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1 A generalized kinetic model describes ion-permeation mechanisms in

2 various ion channels

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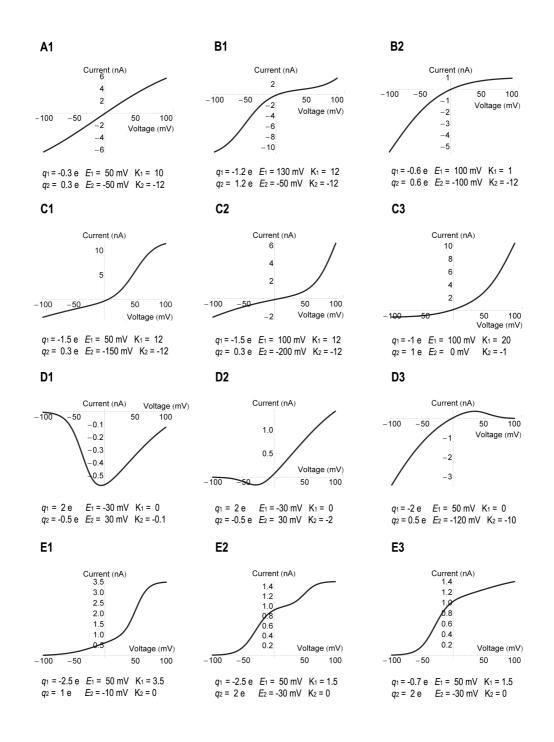
13 Abstract

14 Ion channels conduct various ions across biological membranes to maintain the 15 membrane potential, to transmit the electrical signals, and to elicit the subsequent 16 cellular responses by the signaling ions. Ion channels differ in their capabilities to 17 select and conduct ions, which can be studied by the patch-clamp recording method 18 that compares the current traces responding to the test voltage elicited at different 19 conditions. In these experiments, the current-voltage curves are usually fitted by a 20 sigmoidal function containing the Boltzmann factor. This equation is quite successful 21 in fitting the experimental data in many cases, but it also fails in several others. 22 Regretfully, some useful information may be lost in these data, which otherwise can 23 reveal the ion-permeation mechanisms. Here we present a generalized kinetic model 24 that captures the essential features of the current-voltage relations and describes the 25 simple mechanism of the ion permeation through different ion channels. We 26 demonstrate that this model is capable to fit various types of the patch-clamp data and 27 explain their ion-permeation mechanisms.

29 Introduction

30 Cell membranes, made of lipid bilayers, are impermeable to inorganic ions. 31 Various ions cross the membrane via the specific ion channels down their electrical-32 chemical gradient, or against this gradient at the expense of the extra energy, e.g., via 33 the hydrolysis of the ATP molecules. Due to the availabilities and functions of 34 different ion channels, pumps, and transporters at different locations, the 35 concentrations of the various ions are maintained at the different levels in specific 36 locations separated by the cell membranes, making ions as the important signaling 37 molecules. In these processes, the transmembrane ion channels play important roles, 38 owning to their varied capabilities to select and conduct ions across the membrane. 39 The dysfunction of the ion channels can lead to various diseases (Ackerman & 40 Clapham, 1997; Ashcroft, 2006; Lehmann-Horn & Jurkat-Rott, 1999).

41 Ion channel functions are usually studied by the patch-clamp recording method 42 (Neher & Sakmann, 1976; Neher, Sakmann, & Steinbach, 1978), where the electrical 43 currents are recorded responding to a series of the test voltages elicited. The shapes of 44 the current-voltage curves (typical curves are shown in Fig. 1) are helpful in 45 elucidating the channel functions, e.g., they can provide the information such as the 46 magnitude of the inward and outward currents indicating the inward or outward 47 rectifications, the steepness of the current slopes, the reversal voltage, etc. To 48 compare the functions of the wild-type and mutant channels, or to compare the same 49 channel under the different situations, multiple curves are usually plotted together for 50 a quick and qualitative description of the channel functions. For the quantitative 51 comparisons, and for the explanation of the ion-permeation mechanisms, we must use 52 a model that fits the data.



53

54 Figure 1.

55 Typical current-voltage curves represented by the three-state model. Mechanism 56 m3A3 was used for curves D1 and D2, and mechanism m3A2 was used for all other 57 curves. The model parameters are listed below each curve. The current has the same 58 unit as Ea, which was set arbitrarily to 1 nA for mechanism m3A2 and -1 nA for 59 mechanism m3A3. All temperatures were set to 22 °C.

60

61 In 1952, Hodgkin and Huxley showed that the relation of the channel open 62 probability P_0 and the test voltage V followed the sigmoidal equation containing the 63 Boltzmann factor (Hodgkin & Huxley, 1952) which, through later analyses, led to the 64 formula $P_o = 1/(1 + \exp(-q(V - V_{1/2})/k_BT))$. This equation is commonly referred to as 65 "Boltzmann equation". Here q represents the electric charge, $V_{1/2}$ is the half-activation 66 potential, k_B is Boltzmann's constant and T is the absolute temperature. This equation 67 captures the essential features of the two-state ion permeation processes. It describes 68 not only the channel open probability, but also the normalized current (I/I_{max}) as a 69 function of the applied voltage, which has been quite successful in fitting the patch-70 clamp data obtained from many ion-permeation processes and are still widely used 71 nowadays.

72 Being a two-state model, the Boltzmann equation has limitations, that it does not 73 describe the multistage or the two-direction permeation data. To use the model, all 74 data are converted to within the range of 0 to 1 (normally done by dividing the 75 maximum current value). However, many data also show the negative current in 76 addition to the positive current (curves in the first three rows of Fig. 1), and some 77 curvatures (Fig. 1) are hardly fitted by the Boltzmann equation. Therefore, more 78 sophisticated models are developed, helpful to solve the problems in various aspects 79 (Bezanilla, 2018; Chowdhury & Chanda, 2011, 2012; Chowdhury, Haehnel, & 80 Chanda, 2014; Horng, Eisenberg, Liu, & Bezanilla, 2019; Islas & Sigworth, 2001; 81 Sigg, 2014). Lacroix et al. developed a three-state model (Lacroix et al., 2012), 82 successfully fitting the multistage charge-voltage curves obtained from many gating-83 current experiments of the Shaker K⁺ channel (Carvalho-de-Souza & Bezanilla, 2018; Lacroix, Hyde, Campos, & Bezanilla, 2014; Lacroix et al., 2012). Bezanilla et al. 84

85 employed the sequential Boltzmann equations (two Boltzmann equations of different 86 parameters added together) that fitted the gating-charge data of K^+ ion permeating 87 through the mutant Shaker channel (Bezanilla, Perozo, & Stefani, 1994). More often, 88 the higher-rank models (usually the Markov models) with more than three-states are 89 employed that include all possible ion-permeation pathways (Horn & Vandenberg, 90 1984; Vandenberg & Bezanilla, 1991; Zagotta, Hoshi, & Aldrich, 1994; Zagotta, 91 Hoshi, Dittman, & Aldrich, 1994). These Markov models are very helpful to elucidate 92 the allosteric ion permeation mechanisms. Although useful, these models are usually 93 complicated containing multiple steps, and different permeation processes may 94 employ different models, making the predicted parameters unsuitable for comparison 95 among channels.

96 Therefore, a universal model is needed, not only to fit the data but also to explain 97 the ion-permeation mechanism and compare the functions of different channels. Here 98 we develop a generalized kinetic model. When employing three states, it is able to fit 99 the commonly occurred current-voltage curves (typical curves are shown in Fig. 1) 100 obtained from the patch-clamp experiments, and explain their mechanisms. The 101 model is especially helpful to study the two-direction permeation data, which are 102 usually left unfitted. We call this model a generalized model, because it can include 103 other models that are commonly used nowadays. For example, when employing two 104 states, it includes the Boltzmann equation; when employing three states, it includes 105 the existing three-state model. In addition, this model can include the individual 106 mechanisms suitable for each experimental design. With it, we can easily compare the 107 functions of the different channels or the same channel under the different conditions 108 simply by comparing the model parameters. We demonstrate the applicability of this 109 model using several published patch-clamp data.

