1	Accumulation of ammonium owing to the metabolic imbalance of carbon
2	and nitrogen might inhibit the central metabolism in Methylomonas sp.
3	ZR1
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12 Abstract:

The metabolic intermediates of nitrogen source have been proved to have 13 multiple functions on the metabolism of mehthanotrophs. In this study, 14 accumulation and assimilation mechanism of the nitrate metabolic 15 intermediate ammonium in the fast growing Methylomonas sp. ZR1 was 16 analyzed. Although, nitrate salt was the best nitrogen source supporting 17 the growth of ZR1, its metabolic intermediate ammonium would 18 accumulate and inhibit ZR1. Kinetic studies indicated that accumulation 19 of NH4⁺ was deduced from the imbalance of nitrogen and carbon 20 metabolism. Compensation of carbon skeleton α-keto-glutaramate could 21 effectively relieve the inhibition of NH4⁺ to ZR1, which further approved 22

the assumption. qPCR analysis indicated a third ammonium assimilation 23 pathway Glycine synthesis system may function in ZR1 under high 24 ammonium tension. In the presence of ammonium, ZR1 might employ 25 two strategies to relieve the ammonium stress, one was assimilating the 26 excess ammonium, and another one was cutting off the nitrogen reduction 27 reactions. Investigation of the nitrogen metabolism and its influence to 28 the carbon metabolism is meaningful to systematically understand and 29 control the C1 feedstock bioconversion process in methanotrophs. 30

31 **Importance:**

The nitrogen metabolism in methanotrophs has long been concerned. 32 However, there are lots of research problems yet to be solved. In this 33 study, the accumulation and assimilation mechanism of the nitrogen 34 metabolic intermediate ammonium in the fast growing *Methylomonas* sp. 35 ZR1 was analyzed. Owing to the imbalance metabolism of carbon and 36 nitrogen source, ammonium would accumulate to high concentrations to 37 inhibit cell growth. Compensation of carbon skeleton was an effective 38 strategy to relieve the inhibition of NH₄⁺. A third ammonium 39 assimilation pathway related genes were proved actively expressing in 40 ZR1 when it confronted with high ammonium tension. When confronted 41 with ammonium tension, ZR1 might employ different strategies to 42 relieve the ammonium stress according to the edible carbon source. 43 Revealing the endogenous ammonium accumulation mechanism and its 44

45 metabolic adjustment effect on the central metabolism of methanotrophs,

- 46 was meaningful to reveal the complex coordination metabolic
 47 mechanism of nitrogen and carbon in methanotrophs.
- 48 Keywords: nitrogen metabolism, ammonium, *Methylomonas*
- 49 **Introduction:**

Methanotrophs are excellent C1 compounds assimilating microorganisms, 50 and its industrial application has a long history since 1960s. At the 51 beginning methanotrophs were used to produce single cell protein from 52 methane or methanol. Nowadays, with the development of biotechnology, 53 methanotrophs were genetic modified to produce lactic acid (Henard et 54 al., 2016; Garg et al., 2018), astaxanthin (Ye and Kelly, 2012) and 55 α -hemelune (Sonntag et al., 2015) from methane. However, the slow 56 growth nature of methanotrophs has constrained its application in 57 industrial biotechnology. 58

Efforts to optimize fermentation technologies to improve the growth rates 59 of methanotrophs on methane have been reported over the past years (Yu 60 et al., 2003; Park et al., 1991b; Shah et al., 1996; Xing et al., 2006; Han et 61 al., 2009) (Myung et al., 2016). Other factors such as the addition of 62 citrate acid (Xing et al., 2006) and adjusting the ion concentration (Leak 63 and Dalton, 1986) were also reported to have effect in improving the 64 growth rate of methanotrophs from methane. These studies indicated that 65 the growth rate of methanotrophs might be result from multi-factor 66

combined effects. Many studies have also proved that nitrogen sources 67 have strong effect on the growth of methanotrophs. Park et al. studied 68 several factors including the nitrogen source affected the growth of OB3b, 69 and found that nitrate depletion was responsible for the diauxic growth 70 pattern in the batch cultivation of OB3b in the bioreactor. However, its 71 growth declined much with 40 mM nitrate (Park et al., 1991). Hoefman et 72 al. found niche partitioning among methanotrophic species, with methane 73 oxidation activity responses to changes in nitrogen content being 74 dependent on the in situ methanotrophic community structure (Hoefman 75 et al., 2014). Strains have developed a complex mechanism to balance the 76 carbon and nitrogen metabolism (Commichau et al., 2006). Recent 77 studies have shown that many rapidly proliferating cells are dependent on 78 the nitrogen metabolic intermediate such as serine and glutamine (Yang 79 and Vousden, 2016). These studies indicated that nitrogen and its 80 metabolic intermediate might have multiple functions on the central 81 metabolism of methanotrophs, and subsequently affected its growth. 82

