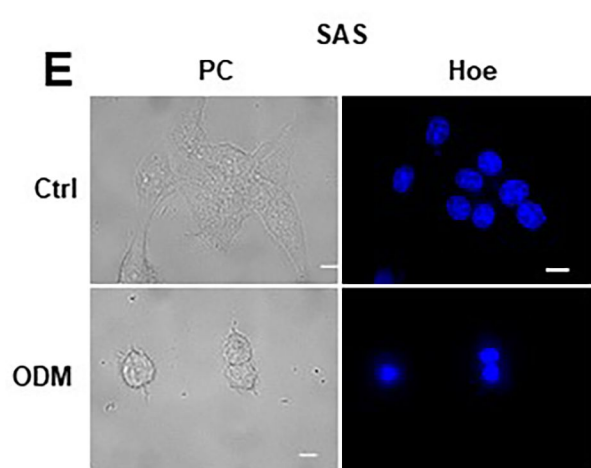
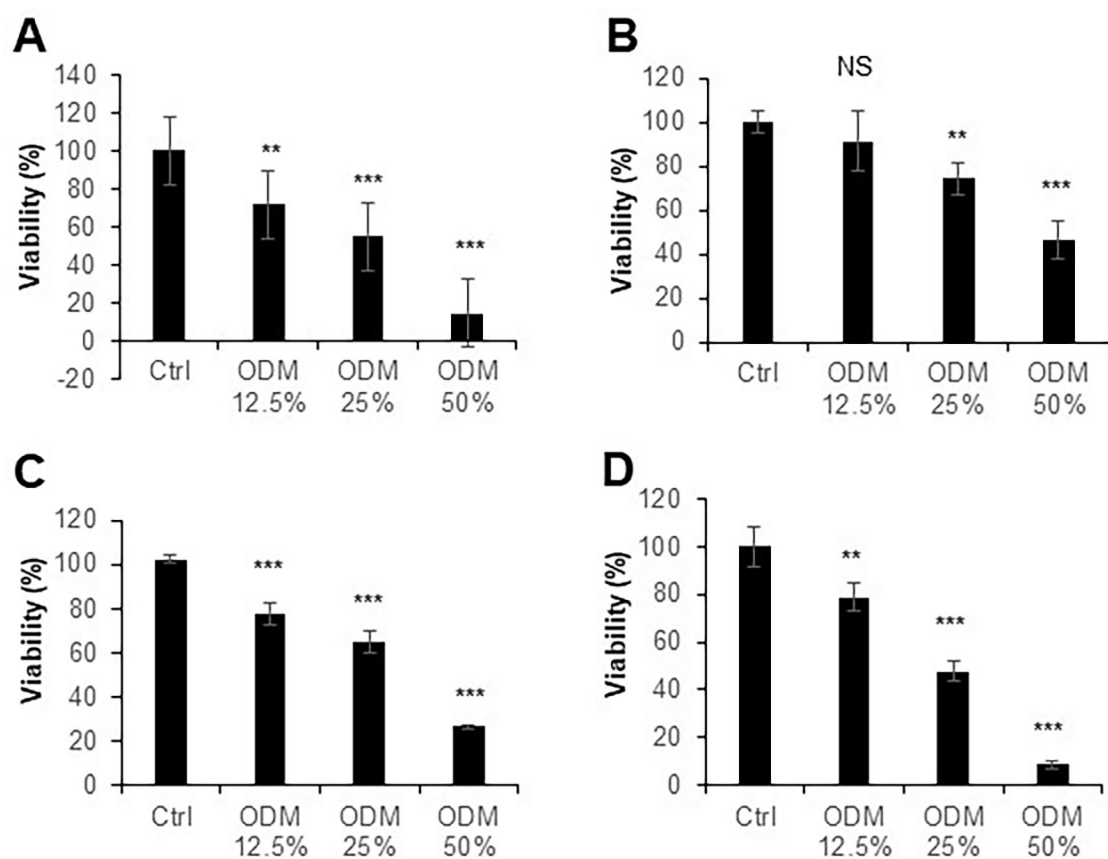