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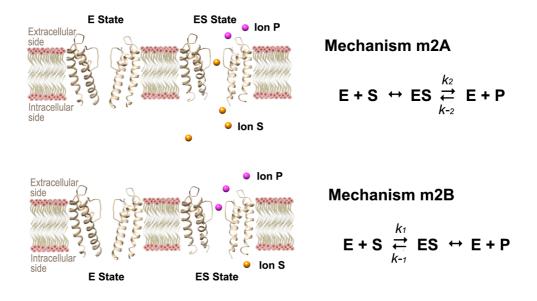
111 Theory and Results

112 Transmembrane ion channels usually contain the gating and the selectivity filter 113 domains, and some contain the extra domains sensing the change of the agonist 114 concentrations or the environmental stimuli, such as the membrane voltage, pH, 115 temperature, pressure, etc. Many physiological studies find that the channel can 116 remain in the closed state (C) when its gate is closed that prohibits ion permeation, or 117 enter the open state (O) when the gate is open that allows the binding and 118 transmission of ions. Therefore, the open and closed states are often used in the two-119 state model: $C \leftrightarrow O$, as was used in deriving the Boltzmann equation (Hodgkin & 120 Huxley, 1952).

121 However, ion permeation through channels can involve more than two states. Indeed, many studies find that the channel can enter the inactivation state as well. For 122 example, the voltage-gated K^+ channel can enter the inactivation state during the 123 124 sustained depolarization stimulus (Choi, Aldrich, & Yellen, 1991; Hoshi, Zagotta, & 125 Aldrich, 1991). The structural studies even revealed the distinct features of the ion-126 channel complex presumably representing the slow inactivation state, e.g., the 127 collapsed conformation of the selectivity filter of the KcsA potassium channel 128 (Cuello, Jogini, Cortes, & Perozo, 2010; Y. F. Zhou, Morais-Cabral, Kaufman, & 129 MacKinnon, 2001), the lack of the ion occupation at the S1 binding site in the 130 selectivity filter of the Kv1.2 potassium channel (Pau, Zhou, Ramu, Xu, & Lu, 2017) 131 with an overall structure perturbed less even in the lipid bilayer environment 132 (Matthies et al., 2018), etc. And many studies suggest that the selectivity filter can be 133 the second gate that modulates the channel inactivation (Cordero-Morales et al., 2006; 134 Liu, Jurman, & Yellen, 1996). Therefore, the inactivation state (I) is commonly

employed in describing the ion-permeation mechanisms in three-state models, which

can be expressed in several ways: $C \rightarrow O \rightarrow I$, $C \Leftrightarrow O$ $\searrow I^{\swarrow} I^{\swarrow}$, etc. However, in addition to 136 137 the inactivation state, the three-state model may also employ the intermediate ion-138 binding state (O_i) when the ion-binding process contains multiple steps or via the 139 multiple subunits (such as when studying the gating charge permeating through the 140 four voltage-sensing subunits of a voltage-gated ion channel) (Lacroix et al., 2014; 141 Lacroix et al., 2012). In these cases, the three-state model can be expressed as $C \rightarrow O_i \rightarrow O$ or some other similar formulas. Hence the three-state model contains 142 143 certainly more than one mechanism, which seems to complicate the model building 144 procedure. Here we try to find a universal model that encompasses all these 145 mechanisms, so that one working equation is enough to handle all types of the ion-146 permeation data. Aimed at deriving a universal model, we no longer use the symbols C, I, O_i, and O in our analyses, instead, we use the general symbols like those often 147 148 used in the kinetic models describing an enzymatic reaction, and denote only the ion-149 unbound states (E and F representing the different states of an apo channel, see 150 captions of Figs. 2-3 for descriptions) and the ion-bound states (ES, FT, and EST 151 representing the channels of state E or F binding ions S or T, see captions of Figs. 2-3 152 for descriptions) in our model. This is because any of these states can have multiple 153 meanings suitable for the specific situation, e.g., the ES state can represent the active channel state O, or the intermediate ion-binding state Oi, or the inactivation state I, 154 155 depending on the different situations. This enables us to write the general equation for 156 the three-state model.

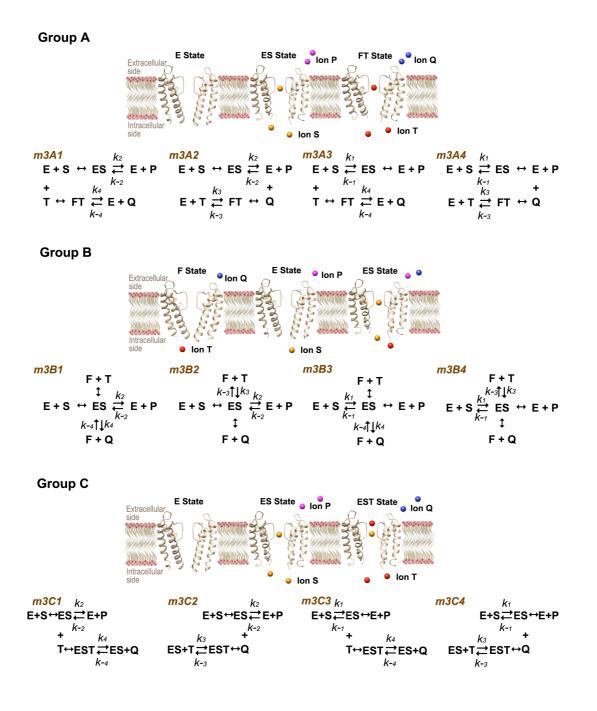




158 *Figure 2*.

159 Mechanisms of the two-state model. The schematic pictures show the E- and ES-state 160 channels embedded in the lipid bilayer. The apo channel is denoted as E, and the ion-161 bound form is denoted as ES. Note that the E state can also have ions bound on it (see 162 text). The crystal structures of the KcsA K^+ channel were used to represent the E (PDB 3FB6) and ES (PDB 3FB7) states. The substrate ion S, located at the 163 intracellular side, changes the symbol to P (the product ion) after transmitted to the 164 165 extracellular side and vice versa. The enzymatic reactions are written beside each 166 schematic picture, describing each process involving two steps: the ion-binding and 167 the subsequent ion-permeation step. The ion-binding step is at equilibrium whenever 168 a double-direction arrow (\Leftrightarrow) shows up. The outward rate constant is written above 169 the right arrow and the inward rate constant is written below the left arrow for each 170 ion-permeation step.

171



172

173 Figure 3.

174 Three groups of mechanisms selected for the three-state model. Mechanisms in group 175 A employ one ion-unbound channel state E and two ion-bound channel states ES and 176 FT, represented by the crystal structures of KcsA K^+ channel (PDB: 3FB6, 3FB7, and 177 3FB8). T can be the same or the different type of ion as S. T is also located at the 178 intracellular side, and changes the symbol to Q after transmitted to the extracellular

179 side. For clarity, we always use the different symbols S and T to describe the mechanism, although they may represent the same type of ion in reality. The ion-180 181 bound state (ES or FT) can have several meanings, representing such as the active 182 channel state, the partial-active or the intermediate state, the active state with the 183 enhanced conductivity, the inactivation state, etc. But these representations are not 184 shown in the schematic picture. Below the picture, four mechanisms of group A are 185 listed, each composed of a pair of equations describing the ion-binding and the 186 subsequent ion-permeation processes. The equilibrium binding step and the relevant 187 rate constants are described similarly as those in Fig. 2. The four mechanisms differ 188 in the ion-permeation directions. Mechanisms in group B employ two ion-unbound 189 states E and F (PDB 3FB5), and one ion-bound state ES, and mechanisms in group C 190 employ one ion-unbound state E and two ion-bound states ES and EST (PDB 3FB7), 191 where the EST state has two ions S and T bound on it. The associated mechanisms are 192 written blow each picture. Note that the F and EST states can also have several 193 meanings, some overlapping those represented by the ES or FT state, but we do not 194 show their multiple meanings in this figure.

