1	Detergent headgroups control ToIC folding in vitro
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17 Abstract

ToIC is the trimeric outer membrane component of the efflux pump system in E. coli 18 19 responsible for antibiotic efflux from bacterial cells. Over-expression of efflux pumps has been 20 reported to decrease susceptibility to antibiotics in a variety of bacterial pathogens. Reliable 21 production of membrane proteins allows for the biophysical and structural characterization 22 needed to better understand efflux and for the development of therapeutics. Preparation of 23 recombinant protein for biochemical/structural studies often involves the production of proteins 24 as inclusion body aggregates from which bioactive proteins are recovered. Here we find that the 25 in vitro folding of ToIC into its functional trimeric state from inclusion bodies is dependent on the headgroup composition of detergent micelles used. Nonionic detergent favors the formation of 26 27 functional trimeric ToIC, whereas zwitterionic detergents induce the formation of a non-native 28 trimeric ToIC fold. We also find that nonionic detergents with shorter alkyl lengths facilitate ToIC 29 folding. It remains to be seen whether the charges in lipid headgroups have similar effects on 30 membrane insertion and folding in biological systems.

31 Introduction

The Gram-negative bacterial cell envelope contains a distinctive extra protective layer called the outer membrane. The outer membrane not only acts as a protective layer to the cell but also functions as a selective permeability barrier controlling the transport of molecules in and out of the cell (1,2). The outer membrane of Gram-negative bacteria is an asymmetric bilayer composed of phospholipids in the inner leaflet, lipopolysaccharide in the outer leaflet, and outer membrane proteins (OMPs) spanning the bilayer (3).

38 OMPs are almost exclusively antiparallel beta-barrels. They direct diverse cellular 39 functions including signal transduction, general and substrate-specific transport, and enzymatic 40 catalysis (4). OMPs are synthesized in the cytoplasm and targeted to the Sec translocon; a

protein complex that helps to transport OMPs across the inner membrane (5). To traverse the
periplasmic space, OMPs are aided by periplasmic chaperones (6-9). The folding of the majority
of OMPs requires the BAM machinery which catalyzes insertion into the outer membrane (1014).

For biochemical and structural studies which often require a high concentration of proteins, isolating OMPs from their native environment is extremely labor-intensive. This is because the small volume of the outer membrane results in low-yield protein extraction. When possible, refolding of recombinant protein is preferred.

Preparation of recombinant outer membrane protein for biophysical or structural studies often involves production of proteins as inclusion bodies, aggregates of unfolded, misfolded, and partially folded protein, from which bioactive proteins are recovered (15). When the signal sequence is removed, single chain outer membrane proteins are known to fold from solubilized inclusion bodies into membrane mimetics like detergent micelles (16,17) or lipid bilayers (18,19). When refolding, the selection of detergent or lipid is often chosen empirically or involves screening mimetics with various chemical properties to achieve proper folding.

56 Previous studies have shown differential effects of various features of membrane 57 mimetics on protein folding. OMP folding rates increase in lipids that form thinner bilayers 58 (20,21), bilayers of increased curvature (21), and bilayers with heat-shock induced lipid 59 dynamics (22,23).

Different types of detergents have been employed in the solubilization of membrane proteins for *in vitro* structural and biophysical studies. Although nonionic detergents are the most commonly used detergent in studying membrane proteins, zwitterionic detergents are also frequently employed. Of the zwitterionic detergents used, LDAO is the most common for biochemical and structural analysis of OMPs (24). OMPs that have been studied with LDAO

range from small 8-stranded OMPs, OmpW and Ail (25,26), 12-stranded autotransporter, EstA
(27),14-stranded long-chain fatty acid transporter, FadL (28),16-stranded OMPs sortase, BamA
(29) to the multi-barrel porin, OmpF, with three 16-stranded barrels, (30)

68 Outer membrane efflux pumps present a unique challenge in refolding since they are 69 composed of three separate chains. We recently found that with an additional concentration 70 step, multimeric outer membrane proteins such as ToIC can be folded into membrane mimetics 71 like single chain barrels are (31). ToIC is an outer membrane protein that serves as a 72 component of the tripartite resistance-nodulation-cell-division (RND) efflux pump in E. coli, 73 responsible for the expulsion of toxins, including antibiotics, from bacterial cells (32,33). 74 Tripartite RND efflux pumps are composed of three protein subunits that span the inner and 75 outer membranes – where energy from the proton motive force of the inner membrane pushes 76 small hydrophobic molecules out through the periplasmic adapter protein, and then the outer 77 membrane channel (34). The assembly of these three subunits of this pump facilitates extrusion 78 of antibiotics and other toxins from bacterial cells (35). Expression of the AcrABToIC efflux 79 pump is directly correlated with antibiotic resistance in clinical isolates as well as in vitro 80 selected mutants of different bacterial pathogens (36-38).

The outer membrane pore of TolC is composed of three identical chains (**Figure 1A**) and the folding of TolC into its native trimeric state is essential for the protein to engage the periplasmic subunit of the efflux pump. Because of the prominent role that TolC plays in efflux pump mediated antibiotic resistance, understanding the factors that promote or inhibit the assembly of functional TolC is important for developing inhibitors of efflux pump assembly.