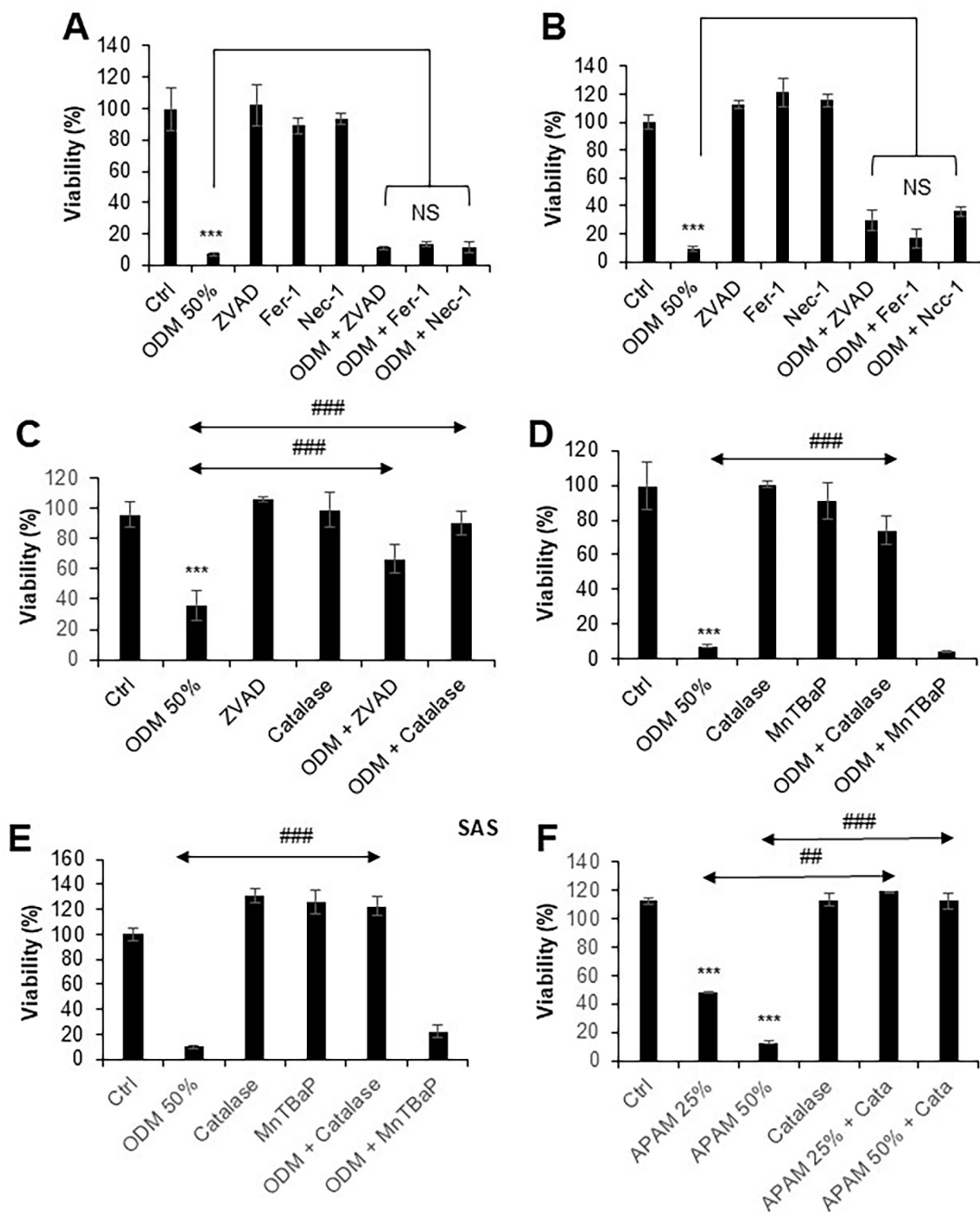


