

1 Main manuscript for

4 *Mycobacterium tuberculosis* DosS binds H₂S through its Fe³⁺ heme iron to regulate the Dos 5 dormancy regulon

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50 **Abstract**

51 *Mycobacterium tuberculosis* (*Mtb*) senses and responds to host-derived gasotransmitters NO and
52 CO via heme-containing sensor kinases DosS and DosT and the response regulator DosR.
53 Hydrogen sulfide (H₂S) is an important signaling molecule in mammals, but its role in *Mtb*
54 physiology is unclear. We have previously shown that exogenous H₂S can modulate expression
55 of genes in the Dos dormancy regulon via an unknown mechanism(s). Here, we tested the
56 hypothesis that *Mtb* senses and responds to H₂S via the DosS/T/R system. Using UV-Vis and EPR
57 spectroscopy, we show that H₂S binds directly to the ferric (Fe³⁺) heme of DosS (K_D = 5.64 μM) but
58 not the ferrous (Fe²⁺) form. No interaction with DosT was detected. Thus, the mechanism by which
59 DosS senses H₂S is different from that for sensing NO and CO, which bind only the ferrous forms
60 of DosS and DosT. Steered Molecular Dynamics simulations show that H₂S, and not the charged
61 HS⁻ species, can enter the DosS heme pocket. We also show that H₂S increases DosS autokinase
62 activity and subsequent phosphorylation of DosR, and H₂S-mediated increases in Dos regulon
63 gene expression is lost in *Mtb* lacking DosS. Finally, we demonstrate that physiological levels of
64 H₂S in macrophages can induce Dos regulon genes via DosS. Overall, these data reveal a novel
65 mechanism whereby *Mtb* senses and responds to a third host gasotransmitter, H₂S, via DosS-Fe³⁺.
66 These findings highlight the remarkable plasticity of DosS and establish a new paradigm for how
67 bacteria can sense multiple gasotransmitters through a single heme sensor kinase.

68

69 **Significance Statement**

70 Hydrogen sulfide (H₂S) is an important signaling molecule in eukaryotes and bacteria, and along
71 with CO and NO, is an important part of host defense against *Mycobacterium tuberculosis* (*Mtb*).
72 However, the mechanism(s) by which *Mtb* senses and responds to H₂S is unknown. Here, we
73 report that the *Mtb* heme sensor kinase DosS, a known sensor of CO and NO, is also a sensor of
74 H₂S. We found that H₂S binds DosS in its ferric (Fe³⁺) state, which is considered as its inactive
75 state, to induce the Dos dormancy regulon during infection. These data highlight the unusual
76 capacity of *Mtb* to sense multiple gasotransmitters through a single sensing protein.

77 **Main Text**

78

79 **Introduction**

80 Tuberculosis (TB) is a global epidemic responsible for ~1.4 million deaths annually (1).
81 *Mycobacterium tuberculosis* (*Mtb*), the causal agent of TB, can persist in a state of clinical latency
82 for decades. *Mtb* survives and establishes an infection due, in part, to its ability to sense and
83 respond to host defenses in the lung, including host-generated gasotransmitters. Carbon monoxide
84 (CO) and nitric oxide (NO) are critical components of the host defense to clear the pathogen and
85 are important to the outcome of *Mtb* infection (2, 3). The most recent addition to the list of
86 gasotransmitters is hydrogen sulfide (H₂S). Notably, enzymes required for the generation of NO,
87 (nitric oxide synthase [iNOS]) (3), CO (heme oxygenase-1 [HO-1]) (2, 4, 5), and H₂S (cystathionine
88 β-synthase [CBS] (6), cystathionine γ-lyase [CSE] (6), and 3-mercaptopyruvate sulfur transferase
89 [3-MPST]) (6), are upregulated in the lungs of *Mtb*-infected mice and human TB patients. The
90 increased levels of these enzymes suggest an abundance of NO, CO, and H₂S at the primary site
91 of infection.

92 The role of NO and CO in TB pathogenesis is well studied compared to that of H₂S. We
93 have recently shown that *Mtb*-infected mice deficient in the H₂S-producing enzyme CBS (6, 7) or
94 CSE (6, 7) survive significantly longer with reduced organ burden, suggesting that host-generated
95 H₂S is beneficial for *Mtb in vivo*. Bacterial two-component regulatory systems sense changes in
96 the host environment and mediate adaptive genetic responses. The *Mtb* genome encodes 11
97 paired two-component systems (8), including the DosS/T-DosR system, comprised of heme-
98 containing sensor kinases DosS and DosT and their cognate transcriptional response regulator
99 DosR (9). DosS, currently regarded as a redox sensor, is considered inactive in the oxidized/met
100 (Fe³⁺) state and is activated by direct binding of NO or CO to the heme iron in the reduced (Fe²⁺)
101 state (10). DosT is an oxygen sensor and is inactive in its oxy-bound form which is activated upon
102 loss of O₂ or direct binding of NO or CO to the heme iron in the reduced (Fe²⁺) state (10, 11) (Fig.
103 1A).

104 We recently reported that exposure of *Mtb* to the H₂S donor GYY4137 induces the
105 expression of genes that regulate cysteine metabolism and genes in the copper and Dos dormancy
106 regulons (7). While H₂S can chemically modify biomolecules directly (12-14), we considered the
107 possibility that alterations in gene expression in response to H₂S are mediated by regulatory
108 proteins in *Mtb*. Since H₂S is known to bind the iron in heme-containing proteins (15-19), and
109 because DosS was among the Dos regulon genes induced upon exposure to H₂S (7), we
110 hypothesize that DosS and/or DosT sense and respond to H₂S to induce the Dos dormancy
111 regulon. To examine the interaction of H₂S with DosS and DosT, we used UV-visible and EPR
112 spectroscopy. We also examined how H₂S modulates DosS autokinase and phosphate transfer,
113 and *Mtb* gene expression *in vitro*. Lastly, we examined how *Mtb* senses H₂S during macrophage

114 infection. We expect that the findings in this work will lead to an improved understanding of *Mtb*
115 persistence.

116

117 **Results**

118

119 **DosS in the Fe³⁺ form binds H₂S**

120 DosS and DosT contain heme and exhibit specific absorption characteristics in the ultraviolet-
121 visible (UV-Vis) range, which are altered upon interaction between the heme iron and ligands like
122 NO and CO (10, 20). To determine whether *Mtb* DosS or DosT sense H₂S via its heme iron, we
123 monitored spectral changes of the ferric (Fe³⁺) and ferrous (Fe²⁺) forms of recombinant DosS and
124 DosT in the presence of sulfide (here, we define sulfide as H₂S and HS⁻) following addition of
125 sodium sulfide (Na₂S). Addition of sulfide red shifts the Soret peak of the Fe³⁺ form of DosS from
126 408 nm to 415 nm, indicative of a high-spin to low-spin transition (21), with increased peak
127 intensities of the α (570 nm) and β (535 nm) bands (Fig. 1B), similar to spectral changes observed
128 upon sulfide binding in other heme-containing proteins (Table 1). Reduction of DosS to the Fe²⁺
129 form using sodium dithionite (DTH) shifts the Soret peak to ~425 nm with emergence of a new peak
130 at 560 nm, as observed previously (10, 22). However, the absorption pattern of DTH-reduced DosS
131 remains unchanged in the presence of sulfide. Similarly, addition of sulfide does not alter the
132 absorption spectrum of DosT, where the heme iron remains in the oxy-bound state (Fe²⁺-O₂) (Fig.
133 1C) (10, 22). These data suggest that sulfide directly interacts with the Fe³⁺ form of DosS. This is
134 mechanistically distinct from the binding of NO and CO, which bind the heme iron of DosS and
135 DosT in the Fe²⁺ state.

136 H₂S is in protonation equilibrium with HS⁻ in solution with a pK_a value of 7.01 (23). To
137 determine whether H₂S or anionic hydrosulfide (HS⁻) binds to the Fe³⁺ form of DosS, we monitored
138 the UV-Vis spectra of DosS at pH values above and below the pK_a of H₂S in the presence of 25
139 μM Na₂S. Notably, as the pH decreased we observed lower peak intensities in the 405 nm range,
140 indicating reduced levels of unbound DosS, and increases in the ~420 nm range corresponding to
141 sulfide-bound DosS. These data suggest that H₂S, and not HS⁻, is the sulfide species that initially
142 binds to the heme iron of DosS (Fig. 1D).

143 Iron is paramagnetic in its Fe³⁺ state. Ligand binding to the heme iron results in
144 perturbations in the *d*-orbitals that can be monitored by electron paramagnetic resonance (EPR)
145 spectroscopy (10, 24). To confirm that H₂S binds directly to the Fe³⁺ form of DosS, we compared
146 the EPR spectra of DosS before and after exposure to sulfide. DosS alone gives a strong axial
147 feature centered at g=5.98, which is indicative of Fe³⁺ in the high-spin (S=5/2) state (Fig. 1E) (10,
148 25) (Table 1). The addition of increasing concentrations of sulfide results in the conversion of the
149 g=5.98 high-spin signal into a rhombic low-spin signal with g values of 2.67, 2.26, and 1.76 (Fig. 1
150 F-H), which are similar to low-spin sulfide-bound species reported for other heme-containing

151 proteins (Table 1). Taken together, these results indicate that H₂S is a ligand of the Fe³⁺ form of
152 DosS and binding of H₂S converts the Fe³⁺ heme iron from the high-spin state to low-spin state.