In this study, we use *in vitro* refolding experiments and biophysical and biochemical assays to identify a differential effect of detergent headgroup composition on ToIC folding *in vitro*. We find that ToIC only folds correctly in nonionic detergents and that ToIC stability is controlled by the particular nonionic headgroup. In contrast, we find that zwitterionic detergents

90 induce the formation of non-native and non-functional trimeric ToIC during *in vitro* refolding from91 inclusion bodies.

92 **Results**

Different types of detergents have been employed in the refolding and solubilization of
membrane proteins for *in vitro* structural and functional studies. Although nonionic detergents
like n-dodecyl β-D-maltopyranoside (DDM) are most commonly used detergents for membrane
protein purification and structural studies, zwitterionic detergent, n-dodecyl-N,N-dimethylamineN-oxide (LDAO) is among the top six detergents used for membrane protein purification and

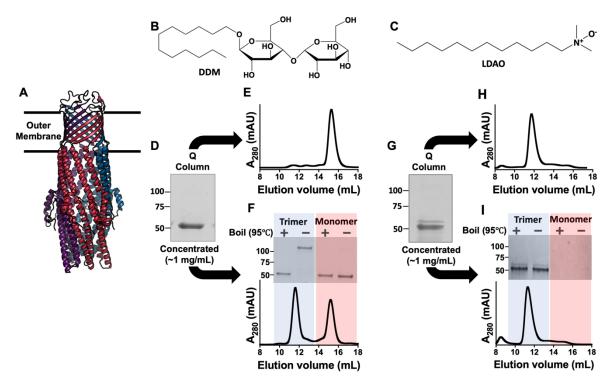


Figure 1: ToIC refolded in DDM and LDAO detergents micelles. (A) Structure of trimeric ToIC (PDB ID: 1EK9). (B, C) Molecular structure of (A) non-ionic DDM and (B) zwitterionic LDAO. (D) SDS PAGE of ToIC in DDM micelles by rapid dilution followed by ion exchange chromatography. (E) SEC chromatogram of ToIC elution fraction from ion exchange column in DDM micelles. (F) *Bottom:* SEC chromatogram of concentrated ToIC eluate from ion exchange column in DDM micelles. *Top:* SDS PAGE of ToIC SEC fractions. (G) SDS PAGE of ToIC in LDAO micelles by rapid dilution followed by ion exchange chromatography. (H) SEC chromatogram of ToIC elution fraction from Q column in LDAO micelles. (I) *Bottom:* SEC chromatogram of concentrated ToIC eluate from ion exchange column in LDAO micelles. *Top:* SDS PAGE of SEC fractions.

98 crystallization (24). Ionic detergents may be more biologically relevant as phospholipids all have 99 at least one negative charge due to the phosphate in their headgroup. In addition the small 100 micellar size of zwitterionic LDAO is reported to be more favorable for protein crystallization 101 (39). DDM is a nonionic detergent with maltose constituting the headgroup and a 12-carbon 102 alkyl group making up the tail (Figure 1B). In contrast, LDAO is a zwitterionic detergent with an 103 amine oxide headgroup and a 12-carbon alkyl tail (Figure 1C). We have previously reported 104 that in vitro refolding of ToIC into DDM micelles from inclusion bodies requires three major 105 refolding steps: rapid dilution into detergent micelles, followed by anion exchange 106 chromatography, and then concentration. We found that the third step which involves 107 concentrating the elution fraction from the Q anionic column is essential for the formation of fully folded and functional trimeric ToIC (31). 108

109 Here we find that the choice of detergent can either cause or prevent ToIC folding into its 110 native, functional state. ToIC from urea-solubilized inclusion bodies was rapidly diluted into detergent and passed through a Q column. When the detergent was DDM, ToIC could then be 111 112 identified by sodium dodecyl sulfate polyacrylamide gel electrophoresis (SDS PAGE) at 113 approximately 51 kDa, corresponding to the size of a single polypeptide chain of ToIC (Figure 114 **1D**), consistent with previous reports (31,40,41). When this sample is assessed by size 115 exclusion chromatography (SEC), only a peak corresponding to monomeric TolC (about 15 ml) is visible (Figure 1E), consistent with previous report (31). When concentrated, ToIC trimerizes 116 as indicated by the appearance of a second SEC peak at about 11.5 ml (Figure 1F Bottom) 117 118 and a heat-modifiable, higher molecular weight band on SDS PAGE (Figure 1F, Top).