195

196 The next question is whether the three-state model is enough to handle all types 197 of the patch-clamp data? This depends on the experimental design and on the shapes 198 of the current-voltage curves obtained. In the normal patch-clamp experiments, the 199 currents are recorded responding to the specified voltage elicited. These recordings do 200 not distinguish the transition of states not involving the change of currents, such as the 201 transitions among several closed states from $C_1 \rightarrow C_2 \rightarrow C_3$, etc. Unless the special 202 experimental designs are employed that can incur current changes from these states, 203 they have to be combined into one closed state due to the lack of information.

204 Similarly, if the transitions among several inactivation states do not incur current 205 changes, they need be combined as well. Although the model can contain multiple 206 open states, whether to employ all these states depends on the shape of the current-207 voltage curve. We suggest starting the fitting procedure with the lower-rank model 208 whenever possible unless otherwise required by the specific experimental design. This 209 is because several parallel or sequential steps can be combined into one step if they do 210 not incur the appreciable current changes. For example, if the events of ion-211 permeation through different subunits (representing the different states) occur 212 simultaneously that each individual state cannot be differentiated by the current curve, 213 these states can be combined as one state. The sequential ion-binding processes E + S \rightarrow ES and ES + S \rightarrow ES₂ are readily combined into one ion-binding process E + 2S 214 215 \rightarrow ES₂ when the data obtained is not sufficient to differentiate the ES state.

Following these ideas, we find that the three-state model is enough to describe the commonly occurred current-voltage curves shown in Fig. 1. Therefore, we focus on explaining the two- and three-state models in this paper.

219

A. The two-state model

220 Two mechanisms exist because the channel can conduct ions in two directions, 221 albeit with different abilities. Here we follow the Michaelis-Menten mechanism and 222 use the equilibrium approximation analysis. In the first mechanism (m2A in Fig. 2), 223 the ion binding at the intracellular side is in equilibrium relative to the outward conduction step, so that $k_2 \ll k_{-1}$ and $k_{-2}[P] \ll k_1[S]$. And in the second mechanism 224 225 (m2B in Fig. 2) the ion binding at the extracellular side is in equilibrium relative to the inward conduction step $(k_2 \gg k_{-1} \text{ and } k_{-2}[P] \gg k_1[S])$. Here we use brackets to 226 227 denote the concentrations of each species, e.g., [S] and [P] represent the 228 concentrations of the substrate ion and the product ion at the intracellular and the extracellular side, respectively. k_i is the rate constant associated with each ionpermeation step labeled on each reaction shown in Fig. 2. Note that the reaction rate is simply the current I, thus I = $k_2[ES] - k_{-2}[E][P]$ and I = $-(k_{-1}[ES] - k_1[E][S])$ for the first and second mechanism, respectively. Now, let Et = [E] + [ES], we obtain a general equation for the two-state model:

234
$$I = \frac{Ea \cdot \left(1 + K_1 \frac{[E]}{[ES]}\right)}{1 + \frac{[E]}{[ES]}}$$
(1)

In the first mechanism, $Ea = k_2 \cdot Et$ and $K_1 = -k_{-2} \cdot [P]/k_2$. And in the second mechanism, $Ea = -k_{-1} \cdot Et$ and $K_1 = -k_1 \cdot [S]/k_{-1}$. Note that the expressions of Ea and K_1 need not be identified at the stage of data fitting that each of them can be considered simply as one parameter of Eq. (1). After fitting the curve, their expressions can be used to explain and compare the channel functions.

The next step is to relate the concentration ratio [E]/[ES] to the test voltage *V*. Hodgkin and Huxley used the open probability theory (Hodgkin & Huxley, 1952), here we use the principles of chemical potentials that enable us to express all the relevant concentration ratios of the different states, such as [ES]/[E], [FT]/[E], [EST]/[ES], etc. According to the thermodynamic principles, at equilibrium, the chemical potentials of the E, S, and ES in the first step of mechanism m2A have the relation: $\mu_E + \mu_S = \mu_{ES}$. Now express each μ using the thermodynamic principle:

247
$$\mu_{\rm E}^{\rm o} + k_B T \ln[\rm E] + q_{\rm E}V + \mu_{\rm S}^{\rm o} + k_B T \ln[\rm S] + q_{\rm S}V$$
$$= \mu_{\rm ES}^{\rm o} + k_B T \ln[\rm ES] + q_{\rm ES}V$$
(2)

Here μ^{o}_{i} denotes the chemical potential of each species *i* under the standard state, *V* is the electrostatic potential (it is the membrane potential in the ion-channel studies), q_i is the electric charge of each species *i*, k_B is Boltzmann's constant, and *T* is

the absolute temperature. Rearranging this equation, we obtain:

252
$$\frac{[E][S]}{[ES]} = e^{-\Delta \mu^{o}/k_{B}T} e^{-q_{I}V/k_{B}T}$$
 (3)

Here $q_1 = q_E + q_S - q_{ES}$, which represents the change of the electric charge before and after the ion binding to the protein. $\Delta \mu^o = \mu_E^o + \mu_S^o - \mu_{ES}^o$. Define that [E] = [ES] when $V = E_1$ (E_1 is the half-activation potential, conventionally denoted as $V_{1/2}$, representing the voltage at which half channels are in the E state and the other half are in the ES state), and $\Delta \mu^o$ can be expressed as a function of E_1 : $e^{-\Delta \mu^o/k_B T} = [S]e^{q_1 E_1/k_B T}$. Inserting this into Eq. (3), and assuming that the solutions contain the abundant permeant ions (so that the ion concentrations change little at *V* and E_1), we obtain:

260
$$\frac{[E]}{[ES]} = e^{-q_1(V-E_1)/k_BT}$$
 (4)

261 Eq. (4) is analogous to the results obtained by the probability theory (Hodgkin & 262 Huxley, 1952), except that q_1 has a slightly different meaning than the one shown in 263 the original Boltzmann equation, here it describes the change of the charges of the 264 ion-protein complex. Because proteins contain the charges, dipoles, quadruples, etc., 265 and their values may also change during the ion-binding process, so the value of q_1 266 need not be integers any more, which is often found in real cases. Normally, a large q_1 267 value can be interpreted as a large number of ions binding to the channel. With this definition of q_1 , the "ion-unbound" state can still have ions bound inside the channel, 268 and this was confirmed by many structural studies that the closed state channels 269 270 bound ions inside their selectivity filters (Doyle et al., 1998; Hite et al., 2015; Tao, 271 Avalos, Chen, & MacKinnon, 2009). Here q_1 shows only the electric charges altered 272 in the ion-protein complex in the "ion-bound" state relative to that in the "ion273 unbound" state. And this is why sometimes the calculated q_1 value is small that is 274 hardly explained by the charges carried by ions. Here, the "ion-unbound" state is 275 merely a name representing one channel state, which may not be the ion-depleted 276 state in reality. However, for the sake of clarity, we still use the ion-depleted structure 277 to denote the "ion-unbound" state in Figs. 2 and 3.

278 Performing the similar analysis for the mechanism m2B, we obtain the general279 formula for the two-state model that relates the current I to the test voltage *V*:

280
$$I = \frac{Ea \cdot \left(1 + K_1 e^{-q_1(V - E_1)/k_B T}\right)}{1 + e^{-q_1(V - E_1)/k_B T}}$$
(5)

When using the normalized current and when $K_1 = 0$, Eq. (5) reduces to the Boltzmann equation. Apparently, the Boltzmann equation is only a limiting case of the two-state model presented by Eq. (5), which occurs when the channel conducts ions in only one direction.