In methanotrophs, the nitrogen metabolism has long been concerned. 83 1983, Dalton et al. studied ammonia assimilation in the type X 84 methanotroph Methylococcus capsulatus Bath, type I methanotroph, 85 *methanica* S1, and the Π *Methylomonas* type methanotroph, 86 Methylosinus trichosporium OB3b, and found that, Bath and S1 possess 87 both the glutamine synthase/ glutamine 2-oxo-glutarate amino transferase 88

(GS-GAGOT) and alanine dehydrogenase (ALAD) pathways for the 89 assimilation of ammonia, but operated according to the nitrogen source 90 (Murrellt and Dalton, 1983). Loginova NV et al. (Loginova et al., 1982) 91 studied enzymes involved in ammonium assimilation by 15 bacterial 92 strains of different taxonomy, and found that bacteria were found to differ 93 in the enzymes for ammonium assimilation according to the pathways of 94 C1-metabolism. Nyerges (Nyerges, 2008) assessed primary the 95 differences in ammonia co-metabolism among four methanotrophs 96 isolates, found the investigated strains exhibited different levels of 97 ammonia and hydroxylamine oxidation, and inhibition of 98 methane-oxidizing activity by ammonia and nitrite. Dam et al. found that 99 ammonium induces differential expression pattern of methane and 100 nitrogen metabolism-related gene in Methylocystis sp. strain SC2 (Dam et 101 al., 2014). Kits et al. found that in Methylomonas denitrifican, sp. nov. 102 type strain FJG1, methane oxidation could couple to nitrate reduction 103 under hypoxia (Kits et al., 2015). These results revealed that nitrogen 104 metabolism might play an important role in the global metabolism of 105 methanotrophs, and nitrogen metabolism mechanism might be distinct 106 among methanotrophs species. 107

In this study, the nitrogen metabolism during the cell growth of *Methylomonas* sp. ZR1 were studied at kinetic and gene expression level.
It was found that, ZR1 might employ different nitrogen metabolic pattern

according to the ediable carbon source, the nitrogen metabolic 111 intermediate ammonium which was also identified as a growth inhibiter 112 was found to accumulate when nitrogen sources were relative surplus in 113 comparison with carbon sources. Carbon skeleton supplementation was 114 found to be an efficient strategy to relieve the inhibition effect of 115 ammonium on the growth of ZR1. qPCR analysis of the carbon and 116 nitrogen metabolic key gene indicated the gene expression diversity when 117 ZR1 was under the ammonium tension condition, and a third ammonia 118 assimilation pathway were found highly expressed with methane as 119 carbon source. All these results further indicated that ammonium might 120 have multidimensional effect on the central metabolism of ZR1. 121

122 Material and Methods

123 Strains and culture method

Methylomonas sp. ZR1 was isolated by our group and deposited in China 124 General Microbiological Culture Collection Center with the accession 125 number CGMCC No. 9873. It can be cultivated using liquid or solid 126 mineral medium with methane or methanol as the growth substrate. The 127 medium used for ZR1 cultivation was nitrate mineral salts (NMS) 128 medium (Whittenbury et al., 1970). For nitrogen source screening, NMS 129 medium without nitrate salt was added with 1 g/l KNO₃, 1 g/l NaNO₃, 0.5 130 g/l NH₄Cl, 0.5 g/l (NH₄)SO₄, 0.5 g/l urea, 1 g/l trypton or 1 g/l yeast 131 extract separately. Samples having no nitrogen substate were performed 132

as control. To test the ammonium inhibition effect, ammonium chloride at 133 different concentration were added into the NMS medium with or without 134 1 g/l KNO₃. And for carbon skeleton compensation test, 0.1 g/l 135 ammonium chloride with 0.3 g/l α -ketoglutaramate (α -KG), 0.4 g/l 136 glutamate, 0.3g/l malic acid, or 0.3g/l pyruvate were simultaneously 137 added into the NMS medium. ZR1 was cultured in flask or bubble 138 column reactor according to the method described by Guo et al. (Guo et 139 al., 2017). When using methanol as carbon source, the initial methanol 140 concentration was 6 g/l without specific instruction. When using methane 141 as carbon source, ZR1 was cultured in bubble column reactor for the 142 ammonium accumulation study; and cultured in flask with gas refreshing 143 every 12 hours for the nitrogen source screening, ammonium inhibition 144 test and carbon skeleton replenishing study. 145

146 Total nitrogen and ammonium concentration analysis method

Total nitrogen concentration in the fermentation broth was analyzed using
the TOC/TN analyzer Multi N/C 2100s (Analytik Jena AG, German).
Fermentation broth was first centrifuged, and the supernatant were diluted
using ddH₂O into suitable concentration for the analysis.