153

154 **Molecular dynamics simulations show that H₂S, but not HS⁻, enters the DosS heme pocket**

155 The DosS heme group is buried within a hydrophobic pocket that is enclosed within the N-terminal
156 GAF-A domain (22, 26). Access to the hydrophobic heme pocket is limited, and ligand entry is
157 influenced by the adjacent amino acid side chains (22). Steered Molecular Dynamics (sMD) has
158 been used to estimate association free energy required for H₂S and HS⁻ to access the heme iron
159 in *L. Pectinata* met-hemoglobin (27) and met-myoglobin (27, 28). Similarly, our sMD simulations
160 estimate the free energy barriers for access to the DosS heme iron to be approximately 5.6 kcal/mol
161 for H₂S and 16.7 kcal/mol for HS⁻ (Fig. 2A). The much higher free energy barrier for HS⁻ indicates
162 that the uncharged H₂S species is strongly favored to enter the heme pocket.

163 Our sMD simulations predict that H₂S accesses the heme iron by passing between Phe⁹⁸
164 and Leu¹¹⁴ of the heme binding pocket (Fig. S1). These residues form a “gate” which when open
165 allows H₂S and a few water molecules into and out of the heme pocket. This is possible due to the
166 neutral charge of H₂S (Fig. S1 A-C, Movie S1). In contrast, our sMD simulations indicate that HS⁻
167 carries a strongly-bound solvation sphere that pushes the Phe⁹⁸ and Leu¹¹⁴ side chains away,
168 resulting in a considerable number of water molecules entering the heme pocket and increasing
169 solvation of the heme active site (Fig. S1 D-F, Movie S2). The process of HS⁻ entry is energetically
170 unfavorable, as indicated by a much higher predicted free energy barrier compared to H₂S.
171 Notably, this H₂S entry “gate” is located away from the “water channel” identified by Cho, et. al., in
172 the DosS GAF-A domain structure (22). Overall, our sMD modeling indicating more favorable heme
173 access for H₂S supports our UV-Vis data demonstrating increased binding of sulfide at lower pH
174 (Fig. 1D).

175

176 **Quantum mechanics simulations show that H₂S deprotonates following heme iron binding**

177 Modeling studies of heme-containing proteins predict that H₂S can deprotonate following binding
178 to heme iron (27, 28). To characterize the heme-bound state of H₂S in DosS, we employed
179 combined quantum mechanics/molecular mechanics simulations with density functional theory
180 calculations (QM[DFT]/MM). Table 2 shows the structural and electronic parameters of both H₂S
181 and HS⁻ bound states of DosS. As expected, the ligand-bound states display a low-spin ground
182 state. The structural analysis shows that HS⁻ forms a tighter bond (a significantly smaller Fe-S
183 distance, [d Fe-S = 2.17 Å]) due to a significant charge transfer (sigma donation). Further, the
184 proximal Fe-His bond shows a slightly positive trans effect, displaying a slightly smaller distance
185 compared to penta-coordinated heme (ca 2.12 Å) (29, 30). Interestingly, in the H₂S-bound state,
186 the two protons become asymmetric and a weakening of the S-H bond is observed (d Fe-S = 2.35
187 Å). On this basis, we evaluated the possibility of deprotonation of Fe-bound H₂S using hybrid

188 QM/MM simulations. Strikingly, a simulation duration of 1 picosecond (ps) was sufficient to observe
189 deprotonation of Fe-bound H₂S (Fig. 2B-D and Movie S3). The proton acceptor is a water molecule
190 located opposite from the distal Tyr (i.e., close to the solvent-exposed heme edge) which
191 subsequently transfers a proton to the heme propionate (Fig. 2D). The heme propionate is
192 accessible to the solvent and can again be deprotonated or remain in a protonated state. Overall,
193 these data strongly suggest that HS⁻ is the tighter binding and predominant bound sulfur species.

194

195 **H₂S does not reduce the heme iron in DosS and binds with low micromolar affinity**

196 H₂S is known to reduce heme iron to the Fe²⁺ state in proteins like myoglobin and hemoglobin (16).
197 Since the Fe²⁺ form of DosS is understood to be the active form of the kinase (10, 20), it is important
198 to determine whether H₂S binding reduces the DosS heme iron. Therefore, we monitored the UV-
199 Vis spectrum of the Fe³⁺ form of DosS in the presence of 200 μM Na₂S over time. As shown in Fig.
200 3A, addition of Na₂S resulted in the expected shift in the Soret peak from 408 nm to 415 with
201 increased peak intensity of the α (570 nm) and β (535 nm) bands. Notably, we observed no
202 additional shift in the Soret peak to 425 nm nor the appearance of the peak at 560 nm, both of
203 which are indicative of the reduced form of DosS (Fig. 1A). Over the course of the experiment we
204 noted a decrease in overall absorbance, which is most likely due to the loss of the heme prosthetic
205 group from DosS (31, 32). These results indicate that H₂S binding does not readily alter the
206 oxidation state of the DosS heme iron.

207 To determine the affinity of DosS for H₂S, we next monitored changes in the absorbance
208 spectrum of DosS in the Fe³⁺ state over a wide range of Na₂S concentrations (Fig. 3B). Using the
209 maximum change in absorbance at ~408 nm, we generated a substrate saturation curve from which
210 a K_D^{app} value of 5.64 ± 0.27 μM was calculated for H₂S binding to DosS (3B and Inset). Notably,
211 this K_D^{app} value is similar to the K_D^{app} value of 7.0 ± 0.4 μM reported for the H₂S-met hemoglobin
212 complex (15, 18). In summary, these results show that measureable, direct binding of H₂S to DosS
213 (Fe³⁺) does not reduce DosS to the active Fe²⁺ form.

214

215 **H₂S increases DosS autokinase activity and DosR phosphorylation**

216 To address the question of whether H₂S binding can alter DosS kinase activity, we performed
217 kinase assays using γ-³²P-labeled ATP and recombinant DosS. We determined relative autokinase
218 activities of DosS (Fe²⁺), DosS (Fe³⁺), and DosS (Fe³⁺-HS⁻) at 5-60 minutes following addition of
219 γ-³²P labeled ATP. Compared to the Fe³⁺ form of DosS, which is considered the least active form
220 (10), the Fe³⁺-HS⁻ form of DosS exhibited increased autokinase activity. In these assays, the Fe²⁺
221 form of DosS showed the highest autokinase activity (Fig 4A). Densitometric analysis of autokinase
222 radiograms shows that the activity of the Fe³⁺-HS⁻ form of DosS is increased by an average of
223 ~30% at all time points compared to unbound DosS in the (Fe³⁺) form (Fig. 4B).

224 Next, we sought to determine whether the H₂S-mediated increase in DosS autokinase
225 activity results in augmented phosphorylation of DosR, the cognate response regulator of DosS.
226 First, to standardize our transphosphorylation assays, unphosphorylated recombinant DosR was
227 added to reaction buffer containing γ -³²P-labeled DosS. Surprisingly, the phosphorylation of DosR
228 by labeled DosS was extremely rapid, and was completed within seconds (Fig. S2). Therefore, to
229 observe differences in the rates of phosphate transfer of DosS (Fe²⁺), DosS (Fe³⁺), and DosS (Fe³⁺-
230 HS⁻), unphosphorylated DosS and DosR were added to the reaction buffer prior to the addition of
231 γ -³²P-labeled ATP. Under these conditions, we observed increased phosphorylation of DosR in
232 the presence of DosS (Fe³⁺-HS⁻) compared to DosR phosphorylation in the presence of DosS (Fe³⁺)
233 at all time points measured. Maximum phosphorylation of DosR was observed in the presence of
234 DosS (Fe²⁺) (Fig. 4C). Densitometric analysis of transphosphorylation radiograms shows an ~2-
235 fold increase in labeled DosR in the presence of DosS (Fe³⁺-HS⁻) compared to the Fe³⁺ form of
236 DosS (Fig. 4D). Overall, these data indicate that the binding of sulfide to DosS increases its
237 autokinase activity. Given the extremely rapid transfer of phosphate from DosS to DosR, we
238 conclude that increased DosR phosphorylation is attributable primarily to changes in autokinase
239 activity.