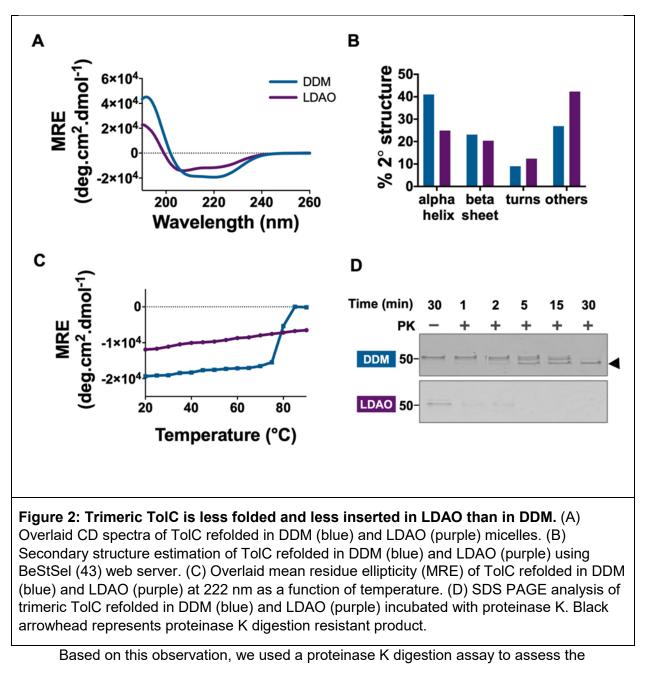
In contrast, when TolC from urea-solubilized inclusion bodies was diluted into LDAO, the protein is visible at approximately 51 kDa, corresponding to the size of a single polypeptide chain of TolC, on SDS PAGE (**Figure 1G**), and a peak corresponding to the molecular weight of trimeric TolC (about 11.5 ml) is visible by SEC (**Figure 1H**), consistent with previous reports of

SEC chromatogram of trimeric TolC (31,42). Upon subsequent concentration, with TolC in LDAO, we find no significant change in SEC (**Figure 1I**, *Bottom*) or SDS PAGE (**Figure 1I**, *Top*). This indicates that TolC can form a trimeric species, even in the absence of sample concentration. However, this trimeric species may be more unstable as suggested by the dissociation of the trimer on SDS-PAGE even without boiling (**Figure 1I**).

128 To better distinguish these two seemingly trimeric states of TolC, we used circular dichroism (CD) spectroscopy to quantify the amount of secondary structure formed in the 129 refolded protein. The mean residue ellipticity (MRE) spectrum reveals the presence of 130 131 secondary structural elements in ToIC refolded in both DDM and LDAO micelles (Figure 2A). 132 The MRE spectrum of ToIC refolded in DDM shows two troughs at about 210 nm and 222 nm and a peak at about 192 nm. For ToIC refolded in LDAO micelles, the MRE spectrum showed a 133 134 significantly shallower trough at about 222 nm compared to DDM micelles consistent with less 135 alpha helical content in LDAO. Furthermore, we observed a shift in the trough from ~210 nm in ToIC refolded DDM micelles to ~206 nm in ToIC refolded in LDAO micelles, consistent with the 136 137 presence of more random coil in LDAO micelles. Also, a general decrease in the signal intensity is observed in the CD spectrum of ToIC refolded in LDAO micelles compared to DDM micelles 138 (Figure 2A). The BeStSel (43) webserver was used to estimate the percentage composition of 139 secondary structural elements in ToIC refolded in LDAO micelles compared to DDM micelles. 140 141 The result indicates that ToIC refolded in LDAO is less structured (Figure 2B).

To further characterize refolded ToIC, we examined the unfolding of the protein by monitoring the change in ellipticity at 222 nm as a function of temperature. For ToIC refolded in DDM micelles, we observed a progressive shift in MRE at 222 nm as temperature increased beyond 75 °C, indicating protein unfolding due to high temperature. However, we observed no melting point in the CD signal at 222 nm with ToIC refolded in LDAO micelles within the

- 147 temperature range of 20 to 90 °C used in this study (Figure 2C and S1). These CD data
- 148 suggest that trimeric ToIC formed in LDAO micelles has a different conformation and is possibly
- 149 less folded than trimeric ToIC formed in DDM micelles.



151 conformation of ToIC refolded in LDAO micelles. Previous studies have reported that correctly

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152 folded TolC becomes resistant to proteinase K digestion, forming a 46 KDa digestion-resistant

- 153 product (44,45). Refolded ToIC in both DDM and LDAO micelles were subjected to proteinase K
- digestion. The digestion products were then boiled and visualized by SDS PAGE. TolC refolded
- in DDM was resistant to proteinase K and formed a 46 kDa digestion resistant product after a 30
- 156 minutes incubation. However, TolC refolded in LDAO micelles was completely digested by
- 157 proteinase K within 30 min of incubation (**Figure 2D**).
- 158 The above data indicate that although ToIC refolded in LDAO micelles exists as a trimer,

Table 1: Detergents used to refold ToIC in this study.

	Name	Chain length	Headgroup	Headgroup type
DDM	n-dodecyl-β-D-maltopyranoside	12	maltose	Nonionic
OM	n-octyl-β-D-maltopyranoside	8	maltose	Nonionic
C12E8	octaethylene glycol monododecyl ether	12	polyoxyethelene	Nonionic
C8POE	n-octylpolyoxyethylene	8	polyoxyethelene	Nonionic
	n-Dodecyldodecyl-N,N- dimethylaminedimethylamine-			
LDAO	N-Oxide oxide	12	amine oxide	Zwitterionic
SB3-12	sulfobetaine 3-12 (SB3-12)	12	sulfobetaine	Zwitterionic