Figure 2. ODM can injure different cancer cell lines

Figure 2





**Figure 4
(Continued)**

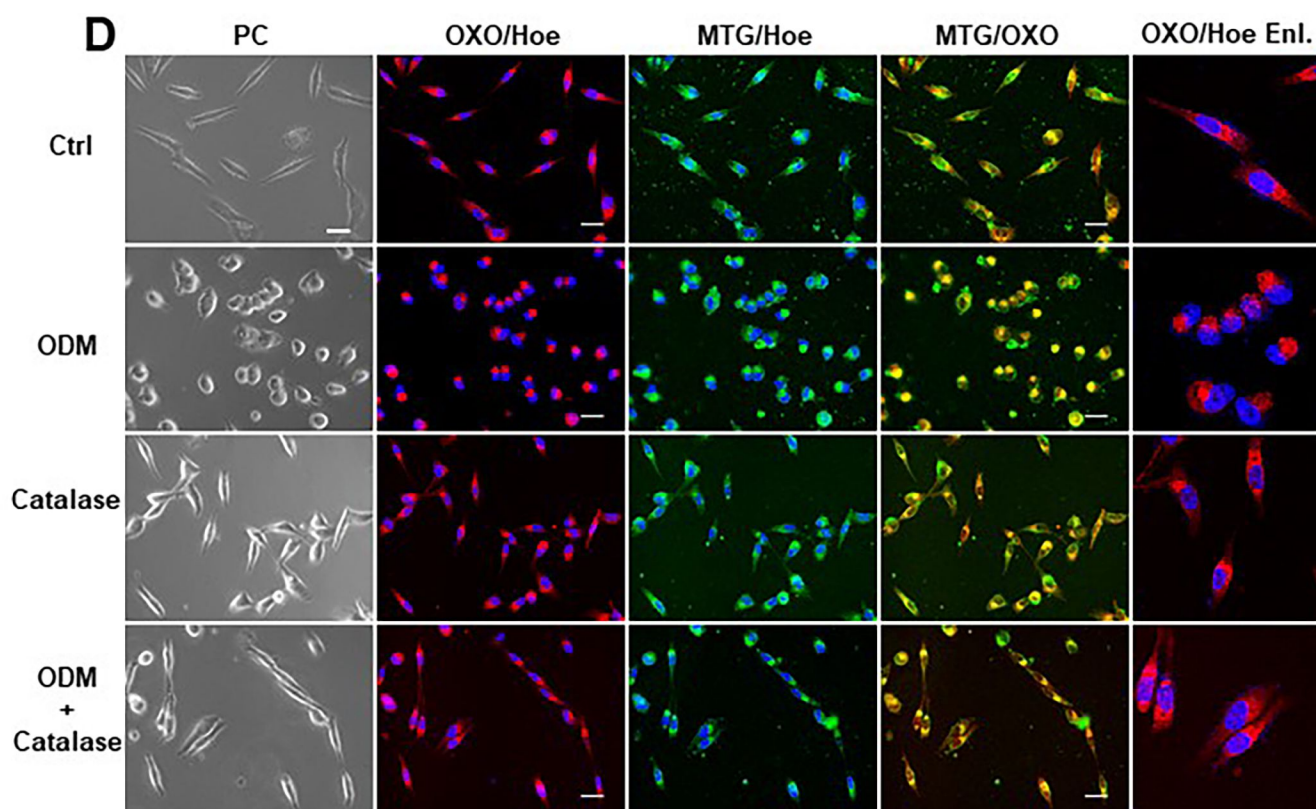
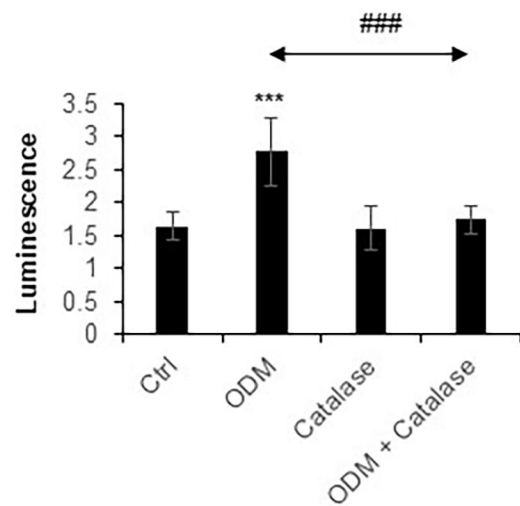
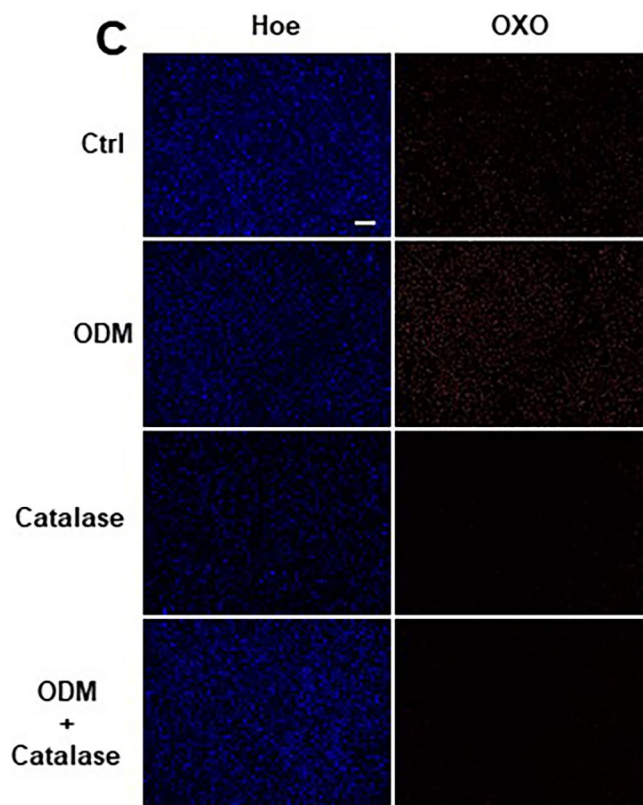