240 Our findings that sulfide binding increases autokinase activity and that HS⁻ is likely the
241 predominant bound ligand have implications for the structural basis of kinase domain activation.
242 To elucidate a mechanism by which sulfide binding increases DosS autokinase activity, we
243 employed MD modeling to detect structural differences between DosS in the Fe³⁺ and Fe³⁺-HS⁻
244 forms. Our analysis indicates that the hydrogen bonding network distal to the heme is significantly
245 different between the ferric high spin (off-state) and the Fe³⁺-HS⁻ low spin state (Fig. 5), but not the
246 Fe³⁺-H₂S state. In the Fe³⁺ state of DosS, there is a loosely coordinated water molecule which
247 cannot act as a hydrogen bond acceptor for Tyr¹⁷¹. Therefore the hydroxyl group of Tyr¹⁷¹ rotates
248 upward and instead forms a tight hydrogen bond with Glu⁸⁷ (Fig. 5A). In contrast, when HS⁻ is
249 bound to DosS in the Fe³⁺ state, Tyr¹⁷¹ hydroxyl group rotates downward and establishes a tight
250 hydrogen bond with HS⁻ with the negative sulfur atom as the hydrogen bond acceptor. This results
251 in the release of Glu⁸⁷, which then moves closer to, and establishes a tight hydrogen bond with,
252 His⁸⁹ and Thr¹⁶⁹ resulting in changes in their relative position, particularly for the loop in the peptide
253 backbone in which His⁸⁹ is located. Disruption of the hydrogen bond between Tyr¹⁷¹ and Glu⁸⁷ is
254 further promoted by the presence of water molecules between them (Fig 5B). These observations
255 suggest that the DosS off-state is characterized by strong Tyr¹⁷¹-Glu⁸⁷ interaction, while the on-
256 state is characterized by a strong Tyr¹⁷¹-HS⁻ interaction that releases Glu⁸⁷, which is consistent with
257 the observation that CO/NO-bound DosS is active and shows the disruption of the hydrogen
258 bonding network between Tyr¹⁷¹-Glu⁸⁷-His⁸⁹ (26). Overall it appears that signal transmission to the
259 histidine kinase domain is initiated by disruption of Glu⁸⁷-Tyr¹⁷¹ H-bonding and further amplified by
260 positional changes in the His⁸⁹-containing loop.

261

262 **DosS senses H₂S to regulate the Dos dormancy regulon**

263 To determine whether the H₂S-mediated activation of DosS is sufficient to drive increased
264 expression of Dos regulon genes, we exposed WT *Mtb* and *Mtb* Δ *dosS*, a *dosS* deletion mutant
265 (33), to Na₂S. This was followed by quantitation of mRNA transcripts of representative Dos regulon
266 genes. As shown in Fig. 6A, *fdxA*, *hspX*, *rv2030c* and *rv2626* transcript levels were markedly
267 increased in WT *Mtb*, but not Δ *dosS*, following exposure to Na₂S. These results indicate that H₂S
268 mediates increased expression of Dos regulon genes via DosS activation, consistent with our
269 previous observations (7).

270 We next sought to determine whether host-generated H₂S can be sensed by DosS to
271 mediate induction of Dos regulon genes. CBS and CSE can produce H₂S using L-cysteine as a
272 substrate (34, 35). To avoid potentially confounding effects of CBS/CSE enzyme inhibitors on *Mtb*
273 as seen previously (7), we chose to use Cys to modulate intracellular levels of H₂S in RAW 264.7
274 macrophages. First, we examined H₂S levels in RAW 264.7 macrophages grown in media
275 supplemented with L-cysteine using the fluorescent WSP-5 H₂S-sensing probe (36). We observed
276 an increased intracellular fluorescence signal in RAW 264.7 macrophages grown in media
277 containing 0.1-2.0 mM L-cysteine compared to cysteine-free media, with maximum signal observed
278 at 0.2 mM L-cysteine (Fig. S3). This finding demonstrates that addition of exogenous L-cysteine
279 increases H₂S production. Importantly, levels of Cys in serum and cells have been reported to
280 range between 30-260 μ M (37-40). On this basis, RAW 264.7 macrophages were grown in
281 cysteine-free media or media containing 0.2 mM L-cysteine and infected with WT or Δ *dosS* *Mtb*.
282 At 24 hours post-infection, intracellular bacteria were recovered and RNA was extracted. We
283 observed marked increases in transcript levels of *fdxA*, *hspX*, *rv2030c*, and *rv2626* in WT *Mtb*, but
284 not Δ *dosS* *Mtb*, isolated from macrophages grown in media containing 0.2 mM L-cysteine (Fig. 6B-
285 I). These data indicate that *Mtb* senses host-derived H₂S via DosS to induce the expression of Dos
286 regulon genes.

287

288 **Discussion**

289 The *Mtb* DosS/T/R signal transduction system is known to sense three host-derived dormancy
290 signals, NO (10, 11, 33), O₂ (33, 41), and CO (4, 5) to induce the 48-gene Dos dormancy regulon.
291 Here, we report that *Mtb* is capable of sensing a fourth gasotransmitter, H₂S, to induce the Dos
292 dormancy regulon. Importantly, we show that DosS is capable of sensing H₂S at levels produced
293 by macrophages, resulting in the upregulation of key dormancy regulon genes. We also show that
294 *Mtb* DosS, but not DosT, can sense H₂S and does so via its heme iron in the ferric state to modulate
295 its autokinase activity resulting in increased phosphorylation of DosR. The ability of sulfide to bind
296 DosS ferric heme iron is clearly distinct from that of NO and CO, which bind DosS only when its
297 heme iron is reduced (Fig. 7). Thus, DosS exhibits remarkable plasticity in sensing multiple

298 gasotransmitters regardless of the oxidation state of the heme iron to ensure induction of the Dos
299 dormancy regulon during infection. These findings establish a new paradigm for how bacteria
300 sense multiple signaling molecules with distinct physicochemical properties through a single heme
301 sensor kinase.

302 The importance of H₂S as a signaling molecule in the regulation of numerous important
303 physiological functions in mammals, including immunity, is well established (42, 43). Hence, it is
304 reasonable that bacterial pathogens have evolved the capacity to detect H₂S to reprogram their
305 transcriptomes to subvert the immune response. However, while numerous signaling studies have
306 been performed in bacteria including, but not limited to, *E. coli*, a gut organism continuously
307 exposed to H₂S, little is known regarding whether H₂S has a direct signal transduction function in
308 bacteria. More specifically, evidence supporting the ability of bacteria to sense and respond to
309 host-derived H₂S via heme sensor kinases in two-component systems is lacking (44, 45).

310 Of note, H₂S is produced at the site of *Mtb* infection, as demonstrated by the presence of
311 H₂S-producing enzymes CSE and 3-MPST around cavities and necrotic lesions in the lungs of TB
312 patients (6). Since H₂S is highly diffusible, intracellular and extracellular *Mtb* will be exposed to
313 host-generated H₂S. *Mtb* is also exposed to NO (3), CO (2) and hypoxia (46) in the lungs of TB
314 patients depending on disease stage. NO, CO and H₂S are produced at different levels and times
315 throughout the course of infection, which likely vary depending on the particular microanatomical
316 location and pathology induced by *Mtb*. This, coupled with differing on- and off-rates of these
317 molecules for DosS and DosT heme iron, strongly suggest that the *Mtb* Dos dormancy regulon is
318 induced throughout the course of infection. Studies in the macaque model of inhalation TB have
319 shown that *Mtb* Δ *dosS*, Δ *dosR* and Δ *dosT* mutants are attenuated (47). However, a *Mtb* Δ *dosS*
320 mutant, but not Δ *dosR* or Δ *dosT* mutants, was shown to be attenuated in C3HeB/FeJ mice and
321 macrophages (48). Notably, these authors demonstrated that DosS phosphorylates proteins other
322 than DosR, suggesting that DosS may modulate expression of genes not in the Dos dormancy
323 regulon (49). These data suggest that the mechanisms by which DosS senses and responds to
324 environmental signals is more complex than previously thought.

325 Several lines of evidence provide insight into the mechanism whereby *Mtb* senses H₂S.
326 Firstly, we found that DosS, but not DosT, is a H₂S sensor as the latter is in the ferrous oxy-bound
327 state (Fe²⁺-O₂) under normoxic conditions. Secondly, sulfide binds DosS under normoxic
328 conditions where its heme iron is in the ferric form, as demonstrated by UV-Vis and EPR
329 spectroscopy. Our spectroscopic data also show that binding of sulfide does not alter the oxidation
330 state of DosS ferric heme iron, and that direct H₂S binding to the ferric iron changes its spin state
331 from high to low. Our findings are consistent with several reports showing that sulfide reacts with
332 heme iron in the ferric (Fe³⁺) state (15-19). This may allow induction of the Dos regulon at earlier
333 stages of infection prior to the onset of hypoxia when ferrous DosS can respond to NO and CO
334 only. Thirdly, our MD modeling using the DosS GAF-A domain X-ray structure (PDB code 2W3E)

335 (22) provides key insight into the molecular interactions of sulfide with DosS. Consistent with our
336 sMD modeling of other hemoproteins (27, 28) H₂S has more favorable access to the DosS heme
337 than does HS⁻, which can explain the increase in H₂S binding to DosS with decreasing pH. This
338 selectivity is due to the strong hydration of HS⁻ compared to H₂S which significantly hinders HS⁻
339 entry into the hydrophobic pocket, particularly *via* a “gate” comprised of Phe⁹⁸ and Leu¹¹⁴ side
340 chains. Further, QM[DFT]/MM calculations demonstrate a stronger Fe-S bond for HS⁻ than for H₂S,
341 suggesting deprotonation is induced upon binding of H₂S. Indeed, our hybrid QM/MM simulations
342 indicate proton transfer from H₂S via a water molecule to a heme propionate resulting in bound HS⁻
343 as the final state.