it assumes a different conformation than in DDM micelles. We further sought to understand 159 160 what property of detergents results in the native-like or non-native-like ToIC folding. We 161 screened four additional detergents varying headgroups and alkyl chain lengths and observed 162 their effects on the folding, stability, and function of ToIC. These include three nonionic 163 detergents n-octyl- β -D-maltopyranoside (OM), n-octylpolyoxyethylene (C8POE), and 164 octaethylene glycol monododecyl ether (C12E8), and one zwitterionic detergent, sulfobetaine 3-165 12 (SB3-12) (Table 1). Like DDM, C12E8 has a hydrophobic tail with a 12-carbon alkyl group 166 while C8POE and OM have an 8-carbon tail length. The headgroup of DDM and OM is 167 composed of maltose while that of C12E8 and C8POE have polyoxyethylene (Figure 3A-D). 168 SB3-12 has a 12-carbon hydrophobic tail length like LDAO, and a positively and negatively charged headgroup contributed by nitrogen and oxygen respectively (Figure 3D). 169

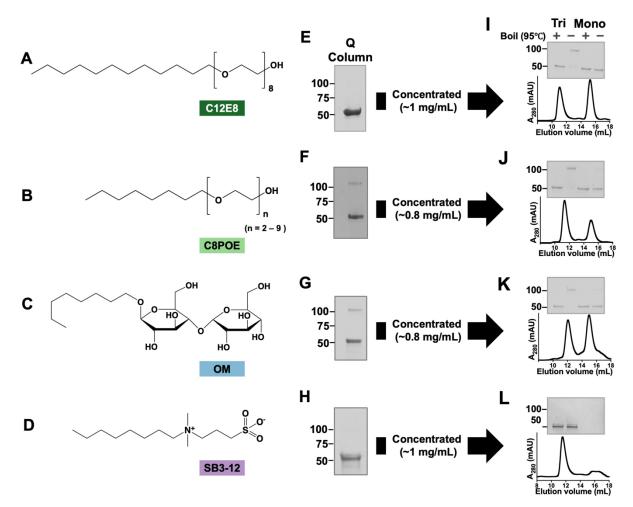


Figure 3: TolC trimerization before concentration depends on detergent headgroup and alkyl chain length. After concentration, TolC trimerization is only dependent on detergent headgroup. (A-D) Molecular structures of (A) n-octyl-β-D-maltopyranoside, (B) octaethylene glycol monododecyl ether, (C) sulfobetaine 3-12, (D) noctylpolyoxyethylene. (E-H) SDS PAGE of TolC refolded in (E) C12E8, (F) C8POE, (G) OM, and (H) SB3-12 micelles by rapid dilution followed by ion exchange chromatography. (I-L) SDS-PAGE and SEC chromatogram of captured monomeric (right) and trimeric (left) TolC refolded in (I) C12E8, (J) C8POE, (K) OM, and (L) SB3-12.

170 Using an identical protocol to previous ToIC folding, urea-solubilized inclusion bodies

- were rapidly diluted into detergent and then passed through a Q-column. TolC diluted into
- 172 C12E8 and SB3-12 and visualized on SDS-PAGE, resolved at a molecular weight
- 173 corresponding to the size of monomeric TolC (Figure 3E and 3H). However, SDS PAGE shows
- that ToIC trimerizes more readily in detergent micelles with shorter alkyl chain length C8POE,
- and OM than longer chain length, DDM and C12E8. (Figure 3F-G). However, when

concentrated, only the detergents with nonionic head groups developed the trimer band
visualized by SDS PAGE which is confirmed with a concomitant trimeric peak on SEC (Figure **3I, 3J and 3K)**. In contrast, in the zwitterionic SB3-12 detergent micelles, we saw a similar trend
observed with zwitterionic LDAO micelles where refolded TolC elutes at a volume
corresponding to the size of trimeric TolC on SEC but resolves at a size corresponding to
monomeric TolC on SDS PAGE (Figure 3L).

We used CD to investigate the folding of ToIC in each of the detergents tested. We observed
overlaping CD spectra of trimeric ToIC refolded in all the nonionic detergents tested (Figure 4A)
indicating a similar level of secondary structure formation. In the same way, ToIC refolded in
zwitterionic SB3-12 has a similar CD spectrum to ToIC refolded in zwitterionic LDAO (Figure
Consistent with ToIC refolded into zwitterionic LDAO, ToIC refolded into zwitterionic SB3shows no melting point in the temperature range used in this study (Figure 4B and S1).

In addition to the differences in the folding in nonionic and zwitterionic detergents, the thermal denaturation assay reveals another head group-based difference in the thermodynamic stability of ToIC within the nonionic detergents. Though ToIC folded in nonionic detergents demonstrates heat modifiability on SDS-PAGE and has a similar secondary structure as measured by CD, ToIC is substantially more stable in the maltose-headgroup lipids. The T_m of ToIC in the maltose headgroup detergents is 30 - 40 °C higher than the T_m of ToIC in polyoxyethylene headgroup detergents (**Figure 4B and S1**).

While trimers form even without concentration for either nonionic eight-carbon chain lipid, the effect of chain length on the stability of TolC in the tested nonionic detergents differs depending on the headgroup composition. For detergents with maltose head group, the 12carbon length chain increased the T_m about 7 °C more than the 8 carbon length. Conversely, while for detergents with polyoxyethylene headgroup, the 8 carbon length chain increased the T_m about 3 °C more than the 12 carbon.