285

B. The three-state model

Many mechanisms exist for the three-state model. We select three groups of 286 287 mechanisms (Fig. 3) that are likely employed in real problems because they can 288 describe all curves shown in Fig. 1. We find that the mechanisms in the first group are 289 especially useful where the channel has one ion-unbound state E and two ion-bound 290 states ES and FT (see Fig. 3). Here ES and FT can represent the channel having the 291 different abilities to conduct ions in the inward and outward directions that results in 292 rectifications (Fig. 1 B1, B2, C1-C3); or they can represent the channel having the 293 different abilities to conduct ions in one direction that results in multistage 294 permeations (Fig. 1 E1-E3); or one of them can represent the inactivation state (bound 295 with ions but not transmitting them) that results in the conditional inactivation curves 296 identified by a noticeable bell shape (Fig. 1 D1-D3). Mechanisms in the second group

297 contain two ion-unbound states E and F, and one ion-bound state ES. Here E and F 298 can represent the different rest-state channels having the different propensities to bind 299 ions competing for the same conductive conformation of the channel, which can 300 include the multistage permeation process; or one of them can represent the 301 inactivation state that results in the conditional inactivation curves. Mechanisms in the 302 third group contain one ion-unbound state E and two ion-bound states ES and EST, 303 where the EST state represents the channel binding ions S and T. Mechanisms in the 304 third group are most suitable for describing the multistage permeation process, but 305 they are not limited to this case, e.g., ES or EST can also represent the inactivation 306 state that results in the bell-shaped inactivation curves. Therefore, multiple 307 mechanisms can lead to the same current-voltage curve. We may select the one that 308 reasonably explains the ion permeation data consistent with the experimental design 309 meanwhile yielding the smallest error in fitting the curve.

Although the mechanisms vary from one another, they all reduce to the same working equation if using the universal parameters Ea, K₁, and K₂ (their expressions for the individual mechanisms are listed in Appendix A):

313
$$I = \frac{Ea \cdot \left(1 + K_1 e^{-q_1(V - E_1)/k_B T} + K_2 e^{-q_2(V - E_2)/k_B T}\right)}{1 + e^{-q_1(V - E_1)/k_B T} + e^{-q_2(V - E_2)/k_B T}}$$
(6)

Here q_1 , q_2 , E_1 , and E_2 have similar meanings as those described for the two-state model (see Appendix A for the definitions of q_1 and q_2 for each mechanism). Ea, K₁, and K₂ have different expressions for each mechanism (Appendix A), but they need not be identified at the stage of the data fitting. This means that we can fit the data simply using Eq. (6), then find the most suitable mechanism based on the parameters. Alternatively, we can choose several mechanisms and fit the data using their specific equations determined by the expressions of Ea, K₁, and K₂ (Appendix A), then select the mechanism that yields the smallest error in fitting the data. Note that when comparing the conduction rates described by the same mechanism, we can directly use the values of Ea·K₁ and Ea·K₂. For example, in the mechanism m3A2, Ea·K₁ = k_2 ·Et and Ea·K₂ = $-k_{-3}$ ·Et, thus comparing the absolute values of Ea·K₁ and Ea·K₂ is similar to comparing k_2 and k_{-3} if Et is kept constant.

326 Eq. (6) is a universal working equation of the three-state model for fitting the 327 data. Can there be a universal mechanism of the three-state model for explaining the 328 data? Among the twelve mechanisms we have proposed (Fig. 3), we find that 329 mechanisms m3A2 and m3A3, differing in the permeation directions, are especially 330 useful that they describe most of the current-voltage curves appearing at different 331 circumstances. Indeed, all curves in Fig. 1 were drawn using these two mechanisms. 332 These curves vary from the simplest ohmic behavior (Fig. 1 A1), to the inward and 333 outward rectifications (Fig. 1 B1, B2, C1-C3), inactivation (Fig. 1 D1-D3), and even 334 the multistage permeations (Fig. 1 E1-E3). Employing the same mechanism enables 335 us to compare the model parameters directly that immediately explains the altered 336 channel functions. For example, the outward rectification curves C1 and C2 look different (Fig. 1), but they differ only in the half-activation potentials ($E_1 = 50 \text{ mV}$, E_2 337 338 = -150 mV in C1 curve, and $E_1 = 100$ mV, $E_2 = -200$ mV in C2 curve). They look 339 different simply because the voltage is restricted to below 100 mV. Besides the shift 340 in the half-activation potentials, changes in q may also contribute to the rectification 341 behavior within a restricted voltage range, e.g., the curve C2 itself is due to a larger 342 number of ions binding to the channel at the intracellular side relative to that at the 343 extracellular side ($q_1 = -1.5$ e, $q_2 = 0.3$ e), accompanied by the shift in the half-344 activation potentials ($E_1 = 100 \text{ mV}$, $E_2 = -200 \text{ mV}$). In addition, a large difference in K₁ and K₂ can also result in rectifications, e.g., the inward rectification curve B2 is 345

346 due to a larger inward conduction rate relative to the outward conduction rate ($K_2 = -$ 347 12, $K_1 = 1$). Thus, comparing model parameters enables us to compare the channel 348 functions directly.

Note that Eq. (6) is not limited to describing the mechanisms shown in Fig. 3, it can also describe other mechanisms employing three channel states, including the previously published three-state model (Lacroix et al., 2012) (see Appendix B). Therefore, Eq. (6) is a more general form of the three-state model.

353

C. The generalized kinetic model (with N states)

If the current curves show apparently two or more intermediate states, then the higher rank model should be used. Generally, the multi-state kinetic model can be expressed as:

357
$$I = \frac{Ea \cdot \left(1 + \sum_{i=1}^{N-1} K_i e^{-q_i (V - E_i)/k_B T}\right)}{1 + \sum_{i=1}^{N-1} e^{-q_i (V - E_i)/k_B T}}$$
(7)

Here the parameters q_i and E_i are the electric charge changed (relative to the reference state) and the half-activation potential associated with each state *i* relative to the reference state defined for each mechanism (e.g., E state is the reference state in mechanism m3A2), and their meanings are explained analogously to those of the twoand three-state models. Like the two- and three-state models, the *N*-state model contains multiple mechanisms, and the values of Ea and K_i depend on the individual mechanisms.

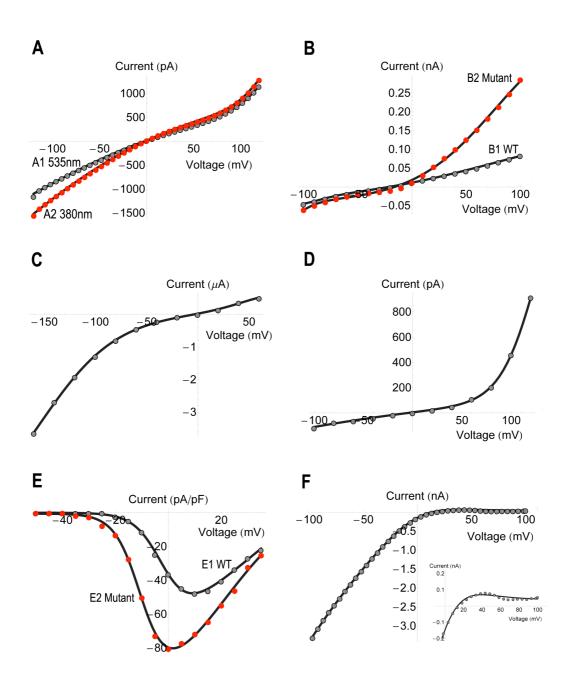
When employing four states, Eq. (7) can include the sequential Boltzmann equations (adding up two Boltzmann equations of the different parameters) (Bezanilla et al., 1994), and their relations are presented in Appendix C. Note that in some circumstances, the four channel states can be regrouped into three states that also reasonably explains the ion-permeation data (Appendix C). Indeed, Bezanilla et al.