344 To the best of our knowledge, this is the first report of a heme sensor kinase that binds H₂S
345 to increase its catalytic activity under physiologically relevant conditions. An important remaining
346 question is how the binding of H₂S to DosS increases autokinase activity. Given that a complete
347 crystal structure for DosS has not been reported, a detailed study of the structural changes in the
348 kinase domain upon H₂S binding remains challenging. Nonetheless, structural studies probing the
349 mechanism by which DosS activity is modulated by the redox state of, and ligand binding to, the
350 heme iron have shown that an intact hydrogen bonding network comprised of Tyr¹⁷¹-Glu⁸⁷-His⁸⁹ is
351 important for inhibition of DosS kinase activity, as seen in the Fe³⁺ state (22). Disruption of this
352 network upon reduction of the heme iron and subsequent binding of NO or CO result in DosS
353 conformations that favor kinase activity (22, 26, 50, 51). Our MD- and QM/MM-based modeling of
354 the DosS GAF-A domain predicts that HS⁻ ligation to DosS disrupts the distal Tyr¹⁷¹-Glu⁸⁷-His⁸⁹
355 hydrogen bonding network. This is consistent with previously published proposals regarding signal
356 transmission between the heme and autokinase domains (26) and with our finding of H₂S-
357 stimulated autokinase and DosR phosphorylation.

358 Our finding that H₂S binds DosS in the Fe³⁺ state only, and not DosT which is in the Fe²⁺-
359 O₂ state, demonstrates that the mechanism of H₂S sensing is distinct from NO and CO, since these
360 molecules bind DosS in the Fe²⁺ state only. We (10) and others (22, 52) have shown that DosS
361 exists in the Fe³⁺ state and requires a reductant to generate ferrous DosS that binds NO or CO.
362 Ferredoxin (FdxA), reduced flavin nucleotides (22), and chorismate synthase (CS) have been
363 posited as reductants of ferric DosS (22, 53, 54). FdxA and CS may be candidate reductants as
364 *fdxA* is part of Dos regulon which is upregulated under hypoxic conditions and CS accelerates
365 NADH-dependent FMN reduction. These potential reducing systems for DosS-Fe³⁺ may represent
366 an additional aspect of Dos regulon induction, and suggests that in the absence of a reducing
367 system, the Fe³⁺ form of DosS can still induce the Dos regulon via binding of H₂S. Hence, in the
368 presence of NO, CO or H₂S, the Dos regulon will be induced via DosS regardless of whether the
369 heme iron is in the Fe²⁺ or Fe³⁺ (unligated or ligated) state. Thus, these findings suggest that DosS
370 signaling arising due to its redox sensor function and ligand binding are equally important. This
371 may have important consequences *in vivo* and again suggests that the Dos dormancy regulon is

372 induced through most of the course of infection. It has been shown that NO and hypoxia (11, 55),
373 and likely CO, inhibit respiration, and all three factors induce the Dos regulon (5, 11, 55). However,
374 we have shown that H₂S increases *Mtb* respiration at low concentrations (7) and induces the Dos
375 regulon under normoxic conditions (Fig. 6 and (7)). These findings suggest that inhibition of
376 respiration (or hypoxia) is not the sole factor that induces the Dos dormancy regulon, and are
377 consistent with the action of the TB drug bedaquiline, which also stimulates respiration (56) and
378 induces the Dos dormancy regulon (57).

379 We (Fig. 1B, (10)) and others (22, 58) have shown that purified DosS exists in the Fe³⁺
380 state; however, others have reported DosS to be in the ferrous-oxy bound (Fe²⁺-O₂) form (20, 54).
381 The reasons for this discrepancy remain unclear but are likely due to differences in protein
382 purification methodology and experimental approaches. Nonetheless, our EPR and UV-Vis
383 spectroscopy data spectroscopy provide compelling evidence that DosS heme is in the ferric state
384 (Fig 1B and 1D). Also, our data showing that DosS can only bind H₂S in the met (Fe³⁺) state is
385 consistent with numerous studies demonstrating that ferric, but not ferrous heme iron in proteins
386 binds H₂S (15, 16, 18, 45).

387 In light of previous studies showing that *Mtb* infection of macrophages induces upregulation
388 of host H₂S-producing enzymes CBS (7) and CSE (6), which leads to excessive levels of H₂S to
389 exacerbate disease, it was important to demonstrate that *Mtb* is capable of sensing physiological
390 levels of H₂S. Indeed, we demonstrated that WT *Mtb*, but not *Mtb* Δ *dosS*, senses endogenous
391 levels of H₂S during infection to induce key genes in the Dos regulon *fdxA*, *hspX*, *rv2030c*, and
392 *rv2626*. This was further confirmed by enhancing H₂S production in macrophages via substrate
393 supplementation (6), which lead to increased expression of these Dos dormancy genes. This
394 provides strong evidence that DosS senses H₂S during infection.

395 To the best of our knowledge, this is the first report of a heme sensor kinase that binds H₂S
396 to increase its catalytic activity under physiologically relevant conditions. Further, our results show
397 that H₂S can function as a signaling molecule in *Mtb*. An important remaining question is how the
398 binding of H₂S to DosS increases autokinase activity. Given that a complete crystal structure for
399 DosS has not been reported, a detailed study of the structural changes in the kinase domain upon
400 H₂S binding remains challenging. However, our MD-based modeling predicts that H₂S binding to
401 DosS disrupts the hydrogen bonding network in the distal domain, which is thought to be a
402 mechanism of DosS activation (26, 50, 52).

403 In summary, we have shown that H₂S binds to the redox sensor DosS to increase its
404 autokinase and phosphate transfer activity, which induces the Dos dormancy regulon (Fig. 7). We
405 have also shown that physiological levels of H₂S are sufficient to induce the Dos regulon via DosS.
406 The ability of *Mtb* to induce the Dos regulon in response to four physiologically relevant
407 gasotransmitters points to a sophisticated signal transduction system to ensure *Mtb* persistence.

408

409 **Materials and Methods**

410 **Mycobacterial strains and culturing conditions.** *Mtb* H37Rv (BEI Resources (NR-123) and the
411 *Mtb* H37Rv deletion mutant $\Delta dosS$ (provided by Dr. David Sherman, University of Washington)
412 were grown at 37 °C with shaking in Middlebrook 7H9 medium (Difco) supplemented with 10%
413 (vol/vol) ADS (albumin, dextrose, sodium chloride), 0.2% glycerol and 0.02% tyloxapol. Sodium
414 sulfide ($Na_2S \cdot 9H_2O$) (Sigma) was added to culture media as an H_2S donor. Stock solutions of Na_2S
415 were prepared in argon-deoxygenated 50 mM sodium phosphate buffer (150 mM NaCl, 5%
416 glycerol, pH 7.4) and contained 100 μM DTPA (Diethylenetriaminepentaacetic acid) as a metal
417 chelator.

418

419 **Expression and Purification of Recombinant DosS and DosT.** DosS and DosT were expressed
420 in Rosetta (DE3) BL21 *E. coli* cells grown at 37 °C in LB medium as reported previously (10). Briefly,
421 bacterial cells were grown at 37 °C to an OD_{600} of 0.5-0.6, at which point hemin (Sigma) was added
422 to a final concentration of 20 μM hemin and protein production was induced by addition of IPTG to
423 a final concentration of 0.4 mM. Cells were grown overnight at 18°C, collected by centrifugation,
424 and lysed by sonication in sodium phosphate buffer (pH 7.4). Soluble proteins were extracted by
425 using Profinity IMAC Ni-charged resin (Bio-Rad, Hercules, CA) as recommended by the
426 manufacturer. Following elution, proteins were dialyzed against sodium phosphate buffer (pH 7.4)
427 to remove imidazole.

428

429 **UV-Vis Absorption Spectroscopy.** The absorbance spectra of recombinant DosS and DosT (3
430 μM) were determined at room temperature using quartz cuvettes with rubber stoppers in a DU800
431 spectrophotometer (Beckman Coulter, Fullerton, CA). The reduction of recombinant DosS was
432 achieved via addition of sodium dithionite (DTH) to a final concentration of 100 μM while inside an
433 anaerobic glove box (Plas-Labs, Inc. Lansing, MI). A Hamilton syringe with a thin-gauge needle
434 was used to add Na_2S to protein solutions contained in quartz cuvettes (Spectrocell) sealed with a
435 screw cap containing a rubber septum.