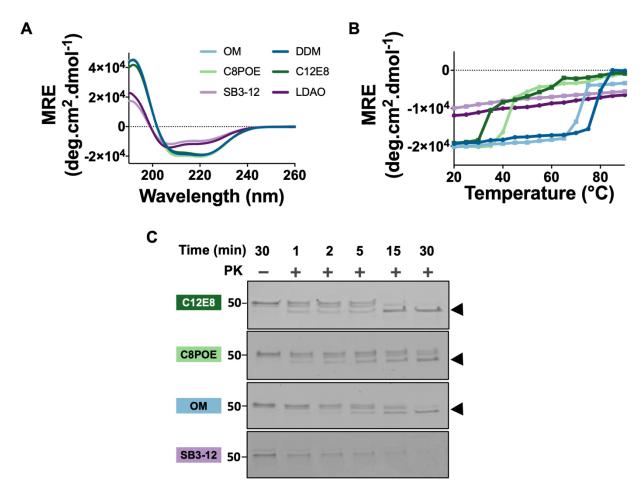


Figure 4: Secondary structure and PK resistance of TolC is dependent on detergent headgroup and not alkyl chain length. (A) Overlaid CD spectra of DDM (dark blue), OM (light blue), C8POE (light green), C12E8 (dark green), LDAO (dark purple), and SB312 (light purple). (B) MRE of TolC at 222 nm as a function of temperature. (C) SDS PAGE analysis of trimeric TolC incubated with proteinase K in detergents.

201 The observed variance in the stability of ToIC in the different detergents tested, led us to

investigate the conformation of ToIC and its insertion into the tested micelles by PK digestion

- assay. The PK digestion assay indicates that, like nonionic DDM, trimeric TolC refolded in all
- 204 nonionic detergents—C12E8, OM, and C8POE—is resistant to proteinase K digestion and
- forms a 46 kDa digestion resistant product. However, like zwitterionic LDAO, TolC refolded in
- 206 zwitterionic SB3-12 detergent micelles was completely digested within 30 min of incubation with
- 207 PK (Figure 4C). This indicates that in all nonionic detergents we tested, ToIC is in a native-like

208 conformation, but that in all zwitterionic detergents we tested, ToIC is in a non-native209 conformation.

210 Because the melting temperature data and proteinase K digestion data indicate a possible difference in conformation between ToIC folded in nonionic detergents and ToIC folded 211 212 in ionic detergents, we sought to determine if ToIC refolded in zwitterionic detergents (LDAO and SB3-12) can bind native ToIC ligand colicin E1 (46,47). Using a coelution assay, a mixture 213 of refolded ToIC and CoIE1 was incubated for an hour and then separated on an SEC column. 214 We observed a shift in the elution peak in the SEC chromatogram of the mixture of CoIE1 and 215 216 ToIC refolded in all of the four nonionic detergents tested when compared with that of ToIC alone. This elution peak shift indicates the presence of a ToIC-CoIE1 adduct formed from the 217 interaction of ToIC with CoIE1 (Figure 5A-D). However, the SEC chromatogram shows no shift 218

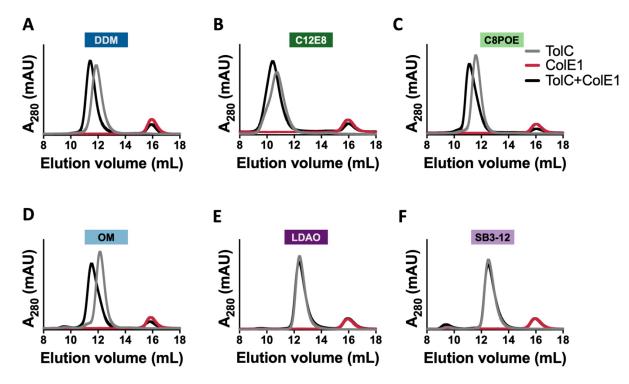


Figure 5: Effect of detergent headgroup on ToIC binding to a native ligand. SEC chromatograms of CoIE1 (red), ToIC+CoIE1 (black) with ToIC refolded in (A) DDM, (B) C12E8, (C) C8POE, (D) OM, (E) LDAO, and (F) SB3-12. 10 μ M of ToIC was incubated with 20 μ M of CoIE1.

in the elution volume when ColE1 is mixed with trimeric TolC refolded in LDAO or SB3-12
micelles. (Figure 5 F-G). Therefore, although TolC forms a monodispersed trimer when refolded
in zwitterionic detergent micelles, the trimer formed has a different conformation from native
TolC and does not bind to its endogenous binding partner.

223 **Discussion**

ToIC exists in its native and functional state in the outer membrane as a trimer with each polypeptide chain contributing to the formation of a cylindrical, β -barrel crowning an α -helical barrel (45,48,49). This conformation is essential for ToIC to perform its function in the outer membrane.