- 370 found that the curves fitted by the sequential Boltzmann equations and the three-state
- 371 model were indistinguishable in their case (Bezanilla et al., 1994), confirming that
- 372 two or more states can be combined into one state if not incurring the appreciable
- 373 current changes.
- 374

375 **Discussion**

How does this model work in reality? Here we show several examples using the

- 377 published patch-clamp data. The fitted curves are shown in Fig. 4 and the model
- arameters are shown in Table 1.



379

380 *Figure 4*.

Experimental current-voltage curves fitted by the three-state model. Source data are represented by the filled circles, and the model data are plotted by the solid black curves. Reference to the source data, the selected mechanism, and the fitted parameters are listed in Table 1. The room temperature (22°C) was used for all

- 385 curve-fitting procedures. The inset in Panel F shows the enlarged plot of a local
- 386 *region (positive voltages) of the current-voltage curve.*

387

- 388 Table 1. Three-state model parameters selected for the current-voltage curves shown
- 389 in Fig. 4.

Curves ^a in Fig.	$q_1(e)$	$q_2(e)$	$E_1 (\mathrm{mV})$	$E_2 (\mathrm{mV})$	K_1^{b}	K_2^{b}	Ea ^c	k'_1 ^d	<i>k</i> ′2 ^e	σ^{f}	Ref. ⁸
4A1	-1.32	0.25	135	-143	2.802	-3.994	903 pA	2530.206	3606.582	0.00868	1
4A2	-1.9	0.25	120	-165	1.684	-4.996	1128 pA	1899.552	5635.488	0.00675	1
$4B1^*$	2.227	-0.349	-90.652	81.927	1.228	-4.116	-0.044 nA	0.054	0.181	0.00902	2
4B2*	2.03	-0.66	-128.719	78	7.682	-11.364	-0.0408 nA	0.313	0.464	0.0132	2
4C	-1.25	0.67	46	-154.597	11.494	-102.241	0.0666 µA	0.766	6.809	0.00628	3
4D	-1.361	0.211	125.366	-192.844	16	-4.795	132.217 рА	2115.472	633.981	0.00532	4
4E1*	4.74	-0.88	0	30	0	-0.46	-102.422 pA/pF	0	47.114	0.0138	5
4E2*	5.11	-1.68	-9.562	21	0.00466	-0.0879	-113.74 pA/pF	0.53	9.998	0.0144	5
4F	-1.355	0.26	-2.41	-246.583	0.0222	-14.901	1.799 nA	0.0399	26.807	0.00271	6

390

^a Mechanism m3A3 was employed for the curves labeled with an asterisk, and
 mechanism m3A2 was used for all other curves.

393 ${}^{b}K_{1}$ and K_{2} are dimensionless.

^c Ea has the same unit as the current reported in the source data file, which is also

395 written following each Ea value in the table.

396 ${}^{d}k'_{1} = Et \cdot k_{2} = |Ea \cdot K_{1}|$ for mechanism m3A2, and $k'_{1} = Et \cdot k_{-1} = |Ea \cdot K_{1}|$ for mechanism

397 m3A3. k'_1 has the same unit as Ea.

398 $k'_2 = Et \cdot k_{-3} = |Ea \cdot K_2|$ for mechanism m3A2, and $k'_2 = Et \cdot k_4 = |Ea \cdot K_2|$ for mechanism

399 m3A3. k'_2 has the same unit as Ea.

400 ^{*f*} See Materials and Methods for the definition of σ .

401 ^g Ref. 1 (Lemoine et al., 2020); Ref. 2 (Zheng et al., 2020); Ref. 3 (Chiasson et al.,

402 2017); Ref. 4 (Syrjanen et al., 2020); Ref. 5 (Huang et al., 2019); Ref. 6 (Zhou et al.,

403 *2017*).

404

405 **Two-direction permeation curves**

406 Lemoine et al. have designed an optical molecule attached close to the pore of the

407 GluD2 receptor that senses the light of the specific wavelength to occlude or open the

408 entrance to the channel pore (Lemoine et al., 2020). The two current curves under the

409 535 nm and 380 nm light illuminations (source data from Figure 4A of Ref. (Lemoine 410 et al., 2020)) were fitted with the three-state model and presented in Fig. 4A. 411 Mechanism m3A2 is suitable to explain these curves. It describes the two-direction ion permeations via the two ion-bound states ES and FT. Here ES represents the 412 413 channel state inclined to direct the outward current, and FT represents the channel 414 state inclined to direct the inward current. Compared with the 535 nm light 415 illumination, the 380 nm light increases the inward conduction rate $(k'_{2,535nm} =$ 416 3606.58 pA, $k'_{2,380nm}$ = 5635.49 pA, see captions of Table 1 for the definition of k') 417 and decreases the outward conduction rate $(k'_{1,535nm} = 2530.21 \text{ pA}, k'_{1,380nm} = 1899.55$ pA), assuming that Et is unchanged. This produces the increased inward current at 418 419 negative potentials, reflecting the optical molecule's ability to sense the specific light 420 and magnify the inward current. We also note that it modifies the ion binding at the intracellular side $(q_{1, 535nm} = -1.32 \text{ e}, q_{1, 380nm} = -1.9 \text{ e})$ that more ions bind to the 421 422 channel under the 380 nm light illumination (when the optical molecule switches to 423 the cis-conformation that opens the pore entrance), consistent with the lock property 424 of the optical molecule. In addition, the guanidinium moiety of the optical molecule 425 may also play a role to bring about these changes, possibly through interacting with 426 the positively charged ions upon switching its locations under the different light 427 illuminations.

Zheng et al. solved the cryo-EM structures of a eukaryotic cyclic nucleotidegated channel TAX-4, and found that the double mutations of the hydrophobic residues F403V and V407A in the cavity of the channel can increase the outward basal current (Zheng et al., 2020). The current curves of the wild-type and mutant channels (source data from Figure 4a in Ref. (Zheng et al., 2020)) were fitted with the three-state model and plotted in Fig. 4B. Mechanism m3A3 is suitable for this case,

434 where the ES and FT states describe the channel abilities for the inward and outward 435 conductions, respectively. The cryo-EM structures reveal that F403 and V407 form 436 the hydrophobic gate inside the channel cavity that they block the permeation 437 pathway during the closed state but pave the way for ion permeation by rotating aside 438 upon switching to the open state (Zheng et al., 2020). Mutations to valine and alanine certainly increase the central space in the cavity that makes the channel a bit "leaky" 439 440 in a sense (Zheng et al., 2020). Comparing the model parameters, the mutations increase the conduction rates in both directions ($k'_{1, \text{wt}} = 0.054 \text{ nA}$, $k'_{1, \text{mutant}} = 0.31 \text{ nA}$, 441 and $k'_{2, \text{ wt}} = 0.18 \text{ nA}$, $k'_{2, \text{ mutant}} = 0.46 \text{ nA}$), conforming the "leaky" property of the 442 443 mutant channel. In addition, ion binding at the intracellular side is roughly doubled $(q_{2, \text{wt}} = -0.35 \text{ e}, q_{2, \text{mutant}} = -0.66 \text{ e})$, but that of the extracellular side is slightly 444 decreased ($q_{1, \text{wt}} = 2.23$ e, $q_{1, \text{mutant}} = 2.03$ e), reflecting the increased ability of the 445 446 mutant channel for the outward conduction. This, together with the increased 447 conduction rates in both directions, exhibits the increased outward basal current as the 448 overall effect.

449 **Rectification curves**

Chiasson et al. have reported a *brush* mutation in the cyclic nucleotide-gated channel that resulted in the gain-of-function, manifested by the inward rectification of the Ca²⁺ current (Chiasson et al., 2017). The typical current curve (source data from the *brush* curve of Figure 3D in Ref. (Chiasson et al., 2017)) was fitted by the threestate model and plotted in Fig. 4C. Employing mechanism m3A2, the inward rectification is mainly due to the larger inward conduction rate ($K_2 = -102.24$) relative to the smaller outward conduction rate ($K_1 = 11.49$).