Figure 4. ODM increases different intracellular ROS in an H₂O₂-dependent manner

Figure 4

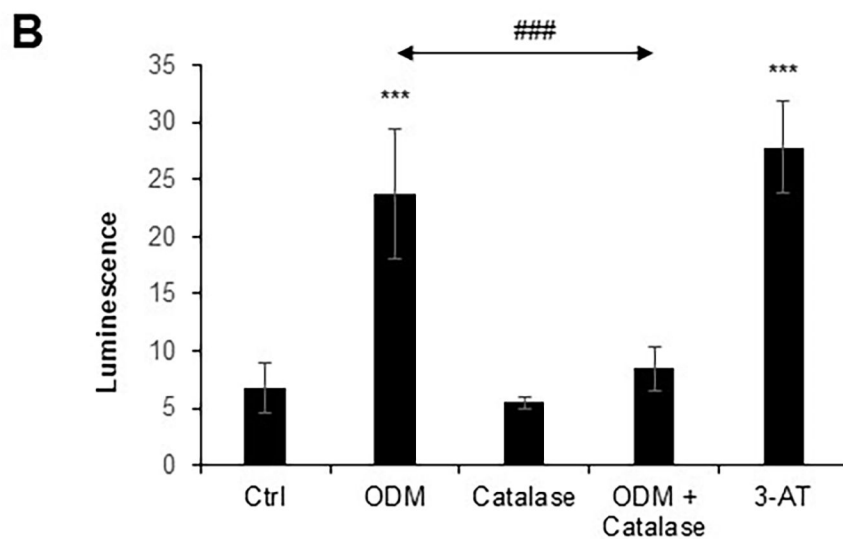
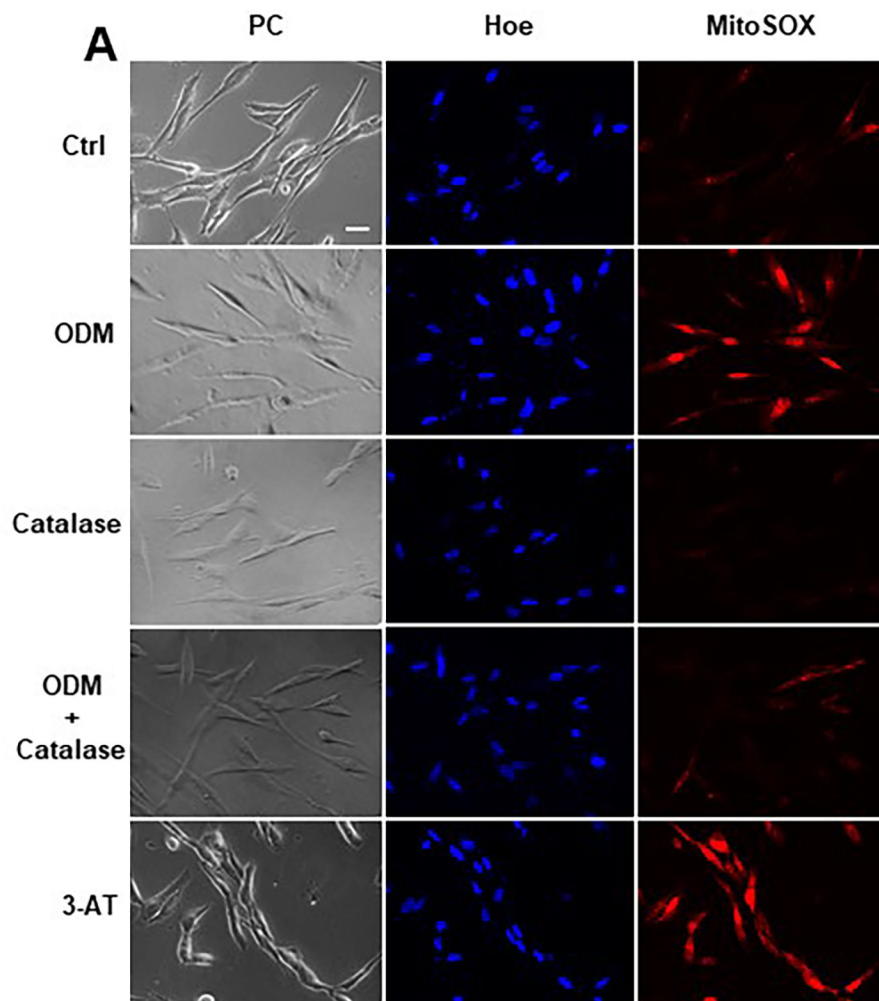


Figure 5. BSO augments ODM-induced H₂O₂ increase and sensitizes to cell death in insensitive cells