It has only recently been determined that ToIC can be folded in vitro (31). Here we find a 228 229 different ToIC folding behavior in zwitterionic detergent micelles compared to nonionic detergent 230 micelles. Although refolded ToIC forms a monodispersed trimer in zwitterionic detergent 231 micelles as observed in the SEC chromatogram (Figure 1G), we find that this is not the native 232 conformation. Structural (Figure 4) and functional (Figure 5) characterization show that trimeric ToIC formed in zwitterionic detergent micelles has a different conformation than native trimeric 233 ToIC. Although we may have expected the soluble, periplasmic alpha- helicalportion to form 234 235 regardless of the detergent used, we find instead that ToIC in zwitterionic micelles have half the 236 alpha-helical content but comparable β -sheet content.

Previous studies have indicated that *in vitro* OMP folding involves the formation of an intermediate step where OMPs associate with the membrane mimetics before they are inserted into the micelles or bilayers (50). The hydrophilic, periplasmic alpha helix domain of ToIC is composed of clusters of charged residues. We speculate that the non-native folded state we observed in zwitterionic detergent micelles may be a low-energy state formed by interactions between residues of the alpha helix domain of ToIC and the charged headgroup of the

zwitterionic detergent micelles. If that were the case, charge-charge interactions would lead to
the formation of non-native, stable, but non-functional trimeric ToIC on the surface of the
detergent micelle.

Our finding that more trimer is visible by SDS page when diluting into detergent micelles with shorter acyl chains before concentration (**Figure 3**) is in agreement with previous studies that showed single-chain OMPs fold more readily into lipids with shorter hydrocarbon chains (20,21,23).

250 Both the polyoxyethelyne detergent micelles and maltoside detergent micelles remain 251 intact upon heating until well above the ToIC unfolding transitions (51-53). Since the unfolding is 252 not due to de-micellation, the unfolding temperature is likely a property of the ToIC interaction 253 with the headgroup rather than a property of the micelle itself. Maltose headgroup detergents 254 (DDM and OM) stabilize the ToIC trimer at higher temperatures. ToIC denaturation in maltoside 255 detergents takes place over a narrow temperature range, starting at about 70 °C to 75 °C and complete denaturation is observed at about 75 °C to 80 °C (Figure 4B, Figure S1). The spectra 256 of the semi-denatured state appear more like an alpha helix before complete denaturation 257 (Figure S1). However, native-like trimeric ToIC is substantially less stable in detergent with 258 259 polyoxyethylene headgroups (C12E8 and C8POE). For polyoxyethylene headgroup detergents, 260 ToIC denaturation starts at about 40 °C and a gradual loss of secondary structure is observed 261 until complete denaturation is reached at about 65 °C. Moreover, as denaturation begins, the 262 spectra appear more like a beta-sheet before complete denaturation is achieved (Figure S1). 263 Thus, the alpha helical part of ToIC is possibly disrupted by polyoxyethylene headgroups.

264 Consistent with previous reports for other OMPs (20,21), we find that shorter alkyl chains 265 facilitate ToIC insertion even without concentration. However, we find different effects of chain 266 length on stability depending on the headgroup. Polyoxyethylene increases stability for shorter 267 alkyl chains and maltose headgroups decreasing stability for shorter alkyl chains. Consistent

with our temperature melt data, perhaps the long polar region of the oxyethylene headgroups
disrupts the helical region of TolC with the longer eight oxyethylene disrupting more than POE
which has a distribution of two to nine oxyethylene. This is consistent with small amounts of
beta structure remaining in the unfolded state of TolC folded in polyoxyethylene micelles
(Figure S1). When there are no oxyethylene present TolC is more stable in the longer (C12)
alkyl chain of DDM than the shorter (C8) alkyl chain of OM.

274 Finally, we anticipate that our data will help to better understand the biogenesis of outer 275 membrane efflux pumps. Overall, we find that the head group composition of detergents plays a 276 significant role in the folding of multimeric outer membrane efflux pumps in vitro, with zwitterionic lipids preventing ToIC folding. These findings may contribute answers to long-277 standing questions of OMP folding: 1) Why can we so easily refold OMPs in vitro but in vivo 278 279 folding requires the BAM complex and 2) Once proto-OMPs are in the periplasm, how does 280 nature prevent OMP folding back into the inner membrane phospholipids. At least for ToIC, the answer may be that the natural negative charge of the phosphate in all phospholipids as well as 281 282 the zwitterionic headgroup of the dominant membrane lipid phosphatidylethanolamine (54) prevents folding into both the outer leaflet of the inner membrane and the inner leaflet of the 283 284 outer membrane. In contrast, the common laboratory use of non-ionic detergents facilitates ToIC 285 folding *in vitro*. It will be important to continue to investigate the biological relevance of this effect and to determine how bacterial cells facilitate efflux pump folding in the presence of zwitterionic 286 287 lipids.