436

437 **EPR Spectroscopy.** Purified DosS protein (10 μM) in sodium phosphate buffer with and without
438 Na_2S was transferred to thin-walled quartz EPR sample tubes (Wilmad Glass, Buena, NJ) and
439 snap-frozen in liquid nitrogen. Cryogenic (7K) EPR was measured on a Bruker EMX spectrometer
440 operating at a frequency of 9.39 GHz, 15-G modulation amplitude, 33 db power, 81.92-ms time
441 constant, and 41.94-s sweep time. Each sample was scanned 4-8 times and the average taken.

442

443 **Determination of K_D^{app} for H_2S binding to DosS.** The K_D^{app} for H_2S binding to DosS was
444 determined by difference spectroscopy using the Soret region. Fractional saturation was

445 determined assuming complete occupancy at 100 μM $[\text{H}_2\text{S}] + [\text{HS}^-]$; this was verified by comparing
446 to 200 μM .

447

448 ***In Vitro* Phosphorylation Assay.** *In vitro* autokinase assays were performed essentially as
449 described (33). Briefly, recombinant DosS (6 μM) alone or in the presence of 100 μM Na_2S was
450 assayed for its ability to autophosphorylate in a reaction containing 50 μCi of $[\gamma\text{-}^{32}\text{P}]\text{-ATP}$
451 (6000 Ci/mmol, PerkinElmer Health Sciences), 100 mM Tris-HCl, pH 7.4, 5 mM MgCl_2 , 50 mM KCl_2
452 in a final volume of 20 μl . The reaction was carried out at room temperature, and 4 μl aliquots of
453 reaction mixture were removed from the reaction at various time points between 0 to 60 mins. The
454 reaction at each time point was stopped by adding 2X SDS-PAGE sample buffer. The samples
455 were resolved on 4-20% gradient PAGE gels (Bio-Rad) without heating. Resolved proteins were
456 transferred to a PVDF membrane which was exposed to a phosphor screen (Amersham) overnight.
457 The phosphor screen was scanned on a FLA7000IP Typhoon Storage Phosphorimager.
458 Transphosphorylation assays were performed as above, except that DosS and DosR proteins were
459 present in a molar ratio of 1:6, respectively.

460

461 ***Mtb* RNA Isolation and qRT-PCR Analysis.** *Mtb* was grown to an OD_{600} of 0.4-0.5, and then
462 Na_2S was added to a final concentration of 50 μM . After incubation at 37 $^\circ\text{C}$ for 30 min, a volume
463 of 4M guanidine thiocyanate solution equal to the culture volume was added, and cells were
464 collected by centrifugation. The cell pellet was washed 1X time with PBS and then suspended in
465 1 ml RNAPro solution (MP Biomedicals). The cells were then added to a 2 mL tube containing 0.1
466 mm Lysing Matrix B beads (MP Biomedicals) and were lysed using a Fastprep-24 bead beater(MP
467 Biomedicals). The lysate was centrifuged to pellet cell debris. 500 μl of chloroform was added to
468 the supernatant fraction which was then vortexed and centrifuged for phase separation. Total RNA
469 was isolated from the aqueous layer using a Qiagen RNA isolation kit, following the manufacturer's
470 instructions. RNA was treated with DNaseI prior to qRT-PCR analysis. 500 ng of DNaseI-treated
471 RNA was used for cDNA synthesis using the iScript cDNA synthesis kit (Bio-Rad, USA).
472 Quantitative RT-PCR was performed using SsoAdvanced SYBR green supermix (Bio-Rad, USA)
473 with the Bio-Rad CFX96 detection system according to the manufacturer's instructions.
474 Quantitative RT-PCR reactions were performed in duplicate using three independent *Mtb* cultures.
475 Relative changes in gene expression were determined using the $2\Delta\text{Ct}$ method (59), where Ct
476 values of target genes were normalized to Ct values of *Mtb sigA* mRNA. Relative changes were
477 determined as the ratio of expression of genes between untreated and Na_2S -exposed cultures.
478 Primers used in this study are listed in [Table 1](#).

479

480 **Macrophage infection and isolation of intracellular *Mtb*.** RAW 264.7 macrophages (ATCC)
481 were cultured in DMEM medium (Gibco) containing 10% heat-inactivated FBS and 10 mM HEPES

482 to maintain pH 7.4. RAW 264.7 macrophages were grown in 75 cm² flasks and infected at a MOI
483 of 1:10 for 2 hours using *Mtb* in log-phase growth. After removing infection media, cells were
484 washed 2X with fresh DMEM medium and incubated in complete media containing the desired
485 concentration of L-cysteine (0-2.0 mM) for 24 hours. The L-cysteine-containing media was
486 removed and cells were washed 1X time with fresh media. Next, intracellular *Mtb* bacilli were
487 isolated essentially as described by Rohde et al. (60) with a few modifications. Briefly, infected
488 macrophages were lysed in a solution of 4M guanidine thiocyanate, 0.5% Na N-lauryl sarcosine,
489 25 mM sodium citrate, and 0.1 M β -mercaptoethanol. Lysate were vortexed and passed through a
490 21 gauge needle ten times to shear genomic DNA and reduce viscosity. Intracellular mycobacteria
491 were recovered by centrifugation at \sim 2,700 x g for 30 min. The bacterial pellet was suspended in
492 100 μ l PBS and lysozyme was added to a final concentration of 0.1 mg/ml and incubated for 30 min
493 at RT. The bacilli were lysed in 1 ml Trizol heated to 65°C with 0.1 mm Lysing Matrix B silicon
494 beads using a Fastprep-24. Next, the lysate was centrifuged to pellet cell debris and the
495 supernatant fraction was treated with 500 μ l of chloroform, vortexed, and centrifuged. RNA was
496 isolated from the aqueous layer using a Qiagen RNA isolation kit, following the manufacturer's
497 instructions. RNA was DNase treated prior to qRT-PCR analysis. A negative control group
498 containing bacteria only was treated with complete DMEM was used to determine baseline gene
499 expression levels.

500

501 **Detection of Intracellular H₂S Using the WSP-5 Probe.** The H₂S-specific WSP-5 fluorescent
502 probe (CAS # 1593024-78-2) was prepared according to the manufacturer's instructions. Cell
503 staining was performed as described (36) with some modifications. Briefly, 5x10⁴ RAW 264.7
504 macrophages were plated in each well of 96 well plate and incubated overnight in DMEM medium
505 containing the desired concentration of L-cysteine. The next day the medium was removed, and
506 fresh medium containing 100 μ M CTAB, the desired L-cysteine concentration, and 50 μ M WSP-5
507 probe was added. After 30 min, the WSP-5-containing medium was removed, the cells were
508 washed once with PBS and placed in fresh PBS for imaging. Fluorescence imaging was performed
509 using Biotek Cytation 5 plate reader.

510

511 **Computational Methods**

512

513 **Starting Structure.** The starting structure of the DosS heme-containing GAF-A domain was built
514 from the corresponding X-ray structure (PDB entry code 2w3e) (22). Protonation states of amino
515 acids were those corresponding to their physiological state at neutral pH (i.e., Asp and Glu
516 negatively charged, Lys and Arg positively charged). His149, corresponding to the heme proximal
517 ligand, was simulated in the HID state (with protonated N δ). The remaining His residues were
518 simulated favoring H-bond formation. Particular care was taken for His⁸⁹ which is part of the distal

519 H-bond network. Since the template crystal structure consists of a truncated GAF-A domain, a C-
520 terminal carboxyl group is present that is not present in the full structure. To account for this, a N-
521 CH₃ “cap” was added to this carboxyl groups to avoid interaction with nearby positive residues. The
522 system was solvated by constructing an octahedral box of 10-12 Å using AmberTools.

523

524 **Classical Molecular Dynamics (MD) Simulations.** The MD simulations were performed using
525 the Assisted Model Building with Energy Refinement (AMBER) package (61). Heme, as well as
526 bound and free H₂S and HS⁻ parameters, were taken from our previous work related to H₂S/HS⁻
527 binding to heme proteins (28). All MD simulations were performed using periodic boundary
528 conditions, SHAKE algorithm and the particle mesh Ewald (PME) summation method for treating
529 the electrostatic interactions using default AMBER 16 parameters (61). Each system was first
530 optimized, and then slowly equilibrated to reach proper temperature and pressure values using the
531 Langevin thermostat and Berendsen barostat (62). All four truncated endpoints of the protein
532 (exposed to solvent) were simulated with mild restraints applied to them to maintain α-helix
533 structure.

534

535 **Ligand Binding Free Energy Profiles.** To determine H₂S/HS⁻ binding free energy profiles, we
536 used our previously developed and extensively used Steered MD (sMD) approach combined with
537 Jarzynski’s equation (63, 64). Briefly, in each simulation the ligand is pulled towards the heme iron
538 inside the protein active site using a harmonic guiding potential at constant speed, and the
539 corresponding work performed on the system is recorded. Several simulations are performed
540 starting from different initial conformations with the ligand outside the protein, and finally the
541 corresponding free energy profile is obtained combining the corresponding work profiles using
542 Jarzynski’s equality. We employed 98 different trajectories for H₂S and 89 for HS⁻. In all cases,
543 the guiding coordinate was the Fe-S distance, using a force constant of 200 kcal/mol.Å² and a
544 speed of 0.0025 Å/ps.