457 Syrjanen et al. have reported the structure of a calcium homeostasis modulator 458 that produced the outward-rectification current (Syrjanen et al., 2020). The typical

459 current curve (source data from the hCALHM1 curve of Figure 1b in Ref. (Syrjanen 460 et al., 2020)) was fitted by the three-state model (mechanism m3A2) and plotted in 461 Fig. 4D. This time, the outward rectification is due to the larger number of ions 462 binding at the intracellular side ($q_1 = -1.36$ e, $q_2 = 0.21$ e), and a larger outward 463 conduction rate ($K_1 = 16, K_2 = -4.8$).

464 The above results, together with the model curves shown in Fig. 1, show the 465 varied reasons that lead to the rectifications. Theoretically, a large difference in K₁ and K₂ can lead to rectifications. But these rectifications may not appear within the 466 467 restricted voltage range selected for the current recordings. Thus the rectification 468 curve shown up in a narrowed voltage range is usually due to a collective action 469 mixed with the individual changes in q, E, and K. Simply comparing q_1 and q_2 or comparing K₁ and K₂ sometimes can lead to inconsistent conclusions, e.g., in Fig. 4C, 470 471 a larger q_1 compared to q_2 ($q_1 = -1.25$ e, $q_2 = 0.67$ e) does not lead to the conclusion of 472 the outward rectification.

473

Bell-shaped curves

474 Huang et al. have reported a mutation of the voltage-gated calcium channel that 475 resulted in the gain-of-function and produced the increased inward current (Huang et 476 al., 2019). The current curves of the wild-type and mutant channels (source data from Figure 3b of Ref. (Huang et al., 2019)) were plotted in Fig. 4E. We find that the 477 mechanism m3A3 is suitable to explain these curves. Here ES represents the 478 479 inactivation channel state at the negative potentials. Note that the "inactivation state" 480 used here is only a general term denoting one nonconducting channel state, which can result from several conditions including the closing of the inner gate. As the test 481 482 voltage increases, the channel gradually recovers from the inactivation state and 483 switches to the E state then to the FT state, which is accompanied by the inward 484 conduction of ions at the less negative to the small positive potentials. At the more 485 positive potentials, the FT state dominates that directs the outward current. This is 486 how the bell-shape curve is formed around 0 mV. Comparing the parameters of the 487 wild-type and mutant channel curves, the mutation increases both outside and inside 488 ion binding $(q_{1, \text{wt}} = 4.74 \text{ e}, q_{1, \text{mutant}} = 5.11 \text{ e}; q_{2, \text{wt}} = -0.88 \text{ e}, q_{2, \text{mutant}} = -1.68 \text{ e})$, shifts 489 both half-activation potentials to the left ($E_{1, wt} = 0 \text{ mV}$, $E_{1, mutant} = -9.56 \text{ mV}$; $E_{2, wt} =$ 490 30 mV, $E_{2, \text{mutant}} = 21 \text{ mV}$), and significantly decreases the outward ion-conduction rate ($k'_{2, wt} = 47.11 \text{ pA/pF}$, $k'_{2, mutant} = 10 \text{ pA/pF}$). These changes lead to the larger 491 492 inward current shifted to the left of the wild-type current curve (Fig. 4E), which can 493 be interpreted as the gain-of-function.

494 Zhou et al. have solved the cryo-EM structures of the human endolysosomal 495 TRPML3 channel and studied its functions (Zhou et al., 2017). The ligand-activated 496 current curve of the wild-type channel (source data from the ML-SA1 ligand-bound 497 current curve in Figure 2b of Ref. (Zhou et al., 2017)) showed not only an inward 498 rectification, but also a bell curvature at positive potentials (Fig. 4F, the inset shows 499 the enlarged plot of the current at positive potentials). We can use the mechanism 500 m3A2 to explain this phenomenon. At negative potentials, the FT state is the active 501 form that directs the large inward currents ($K_2 = -14.9$). As the test voltage increases, 502 the ion permeation changes the direction, and the FT state gradually switches to the E 503 then to the ES state. The ES state is the inactivation state because the channel nearly prohibits ion conduction at the very positive potentials ($K_1 = 0.022$). 504

505 Some

Some further points

The above studies show only the applicability of the three-state model, that it can fit the curve and explain their mechanisms in accordance with the experimental findings. But they tell nothing about the reliability of the model parameters, because 509 only one curve (the averaged data in most cases) is used for each case in our study. 510 The reliability of the model parameters depends on the reproducibility of each data 511 set, and hence is beyond the scope of this paper. However, the model parameters do 512 provide a way to examine the reproduced data sets (by comparing the model 513 parameters obtained from fitting the curve using each data set) and check whether 514 these data values are consistent with one another.

515 Many patch-clamp recordings show the continuously increasing currents not 516 reaching the saturation level, and we may wonder when the current will saturate. This 517 information is easily obtained from the three-state model if the parameters are 518 reliable. For example, in mechanism m3A2, the saturation currents are $Ea \cdot K_1$ and 519 $Ea \cdot K_2$ at the positive and negative potentials, respectively. The saturation currents can 520 occur within or beyond the physiological recording range, and the model can help 521 predict these values or help record these values based on the predicted voltage range.

522 Do the above analyses confirm that the channels switch among only three 523 possible states in those ion-permeation processes? No, the model only suggests that 524 each permeation event employs only three major channel states, that the currents 525 elicited from the minor states are either too small to be detected or merged into the 526 major states during the concerted movements. Employing more states certainly is 527 possible, but not suggested because extra parameters can incur large inaccuracies if 528 not supported by the enhanced recordings that yield adequate curvatures in the 529 current-voltage curves to discriminate the intermediate states. However, these 530 analyses do suggest a simplified but universal mechanism, sufficient to explain most 531 ion-permeation events with only three major channel states. The essential feature of 532 this mechanism is the voltage-dependent switch between the two ion-bound states ES 533 and FT, reflecting channel's altered abilities to conduct ions in the uni- or opposite

directions. In fact, their presumable equivalents are already found in the structural studies. Cuello et al. have reported a series of the crystal structures of the KcsA K⁺ channel (Cuello et al., 2010), differing in the opening scale at the intracellular gate, some even accompanied by the structural change in the selectivity filter, that any of these structures may represent the ES or FT state with the altered abilities to conduct ions.

540

541 Appendices

542 Appendix A: Parameters of the three-state model

543 The four mechanisms in group A employ the concentration ratios [ES]/[E] and 544 [FT]/[E]. Following similar analyses using the thermodynamic principles (as those used in deriving Eq. (4)), we obtain: $[ES]/[E] = e^{-q_1(V-E_1)/k_BT}$ and $[FT]/[E] = e^{-q_2(V-E_2)/k_BT}$. 545 Here $q_1 = q_{ES} - q_E - q_S$ for mechanisms m3A1 and m3A2, and $q_1 = q_{ES} - q_E - q_P$ for 546 mechanisms m3A3 and m3A4. $q_2 = q_{FT} - q_E - q_T$ for mechanisms m3A1 and m3A3, 547 548 and $q_2 = q_{\rm FT} - q_{\rm E} - q_{\rm Q}$ for mechanisms m3A2 and m3A4. E_1 and E_2 are the half-549 activation potentials associated with the portion of the channels changing from the E state to the ES and FT states, respectively. The expressions of Ea, K1, and K2 for the 550 551 individual mechanisms of this group are:

552 m3A1:
$$E_a = -(k_{-2} \cdot [P] + k_{-4} \cdot [Q]) \cdot E_t$$
 $K_1 = -k_2/(k_{-2} \cdot [P] + k_{-4} \cdot [Q])$
 $K_2 = -k_4/(k_{-2} \cdot [P] + k_{-4} \cdot [Q])$