Figure 5

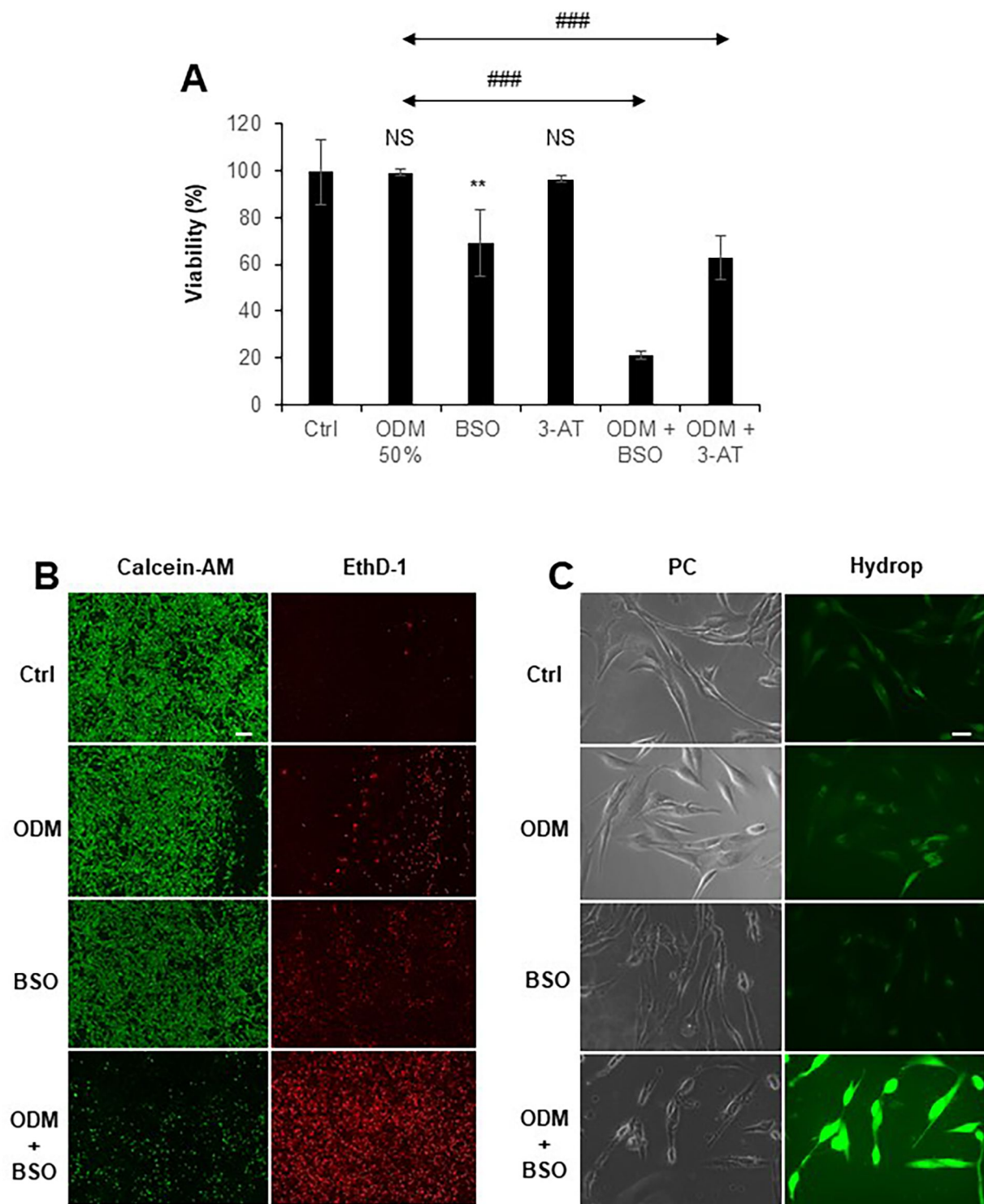


Figure 6. ODM induces MPMC and tubulin remodeling in H₂O₂-dependent manner

Figure 6

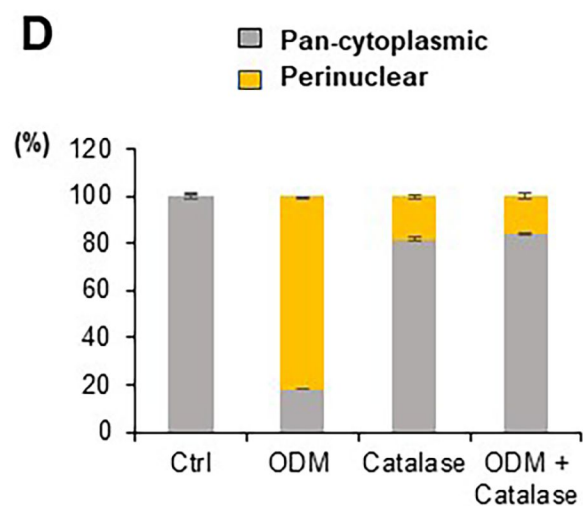