288 Materials and Methods

289 E. coli strains

E. coli BL21(DE3) was used for the expression of ToIC as inclusion bodies.

291 Expression and purification of Colicin E1

292 Colicin E1 was expressed and purified as previously described (47). Briefly, residues 1-190 of colicin E1 (colE1-T) were cloned into pET303 containing a C-terminal 6x histidine tag. Plasmids 293 were transformed into BL21(DE3) and plated on LB + agar + 100 µg/ml carbenicillin. A single 294 colony was inoculated into 20 ml of LB + 100 µg/ml carbenicillin and grown at 37 °C with 295 296 shaking at 250 rpm overnight. The next morning 1L TB broth + 10 mM MgCl₂ + 100 µg/ml carbenicillin was inoculated with 20 ml of the overnight starter culture and grown at 37 °C with 297 298 shaking at 250 rpm until OD₆₀₀ reached 1.0 and induced with 1 mM IPTG. The temperature was 299 reduced to 15 °C with shaking at 250 rpm and incubated for 24 hours. Cell pellets were harvested by centrifugation at 4,000 g for 30 minutes at 4 °C. Cell pellets were resuspended in 300 lysis buffer (TBS, 5 mM MgCl₂, 0.2 mg/ml lysozyme, 5 µg/ml DNase, 1mM PMSF, 20 mM 301 302 imidazole) at 3 ml/g of cell pellet. Cells were lysed by sonication in an iced water bath (3 min, 2 sec on, 8 sec off, 40 % amplitude, QSonica Q500 with 12.7 mm probe). The cytoplasmic 303 304 fraction was isolated by centrifugation at 50,400 g at 4 °C for 1 hour. The supernatant was filtered through 0.22 µm membrane filter and applied to a 5 ml HisTrap FF column and purified 305 306 using an ÄKTA FPLC system with 20 column volumes wash step with binding buffer (20 mM 307 Tris, 40 mM NaCl, 25 mM imidazole) and eluted using a linear gradient from 0 – 100% elution 308 buffer (20 mM Tris, 400 mM NaCl, 500 mM imidazole) in 10 column volumes. Colicin-containing 309 fractions were pooled and concentrated to 2 ml and applied onto a HiLoad 16/60 Superdex 200 310 pg column and eluted with 1.5 column volumes in 20 mM Tris, 40 NaCl.

311 Expression and purification of ToIC as inclusion bodies

ToIC was expressed and purified as previously described (31). ToIC gene was received from R. 312 Misra and cloned into pTrc99a. The signal sequence was deleted for inclusion bodies 313 314 expression using the forward primer 5'- CATGGTCTGTTTCCTGTGTGAAATTG -3' and reverse 315 primer 5'- GAGAACCTGATGCAAGTTTATCAGC -3'. A sequence confirmed plasmid of ToIC was transformed into E. coli BL21(DE3) cells and plated on an LB agar plate with 100 µg/ml of 316 317 carbenicillin. A single colony from the plate was used to inoculate a 20 ml of LB media containing 100 µg/ml carbenicillin which was then incubated at 37 °C, 250 rpm overnight (~ 16 318 h). This starter culture was used to inoculate 1L of LB media containing 100 µg/ml carbenicillin. 319 320 The culture was grown at 37 °C, 250 rpm until it had an optical density (OD) at 600 nm of ~0.6. 321 Then expression of ToIC was induced with 1 mM of Isopropyl ß-D-1-thiogalactopyranoside 322 (IPTG) while the culture remained incubated at 37 °C, 250 rpm for 4 h. Cells were harvested by 323 centrifugation at 4,000 g for 25 min at 4 °C. Cell pellets were resuspended in lysis buffer (TBS, 5 mM MgCl₂, 0.2 mg/ml lysozyme, 5 µg/ml DNase, 2 mM phenylmethylsulfonyl fluoride (PMSF), 1 324 325 % (v/v) Triton X-100) at 3 ml/g of cell pellet and lysed via sonication (3 min, 3 sec on, 7 sec off, 326 40% amplitude, QSonica Q500 with 12.7 mm probe) in an iced water bath. The lysate was 327 centrifuged at 4,000 g for 25 min at 4 °C and the pellet was resuspended in 40 ml wash buffer (TBS, 1% (v/v) Triton X-100). The lysate was sonicated on an ice bath for an additional three 328 329 minutes and the inclusion bodies were recovered by centrifugation at 4,000 g for 25 min at 4 °C. The inclusion bodies pellet was washed by resuspension in 40 ml wash buffer and centrifuged 330 331 at 4,000 g at 4 °C for 25 min. The inclusion bodies pellet was then resuspended in 20 mM Tris-HCl, pH 8, aliquoted, and stored at -20 °C until further use. 332

333 Refolding TolC

TolC was refolded as previously reported (31). Briefly, inclusion bodies were thawed and
 centrifuged at 4,000 g for 25 min at 4 °C. The inclusion body pellet was dissolved in urea buffer