553 m3A2:
$$E_a = (k_3 \cdot [T] - k_{-2} \cdot [P]) \cdot E_t \qquad K_1 = +k_2 / (k_3 \cdot [T] - k_{-2} \cdot [P]) \\ K_2 = -k_{-3} / (k_3 \cdot [T] - k_{-2} \cdot [P])$$

554 m3A3:
$$E_a = -(k_{-4} \cdot [Q] - k_1 \cdot [S]) \cdot E_t$$
 $K_1 = +k_{-1}/(k_{-4} \cdot [Q] - k_1 \cdot [S])$
 $K_2 = -k_4/(k_{-4} \cdot [Q] - k_1 \cdot [S])$

555 m3A4:
$$E_a = (k_1 \cdot [S] + k_3 \cdot [T]) \cdot E_t$$
 $K_1 = -k_{-1}/(k_1 \cdot [S] + k_3 \cdot [T])$
 $K_2 = -k_{-3}/(k_1 \cdot [S] + k_3 \cdot [T])$

556 The four mechanisms in group B employ the concentration ratios [E]/[ES] and [F]/[ES]. Here [E]/[ES] = $e^{-q_1(V-E_1)/k_BT}$ and [F]/[ES] = $e^{-q_2(V-E_2)/k_BT}$, where $q_1 = q_E + q_S - q_S - q_E + q_S - q_S + q_S +$ 557 558 $q_{\rm ES}$ for mechanisms m3B1 and m3B2, and $q_1 = q_{\rm E} + q_{\rm P} - q_{\rm ES}$ for mechanisms m3B3 559 and m3B4. $q_2 = q_F + q_T - q_{ES}$ for mechanisms m3B1 and m3B3, and $q_2 = q_F + q_Q - q_F$ $q_{\rm ES}$ for mechanisms m3B2 and m3B4. E_1 and E_2 are the half-activation potentials 560 561 associated with the portion of the channels changing from the ES state to the E and F 562 states, respectively. The expressions of Ea, K1, and K2 for the individual mechanisms 563 of this group are:

564 m3B1:
$$E_a = (k_2 + k_4) \cdot E_t$$
 $K_1 = -k_{-2} \cdot [P]/(k_2 + k_4)$
 $K_2 = -k_{-4} \cdot [Q]/(k_2 + k_4)$

565 m3B2:
$$E_a = (k_2 - k_{-3}) \cdot E_t$$
 $K_1 = -k_{-2} \cdot [P]/(k_2 - k_{-3})$
 $K_2 = +k_3 \cdot [T]/(k_2 - k_{-3})$

566 m3B3:
$$E_a = -(k_{-1} - k_4) \cdot E_t$$
 $K_1 = -k_1 \cdot [S]/(k_{-1} - k_4)$
 $K_2 = +k_{-4} \cdot [Q]/(k_{-1} - k_4)$

567 m3B4:
$$E_a = -(k_{-1} + k_{-3}) \cdot E_t$$
 $K_1 = -k_1 \cdot [S]/(k_{-1} + k_{-3})$
 $K_2 = -k_3 \cdot [T]/(k_{-1} + k_{-3})$

The four mechanisms in group C employ the concentration ratios [E]/[ES] and [EST]/[ES]. Here [E]/[ES] = $e^{-q_1(V-E_1)/k_BT}$ and [EST]/[ES] = $e^{-q_2(V-E_2)/k_BT}$, where $q_1 = q_E$ $+ q_S - q_{ES}$ for mechanisms m3C1 and m3C2, and $q_1 = q_E + q_P - q_{ES}$ for mechanisms m3C3 and m3C4. $q_2 = q_{EST} - q_{ES} - q_T$ for mechanisms m3C1 and m3C3, and $q_2 = q_{EST}$ $- q_{ES} - q_Q$ for mechanisms m3C2 and m3C4. E_1 and E_2 are the half-activation potentials associated with the portion of the channels changing from the ES state to 574 the E and EST states, respectively. The expressions of Ea, K_1 , and K_2 for the

575 individual mechanisms of this group are:

576 m3C1:
$$E_a = (k_2 - k_{-4} \cdot [Q]) \cdot E_t \qquad K_1 = -k_{-2} \cdot [P]/(k_2 - k_{-4} \cdot [Q]) \\ K_2 = +k_4/(k_2 - k_{-4} \cdot [Q])$$

577 m3C2:
$$E_a = (k_2 + k_3 \cdot [T]) \cdot E_t$$
 $K_1 = -k_{-2} \cdot [P]/(k_2 + k_3 \cdot [T])$
 $K_2 = -k_{-3}/(k_2 + k_3 \cdot [T])$

578 m3C3:
$$E_a = -(k_{-1} + k_{-4} \cdot [Q]) \cdot E_t$$
 $K_1 = -k_1 \cdot [S]/(k_{-1} + k_{-4} \cdot [Q])$
 $K_2 = -k_4/(k_{-1} + k_{-4} \cdot [Q])$

579 m3C4:
$$E_a = -(k_{-1} - k_3 \cdot [T]) \cdot E_t \quad K_1 = -k_1 \cdot [S]/(k_{-1} - k_3 \cdot [T]) K_2 = +k_{-3}/(k_{-1} - k_3 \cdot [T])$$

580

581 Appendix B: Relation of the three-state models

Lacroix et al. developed a three-state model suitable to calculate the multistage gating charge as a function of the test voltage (Lacroix et al., 2012). The gating charge per voltage-sensing domain is defined as:

585
$$Q_{g} = \frac{q_{2}' + q_{1}' \left(1 + e^{-q_{2}'(V-E_{2})/k_{B}T}\right)}{1 + e^{-q_{2}'(V-E_{2})/k_{B}T} \cdot \left(1 + e^{-q_{1}'(V-E_{1})/k_{B}T}\right)} = \frac{q_{1}' + \left(q_{1}' + q_{2}'\right)e^{q_{2}'(V-E_{2})/k_{B}T}}{1 + e^{-q_{1}'(V-E_{1})/k_{B}T} + e^{q_{2}'(V-E_{2})/k_{B}T}}$$
(B1)

In their definitions, q_1' is the charge associated with the transition from the resting to the intermediate state, and q_2' is the charge associated with the subsequent transition from the intermediate to the active state. Now let's define the total gating charge per voltage-sensing domain $Q_{g,max} = q_1' + q_2'$, and the fraction of the charge in the intermediate state $f = q_1'/Q_{g,max}$, then we obtain:

591
$$\frac{Q_g}{Q_{g,\max}} = \frac{f\left(1 + \frac{1}{f}e^{q_2'(V-E_2)/k_BT}\right)}{1 + e^{-q_1'(V-E_1)/k_BT} + e^{q_2'(V-E_2)/k_BT}}$$
(B2)

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593
$$\mathbf{I} = \frac{\left(k_2 - k_{-4} \cdot [\mathbf{Q}]\right) \cdot \mathbf{E}_{t} \cdot \left(1 - \frac{k_{-2} \cdot [\mathbf{P}]}{k_2 - k_{-4} \cdot [\mathbf{Q}]} e^{-q_1(V - E_1)/k_B T} + \frac{k_4}{k_2 - k_{-4} \cdot [\mathbf{Q}]} e^{-q_2(V - E_2)/k_B T}\right)}{1 + e^{-q_1(V - E_1)/k_B T} + e^{-q_2(V - E_2)/k_B T}}$$
(B3)

594

595 If k_{-2} ·[P] is small enough or k_{-2} ·[P] $\leq k_2 - k_{-4}$ ·[Q], the second term in the

numerator of Eq. (B3) can be neglected, and we obtain

597
$$\frac{I}{I_{max}} = \frac{\frac{\left(k_2 - k_{-4} \cdot [Q]\right)}{k_4} \cdot \left(1 + \frac{k_4}{k_2 - k_{-4} \cdot [Q]} e^{-q_2(V - E_2)/k_B T}\right)}{1 + e^{-q_1(V - E_1)/k_B T} + e^{-q_2(V - E_2)/k_B T}}$$
(B4)

Here $I_{max} = k_4$ ·Et. Comparing Eq. (B4) with Eq. (B2), $q_1 = q_1'$ and $q_2 = -q_2'$, because we have defined the transitions both starting from the intermediate state in mechanism m3C1. The fraction of charge *f* in Eq. (B2) is equivalent to the fraction of the rate constant $(k_2 - k_{-4} \cdot [Q])/k_4$ in Eq. (B4). Therefore Eq. (B2) can be considered as a special case of mechanism m3C1 defined by Eq. (B3) when k_{-2} ·[P] ≈ 0 or k_{-2} ·[P] $<< k_2 - k_{-4}$ ·[Q].