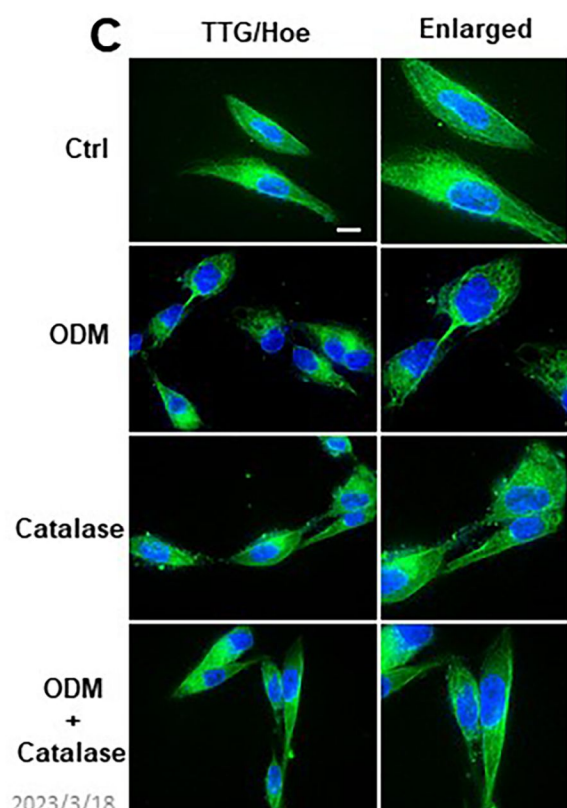
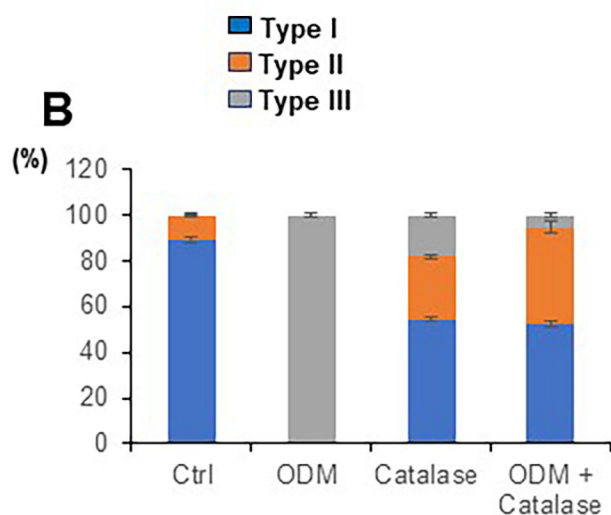
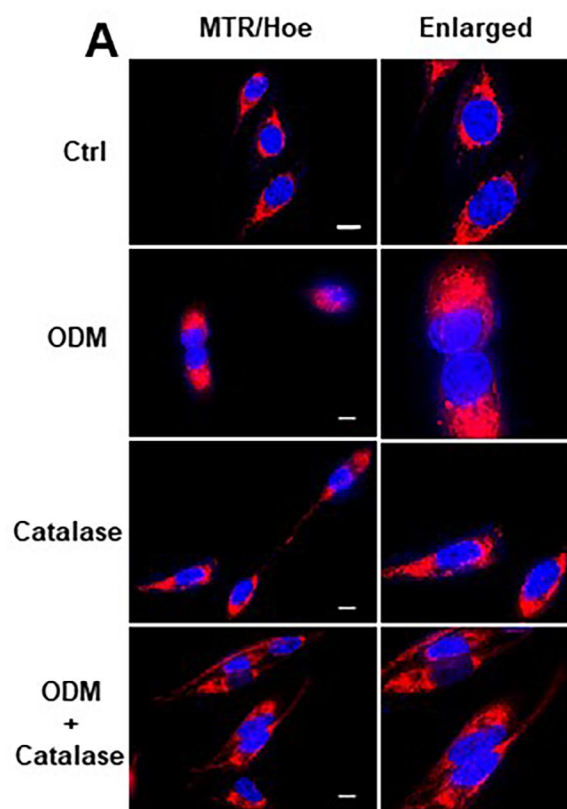


Figure 7. ODM has low cytotoxicity in nonmalignant cells

Figure 7