336 (20 mM Tris-HCl, pH 8, 8 M urea) and incubated at 37 °C for about 15 min. Solubilized inclusion was then centrifuged at max speed on a table-top centrifuge for 5 min to get rid of insoluble 337 inclusion bodies. To initiate refolding, 4 mg/ml of solubilized ToIC inclusion bodies was rapidly 338 339 diluted by 25-fold into refolding buffer (20 mM Tris-HCI, pH 8.0) containing detergent of interest 340 (0.5 % (w/v) n-dodecyl β-D-maltopyranoside (DDM, Anatrace) or 2 % (w/v) n-Octyl-β-D-341 maltopyranoside (OM, Anatrace) or 1 % (w/v) n-dodecyl-N,N-dimethylamine-N-Oxide (LDAO, 342 Anatrace) or 1 %(w/v) sulfobetaine 3-12 (SB3-12, Sigma-Aldrich) or 1 % (v/v) noctylpolyoxyethylene (C8POE, Bachem) or 0.5 % (w/v) octaethylene glycol monododecyl Ether 343 (C12E8, Anatrace) and gently mixed on a rotisserie. The solution was passed through a 0.22 344 345 µm membrane filter and loaded onto a 5 ml HiTrap Q HP anion exchange column (GE 346 Healthcare, USA) equilibrated with wash buffer (20 mM Tris-HCl pH 8.0) containing detergent of 347 interest (0.05 % (w/v) DDM or 2 % OM or 0.2 % (w/v) LDAO or 2 % (w/v) SB3-12 or 0.6 % (v/v) 348 C8POE or 0.05 % (w/v) C12E8). The column was then washed with 5 column volumes of wash buffer. The protein was then eluted with 3 column volumes of elution buffer (20 mM potassium 349 350 phosphate buffer pH 8, 500 mM sodium fluoride) containing the same concentration of 351 detergent of interest in wash buffer. 2X Laemmli SDS sample buffer was added to each sample and loaded on a 4 – 20 % polyacrylamide gel for SDS-PAGE analysis. Gels were imaged with 352 an Epson Perfection V600 photo scanner. 353

354 Size exclusion chromatography

Elution fractions from TolC refolding above were pooled and concentrated using an Amicon centrifugal protein concentrator of 10K molecular-weight cutoff at 4,000 g and 4 °C to a concentration of about 1 mg/ml unless otherwise stated. The sample was then filtered with an 0.22 µm filter. 500 µl of sample was loaded and the loop was emptied with 2.5 ml elution buffer onto a Superdex 200 Increase 10/300 pg size exclusion column, pre-equilibrated with the

elution buffer. The sample was then eluted with the elution buffer containing the detergent ofinterest.

362 CD spectroscopy

CD spectra were recorded using a J-815 spectrometer (Jasco, Germany) from 260 to 190 nm. 363 Thermal melt spectra were obtained by CD between 20 °C and 90 °C at 0.5 nm intervals in a 0.1 364 cm pathlength guartz cuvette. Spectra were collected at 5 °C temperature interval and 5 °C/min 365 366 temperature ramp time with two minutes of equilibration time at each target temperature. Two 367 replicates were collected for each protein sample and baseline (elution buffer) at 100 nm/min 368 scanning speed, a digital integration time of two seconds, and a bandwidth of 1.5 nm. The spectra were smoothed with a Savitsky-Golay filter, and the baseline was subtracted from each 369 370 protein sample. Data were converted to mean residue ellipticity and secondary structure 371 analysis was done using BeStSel web server (43).

372 Proteinase K digestion

The assembly of refolded ToIC was assessed by proteinase K digestion. 15 µg of protein was treated with 15 µg of Proteinase K on ice for 1, 2, 5, 15, and 30 min. 2 mM PMSF was added to inactivate the protease and the reaction was boiled at 95 °C for 5 min. The susceptibility of protein to the proteinase K was determined by SDS PAGE.

377 Coelution

ToIC and Colicin E1 were both buffer exchanged into coelution buffer (20 mM Tris pH 8.0, 40
mM NaCl, containing detergent of interest) using PD-10 desalting columns. Binding was
determined by coelution on a Superdex 200 Increase 10/300 size exclusion column. 1:2 molar
ratio (5 µM ToIC trimer to 10 µM CoIE1) of ToIC and CoIE1 respectively were mixed and
incubated at room temperature for 1 hour.

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538 Supporting Information

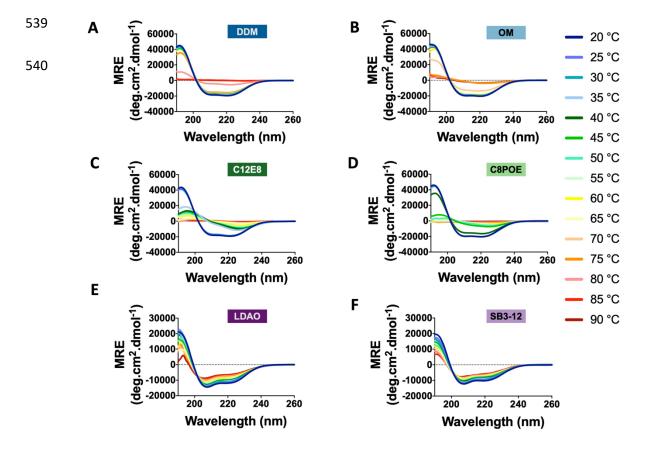


Figure S1: Thermal spectra of TolC refolded in detergents. Change in circular dichroism spectrum per 5 °C increase in temperature from 20 °C to 90 °C of TolC with alpha-helical intermediates in A and B and beta-sheet intermediates in C and D refolded in (A) DDM, (B) OM, (C) C12E8, (D) C8POE, (E) LDAO, and (F) SB3-12.