604

Appendix C: Relation of the four-state model and the sequential Boltzmann equations

The sequential Boltzmann equations proposed by Bezanilla et al. were also used to calculate the gating charge of the Shaker channel, that yielded a charge-voltage curve identical to that obtained by the three-state model (Bezanilla, Perozo, & Stefani, 1994). The sequential Boltzmann equations can represent the two independent ionbinding processes, that include an intermediate ES state with the faction of charge f = $q_1'/Q_{g,max}$, where $Q_{g,max} = q_1' + q_2'$:

613

$$\frac{Q_g}{Q_{g,\text{max}}} = \frac{f}{1 + e^{-q_1'(V - E_1)/k_B T}} + \frac{1 - f}{1 + e^{-q_2'(V - E_2)/k_B T}}$$

$$= \frac{f\left(1 + \frac{1 - f}{f} \cdot e^{-q_1'(V - E_1)/k_B T} \cdot e^{q_2'(V - E_2)/k_B T} + \frac{1}{f} \cdot e^{q_2'(V - E_2)/k_B T}\right)}{1 + e^{-q_1'(V - E_1)/k_B T} + e^{-q_1'(V - E_1)/k_B T} \cdot e^{q_2'(V - E_2)/k_B T} + e^{q_2'(V - E_2)/k_B T}} \tag{C1}$$

And this is equivalent to a four-state model involving the E, ES, FT, and EST

615 states. Let's write one simple mechanism for this four-state model:

$$E + S \Leftrightarrow ES \xleftarrow{k_{1}} E + P$$

$$E + T \Leftrightarrow FT \xleftarrow{k_{2}} E + Q$$

$$E + S + T \Leftrightarrow EST \xleftarrow{k_{3}} E + P + Q$$
(C2)

617 If we define $I_{max} = k_3 \cdot E_t$, then the normalized current for this mechanism is:

618
$$\frac{I}{I_{max}} = \frac{\frac{k_1}{k_3} \left(1 - \frac{k'}{k_1} e^{-q_1(V-E_1)/k_BT} + \frac{k_2}{k_1} e^{-q_2(V-E_2)/k_BT} + \frac{k_3}{k_1} e^{-q_3(V-E_3)/k_BT} \right)}{1 + e^{-q_1(V-E_1)/k_BT} + e^{-q_2(V-E_2)/k_BT} + e^{-q_3(V-E_3)/k_BT}}$$
(C3)

620 (Eq. (7)) involving four states. Here
$$k' = k_{-1} \cdot [P] + k_{-2} \cdot [Q] + k_{-3} \cdot [P] \cdot [Q]$$
,

621 [E]/[ES] =
$$e^{-q_1(V-E_1)/k_BT}$$
, [FT]/[ES] = $e^{-q_2(V-E_2)/k_BT}$, and [EST]/[ES] = $e^{-q_3(V-E_3)/k_BT}$. When

622 k' is small enough or $k' \ll k_1$, the second term in the numerator can be neglected, and 623 we obtain:

624
$$\frac{I}{I_{\text{max}}} = \frac{\frac{k_1}{k_3} \left(1 + \frac{k_2}{k_1} e^{-q_2(V - E_2)/k_B T} + \frac{k_3}{k_1} e^{-q_3(V - E_3)/k_B T} \right)}{1 + e^{-q_1(V - E_1)/k_B T} + e^{-q_2(V - E_2)/k_B T} + e^{-q_3(V - E_3)/k_B T}}$$
(C4)

625 For the independent binding processes defined by Eq. (C2), 626 $e^{-q_2(V-E_2)/k_BT} = e^{-q_1(V-E_1)/k_BT} \cdot e^{-q_3(V-E_3)/k_BT}$, inserting this into Eq. (C4), we obtain:

627
$$\frac{I}{I_{max}} = \frac{\frac{k_1}{k_3} \left(1 + \frac{k_2}{k_1} e^{-q_1(V-E_1)/k_BT} \cdot e^{-q_3(V-E_3)/k_BT} + \frac{k_3}{k_1} e^{-q_3(V-E_3)/k_BT} \right)}{1 + e^{-q_1(V-E_1)/k_BT} + e^{-q_1(V-E_1)/k_BT} \cdot e^{-q_3(V-E_3)/k_BT} + e^{-q_3(V-E_3)/k_BT}}$$
(C5)

Now comparing Eq. (C5) with Eq. (C1), $q_1 = q_1'$, $q_3 = -q_2'$. The fraction of charge *f* at the intermediate state is equivalent to the rate constant ratio k_1/k_3 , and (1-f)/f is equivalent to $(k_2/k_3)/(k_1/k_3) = k_2/k_1$. So the sequential Boltzmann equations can be described by the four-state model defined by Eq. (C5), which is a special case of Eq. (C3) that occurs when $k' \approx 0$ or $k' \ll k_1$. Therefore the sequential Boltzmann equations can be included in the general kinetic model defined by Eq. (7).

634 Now let's compare Eq. (C1) and Eq. (B2). In Eq. (C1), if the second term $(1 - f) / f \cdot e^{-q_1'(V - E_1)/k_B T} \cdot e^{q_2'(V - E_2)/k_B T}$ is much smaller than the other terms in the 635 numerator, and the third term $e^{-q_1'(V-E_1)/k_BT} \cdot e^{q_2'(V-E_2)/k_BT}$ is much smaller than the other 636 637 terms in the denominator, they can be neglected from the numerator and denominator, and hence Eq. (C1) is reduced to Eq. (B2). This is when the three-state model and the 638 639 sequential Boltzmann equations yield the indistinguishable charge-voltage curves like 640 those found by Bezanilla et al. (Bezanilla, Perozo, & Stefani, 1994). Following the 641 mechanism defined by Eq. (C2), this situation can happen when the FT state is 642 merged into another ion-bound state, so that the four-state model is readily reduced to a three-state model involving only the E, ES, and EST states. And this is why we 643 644 suggest using the lower-rank model whenever the extra intermediate states cannot be 645 differentiated by the current-voltage curves.

646

647 Materials and Methods

648 Source data of the current-voltage curves plotted in Fig. 4 were obtained from the 649 publications listed in Table 1, which were also cited in the Discussion section. For 650 most of them, we directly used the source data as presented in the source data file 651 along with the publications, except for the following data. The source data of the current-voltage curves in Figure 4A of Ref. (Lemoine et al., 2020) each contained 400 pair of values. We had reduced each data size to 41 and used them for the curve fitting procedure. The data of the reduced size are presented in the supplementary table.

The source data of Figure 2b in Ref. (Zhou et al., 2017) contained 395 pair of values, and we had reduced the data size to 41. The data with the reduced size are presented in the supplementary table.

All source data were fitted following the method of the nonlinear least squares. The model parameters that yielded the smallest σ value were selected, where σ is defined as:

662
$$\sigma = \sqrt{\frac{\sum_{i=1}^{N} (m_i - s_i)^2}{N \cdot |s_{\max}|^2}}$$

Here s_i represents the *i*-th current value in the source data of size *N*. m_i represents the *i*-th current value calculated by the selected three-state model. $|s_{max}|$ is the absolute value of the maximum current in the source data file. The parameters selected for each curve, together with the calculated σ value, are presented in Table 1.

667

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675 Competing Interests

676 The author declares no competing financial interests.

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