545

546 **Hybrid Quantum Mechanics/Molecular Mechanics Simulations.** QM/MM simulations were
547 performed with our own extensively tested and developed code, called LIO which combines a
548 Gaussian-based density functional theory (DFT) approach implemented in CUDA with the AMBER
549 force field and is implemented as a QM/MM option in AMBER (65). System parameters were the
550 same as in our previous QM/MM work on heme proteins (26, 28). Briefly, the heme, its proximal
551 and distal ligands, a nearby water molecule and a heme propionate define the QM subsystem, and
552 the remaining protein and solvent atoms the classical system. Covalent bonds between QM and
553 MM subsystems were treated using the Link atom method (66). The QM system is simulated using
554 a double Z plus polarization basis set (67) and the PBE exchange-correlation functional (68). Heme

555 iron in the Fe³⁺ state bound to H₂S/HS⁻ was simulated in the low-spin state. We performed a 2.0
556 ps MD simulation, starting from a snapshot extracted from the previously optimized MD simulation.
557

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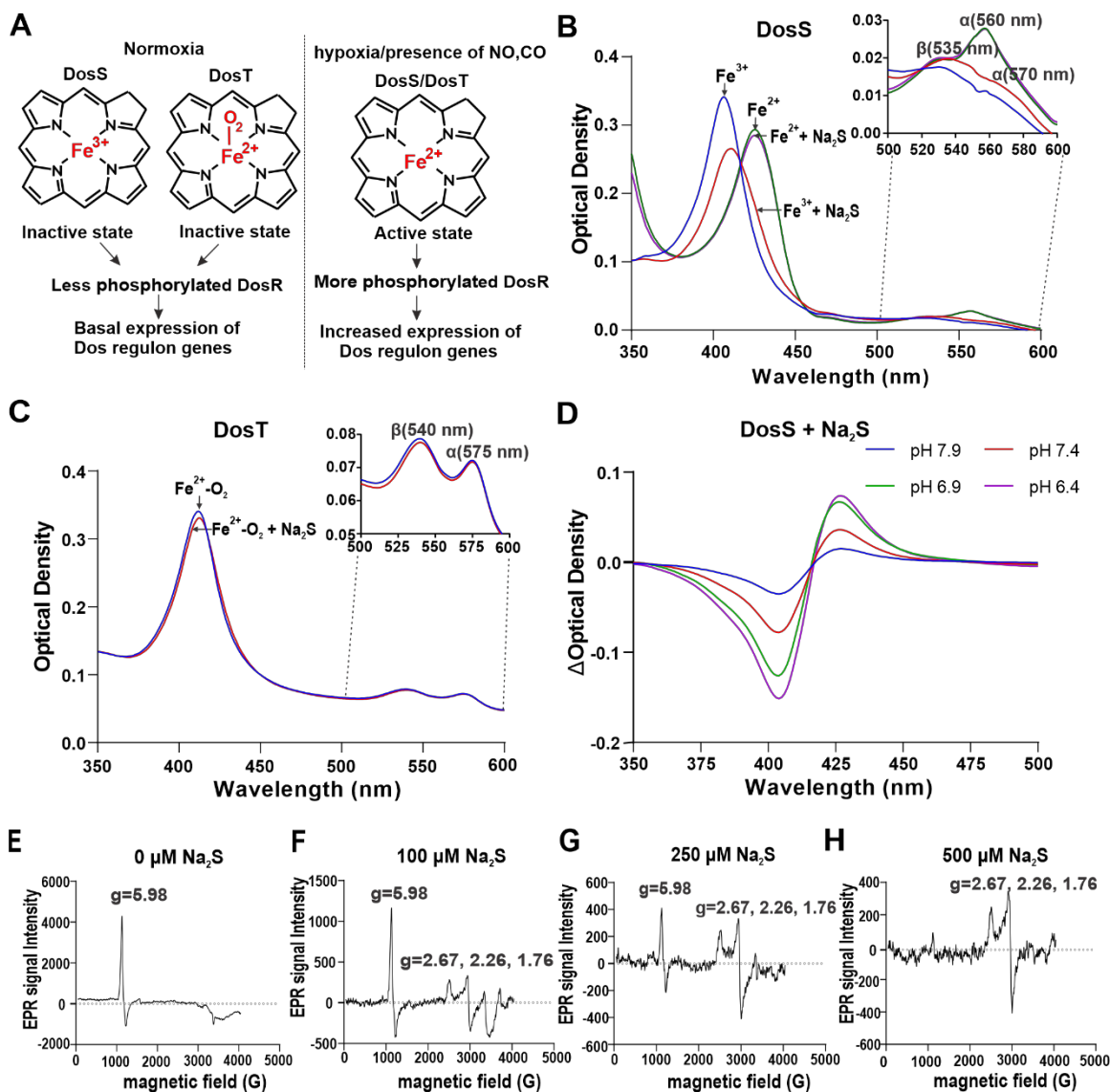
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738 **Figures**

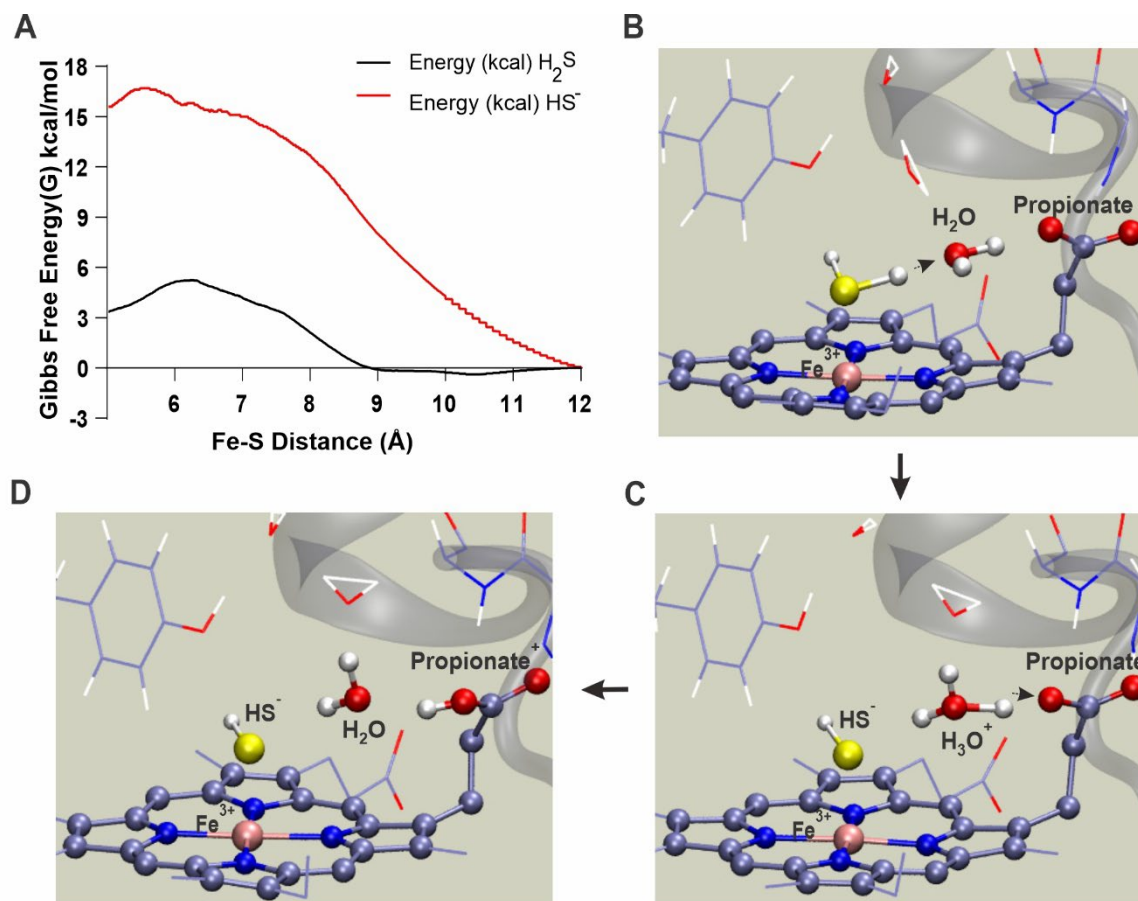
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741 **Fig. 1. Characterization of DosS binding to H₂S.** (A) Depictions of DosS and DosT sensing of
 742 O₂, NO and CO under various conditions. (B) Representative UV-Visible absorption spectra of
 743 recombinant DosS (3 μM) in the Fe³⁺ form (blue curve), the Fe³⁺ form in the presence of 100 μM
 744 Na₂S (red curve), the Fe²⁺ form in the presence of 100 μM DTH (green curve) and the Fe²⁺ form in
 745 the presence of 100 μM DTH and 100 μM Na₂S (orange curve). (Inset) Absorption spectra replotted
 746 to highlight the α and β absorption peaks of the Fe³⁺ form (α at 570 nm and β at 535 nm), and the
 747 Fe²⁺ form (α at 560 nm). (C) Representative UV-Visible absorption spectra of recombinant DosT
 748 (3 μM) in the Fe²⁺-O₂ form (blue curve) and the Fe²⁺-O₂ form in the presence of 100 μM Na₂S (red
 749 curve). (Inset) Absorption spectra replotted to highlight the α (570 nm) and β (535 nm) peaks. (D)
 750 Changes to the UV-Visible absorption spectra of the Fe³⁺ form of recombinant DosS (3 μM)

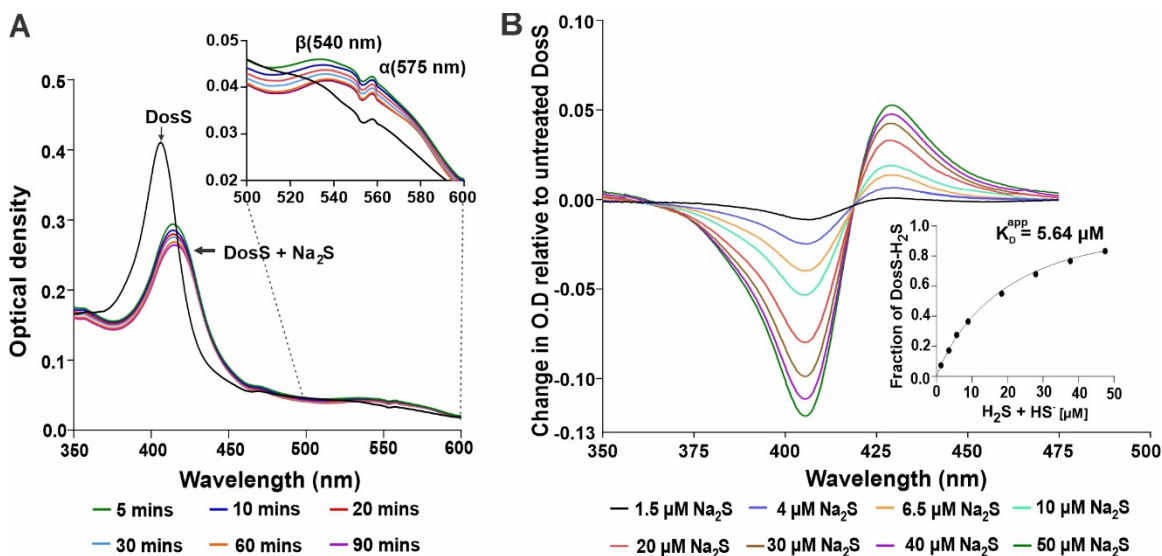
751 resulting from the addition of 25 μM Na_2S at different pH conditions, relative to DosS without Na_2S .
752 (*E-H*) EPR spectroscopic analysis of recombinant DosS showing a single axial peak characteristic
753 of paramagnetic high-spin Fe^{3+} heme iron. DosS alone (*E*), and in the presence of increasing
754 concentrations of Na_2S (*F-H*) which shows the appearance of additional peaks characteristic of the
755 low-spin state of heme iron. UV-Vis spectra are representative of at least 5 independent
756 measurements.
757



758

759 **Fig. 2. Molecular modeling of DosS interaction with H₂S.** (A) Calculated average association
760 free energy profiles for H₂S and HS⁻ as a function of intermolecular distance between the DosS
761 heme iron and sulfur. Free energy values were generated by employing 98 separate trajectories
762 for H₂S and 89 trajectories for HS⁻ using a steered Molecular Dynamics (sMD) approach. (B-D)
763 QM/MM MD simulation snapshots depicting the steps of a predicted proton transfer from H₂S to a
764 nearby heme propionate group through a water bridge within 0.3 ps of MD. The QM subsystem
765 atoms are shown in ball and stick representation.

766



767

768 **Fig. 3. H₂S does not reduce the DosS heme iron and binds with low micromolar affinity (A)**

769 Representative UV-Visible spectra of the Fe³⁺ form of DosS (3 μM) for 5-90 minutes following the

770 addition of 200 μM Na₂S. (*Inset*) Absorption spectra replotted to highlight the α (570 nm) and β

771 (535 nm) peaks. The lack of red shift over time following addition of Na₂S as well as the absence

772 of a strong α peak at 560 nm indicate that the DosS heme iron is not reduced in the presence of

773 sulfide within 90 minutes. A small spectral artifact caused by spectrophotometer filter switching is

774 present at ~550 nm. (B) Representative changes to the UV-Visible spectra of recombinant DosS

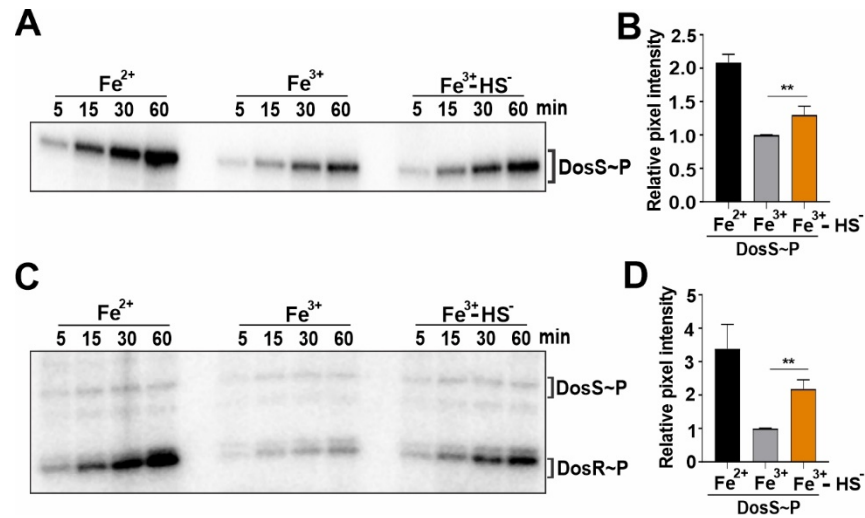
775 (3 μM) resulting from the addition of different concentrations of Na₂S, relative to DosS without Na₂S.

776 (*Inset*) Substrate saturation curve of DosS-H₂S binding was generated using UV-Vis absorption

777 data points obtained from the titration DosS with increasing concentrations of Na₂S. UV-Vis spectra

778 are representative of at least 3 independent assays.

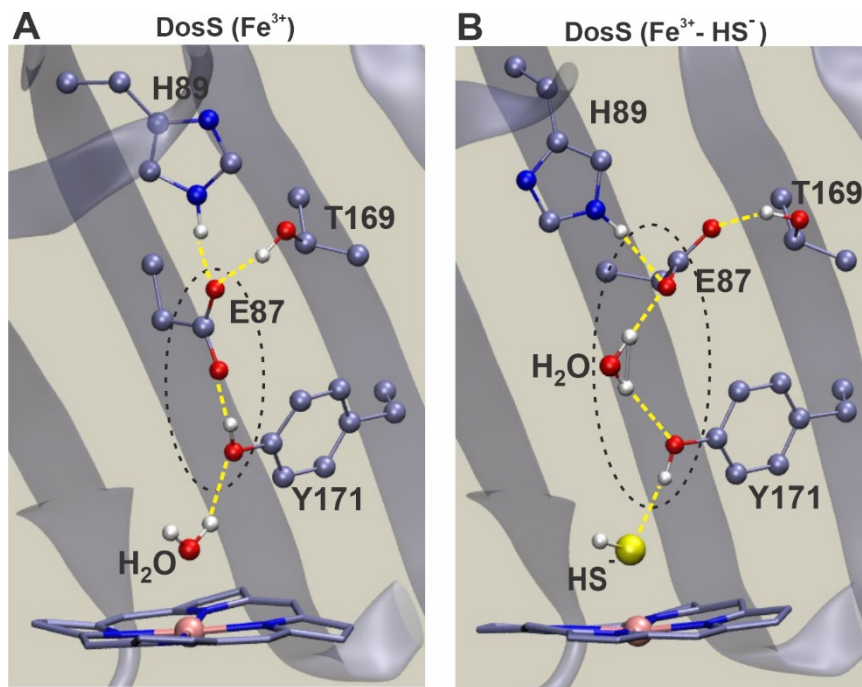
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781 **Fig. 4. H₂S stimulates DosS and DosR phosphorylation.** (A) Representative autoradiogram of
782 PAGE-resolved ³²P-labeled recombinant DosS following autophosphorylation in the presence of γ -
783 ³²P-ATP alone (left and center panels) or with γ -³²P-ATP in the presence of 100 μ M Na₂S (right
784 panel). (B) Densitometric quantitation of DosS bands in (A) at 60 min (n=3). (C) Representative
785 autoradiogram of PAGE-resolved recombinant DosS and DosR following phosphorylation of DosR
786 by γ -³²P-labeled DosS alone (left and center panels) or with γ -³²P-labeled DosS in the presence of
787 100 μ M Na₂S (right panel). These reactions were performed by adding γ -³²P-labeled ATP to a
788 reaction containing both DosS and DosR and analyzed at different time points. (D) Densitometric
789 quantitation of DosR bands in (C) at 60 min (n=3). Data in (B) and (D) are shown as the mean \pm
790 SEM and were analyzed using one-way ANOVA with Tukey's multiple comparisons test performed
791 using GraphPad Prism version 9. **P < 0.01. Autoradiograms are representative of at least 3
792 independent assays.

793



794

795 **Fig. 5. Sulfide binding alters the hydrogen bonding network in the DosS heme pocket (A)**

796 Hydrogen bonding network in the distal domain of ferric (Fe³⁺) DosS showing an intact H-bond

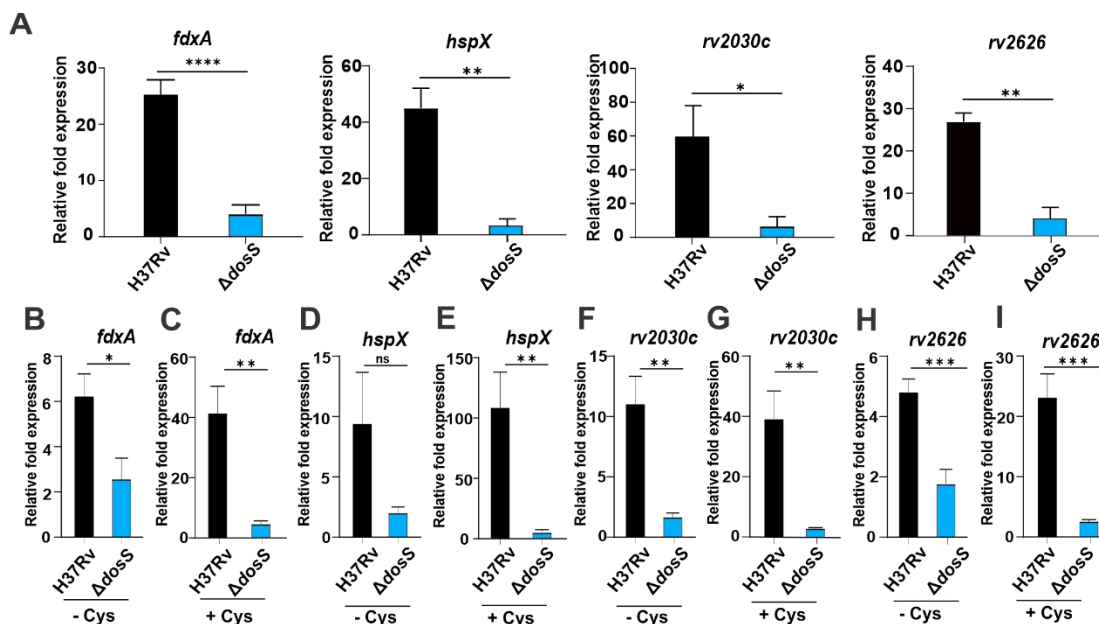
797 between glutamate (E87) and tyrosine (Y171). (B) MD-modeled hydrogen bonding network in the

798 distal domain of ferric (Fe³⁺) DosS in the presence of sulfide showing disrupted H-bonding between

799 glutamate (E87) and tyrosine (Y171). Predicted changes in the H-bonding patterns may lead to

800 structural changes which alter DosS kinase activity upon sulfide binding.

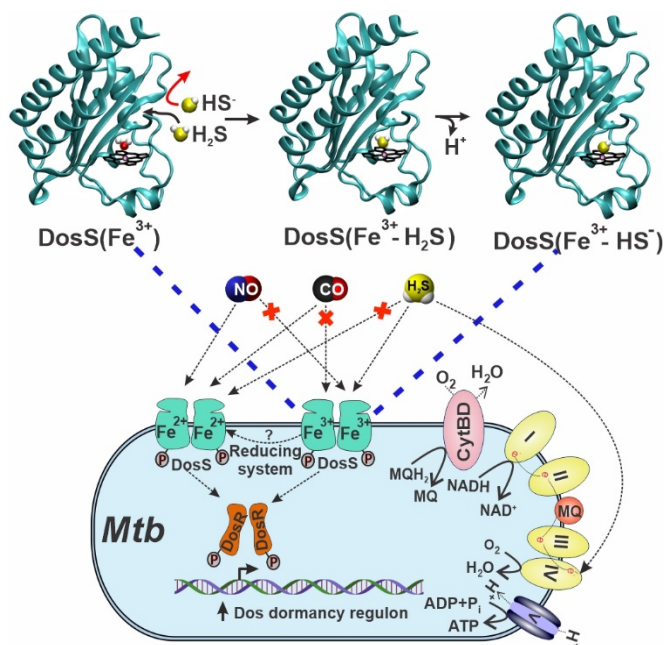
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803 **Fig. 6. Effect of H₂S on expression of DosR regulon genes.** (A) Expression of representative
 804 Dos regulon genes in WT and Δ dosS *Mtb* cells exposed to 50 μ M Na₂S for 30 mins, relative to
 805 unexposed *Mtb* cells (n=6, 3 independent experiments with qRT-PCR performed in duplicate). (B-
 806 I) Expression of representative Dos regulon genes in WT and Δ dosS *Mtb* isolated from infected
 807 RAW 264.7 macrophages grown with or without 0.2 mM L-cysteine for 24 hours (n=6, 3
 808 independent experiments with qRT-PCR performed in duplicate). RT-qPCR gene expression data
 809 from macrophage-isolated *Mtb* are shown relative to gene expression in WT and Δ dosS *Mtb*
 810 cultures exposed to DMEM. Data are shown as the mean \pm SEM and were analyzed using a
 811 unpaired t-test performed using GraphPad Prism version 9. *q < 0.05, **q < 0.01, ***q < 0.001 and
 812 ****q < 0.0001.

813



814

815 **Fig. 7. Schematic of a proposed mechanism of H₂S sensing by *Mtb*.** H₂S, and not HS⁻ (red
816 arrow), can enter the DosS heme pocket. Upon entering the heme pocket, H₂S binds to DosS only
817 in the ferric (Fe³⁺) state, demonstrating that sulfide sensing is different than for NO and CO, which
818 require ferrous (Fe²⁺) DosS. Similar to NO and CO, binding of H₂S to DosS increases autokinase
819 activity which ultimately leads to increased expression of Dos dormancy regulon genes.

820

821 **Table 1: Comparison of Biophysical Parameters of Sulfide-Binding Proteins**

Protein	UV-vis (nm)	EPR (g-values)	K _D (μM)	Reference
<i>Mtb</i> DosS	415, 535, 570 (s)	2.67, 2.26, 1.76	5.64 (H ₂ S)	This work
<i>L. pectinata</i> Hb I	425, 545, 573 (s)	2.67, 2.24, 1.84	0.09 (X _{ST})	(1)
Human Hemoglobin	423, 577, 541	2.51, 2.25, 1.86	17 (X _{ST})	(2)
Bovine Hemoglobin	425, 542, 575(s)	2.55, 2.26, 1.88	7(X _{ST})	(3)
Equine Myoglobin	427, 547, 580	2.56, 2.25, 1.83	96 (X _{ST})	(4), (5)
Human Myeloperoxidase	432, 625	2.567, 2.274, 1.850; 2.512, 2.262, 1.875	< 12 (X _{ST})	(6)
Microperoxidase	414, 536, 568	--	219 (calc) (X _{ST})	(7)
Truncated <i>B. subtilis</i> Hb	427, 550, 577	--	0.2 (X _{ST})	(8)
Truncated <i>T. fusca</i> Hb	425, 550, 575	--	0.36 (X _{ST})	(8)
<i>E. coli</i> DOS (heme-regulated TCS PDE)	427,546, 579	--	ca. 200 (X _{ST}) (based on PDE activation)	(9)
<i>Anaeromyxobacter</i> AfGcHK (heme-regulated TCS His kinase)	423-426,549-551	--	200-5000 (X _{ST}) for binding, although no autokinase activation with 10 mM	(10)

822

823 Note: Hemeproteins that get reduced on sulfide binding do not show the EPR spectrum and
824 hence were not used for comparison eg., cytochrome c oxidase, neuroglobin, cytochrome c.

825 H₂S = based on H₂S concentration

826 X_{ST} = based on concentration of total sulfide species (H₂S + HS⁻ + S²⁻)

827 TCS = two-component system

828 (s) = shoulder

829 calc = calculated

830

831 **Table 2. Comparison of structural and electronic parameters of H₂S and HS⁻ bound**
832 **states obtained from QM/MM calculations**

Parameter	H ₂ S	HS ⁻
dFe-S	2.35 Å	2.17 Å
H ₂ S or HS ⁻ Mulliken population	0.5543	0.0984
dFe-NHis	2.06 Å	2.09 Å
dS-H1	towards Tyr 1.36 Å	1.35 Å
dS-H2	towards water 1.39 Å	
H1 Mulliken population	towards Tyr 0.2242	S-H 0.1698
H2 Mulliken population	towards water 0.2644	Tyr-H 0.4458

833