Ammonium was analyzed using the indophenol blue reaction according to the method described by Xie et al. (Xie et al., 2005). The fermentation broth was first centrifuged, and 0.1 ml of the supernatant were added with 0.5ml reaction solution 1 (3.5 g phenol and 0.04 g sodium nitroprusside in 100 ml ddH₂O), and 0.5 ml reaction solution 2 (1.8 g sodium sodium hydroxide and 4.0 mmol sodium hypochlorite in 100 ml ddH₂O). The mixture was maintained at 37° C for 1 hour, and the absorbance of the solutions was read on a spectrophotometer at 625 nm. The concentration of the ammonium in the samples was calculated according to the calibration curve established using ammonium chloride.

161 Methanol concentration

Methanol concentration in the fermentation broth were measured by a GC (GC9790, Fuli Instrument, China) equipped with a flame ionization detector (FID) and a capillary column ($0.25\mu m$, $60m \times 0.25mm$, 7KG-G013-11 ZebronTM, Phenomenex).

166 Kinetic analysis of the carbon and nitrogen metabolism to cell growth

The growth, carbon and nitrogen assimilation curve were first fitted using the logistic model in Origin 9.0. Then the simulated curve were took derivative with respect to OD_{600} of ZR1 using the mathematic module of origin 9.0.

171 qPCR analysis of expression level of the genes concerns carbon and

172 **ammonium metabolism**

Strain ZR1 was initially grown up to log phase (OD₆₀₀ achieved 0.8) in 50 ml of NMS medium in a 250 ml bottle fitted with butyl rubber septum and with 20% methane (v/v) in the headspace. For methanol as carbon sources, strains were cultured with 6 g/l methanol in 50 ml NMS medium

in 250 ml flask. For preparing ammonium inhibition samples, 0.1 g/l of 177 NH₄Cl were then added into the medium. For carbon compensation 178 samples, 0.1 g/l of NH₄Cl and 0.3 g/l of α -keto-glutaramate $(\alpha - KG)$ 179 were added, and samples without any operation were performed as 180 control. All samples were incubated for 2 hours, then, cells were collected 181 by centrifugation and washed with TE buffer. RNA was extracted 182 immediately using the RNAprep Pure Cell/Bacteria Kit (Tiangen Biotech 183 (Beijing) CO., Ltd), according to the product instructions. The extracted 184 RNA was used as template to construct cDNA using the Takara Prime 185 Script RTreagent Kit with gDNA Eraser (Perfect Real Time). Then genes 186 were amplified using the SYBR Premix EX Tag kit, using Applied 187 Biosystems 7500 fast Real-Time PCR system. Standard housekeeping 188 gene 16S rDNA was selected as internal control gene. qPCR data 189 obtained were analyzed using the method described by Pfaffl (Pfaffl, 190 2001). 191

192 Primers used are listed in table 1.

Gene	Forward primer	Reverse primer	Product
			length bp
16s rDNA	ATGCAAGTCGAACGGTAGCA	AGGGCGTATGCGGTATTAGC	125
glnA	TGGCGTTTTCTGCCATGTTG	ACGAGTCAAGAATTCGCGGT	185
gltB	CGCTCTGGTTAAATCGCACG	CAACGGGATGTTGCGTTTGT	143

alaD	CAACAAGCATTGCGGGACAA	TCAGCTCAGAAACTGCTCCG	129
gcvT	GCAGTTTCTCGCCGAGTCTA	GTAACGGTGCGGAACGAAAG	149
nir	ACCTGGATAGATGTGTGCGG	CCGCCTATGTCGCCTATCAT	185
nif	TTGCGACCCTAAAGCCGATT	TTTAACGTCGCGGTAACCCA	142
fae	GCACTATCCCTGCTGACGAA	GCGATAGCTTCTTTGGTGGC	124
hps	CACCGGTTTGGATGCACAAG	CGCCATAGATAGCAGCACCA	176
pdh	AACAGCAAATGCGTTGTCCC	CGGCAAATCACCGCCTTTAG	146
CS	ATGACCAGGACCGCGTTATC	GGGTAGCCCCGATACAACAG	193
194	All culture conditions were performed	l in triplicate biological replicates,	

and for qPCR analysis, each biological sample was carried out in triplicate. Raw data of the qPCR result were shown in supplementary Table S1.

198 **Result and discussion**

199 1. Nitrogen source effect for the growth of ZR1 from methane and200 methanol.

201 methanol methane 202 6 g[®] 4 203 204 0 205 СК KNO, NaNO₂ (NH_)2SO4 CO(NH2)2 NH,CI Trypton Yeast extract Nitrogen source

Figure. 1 Growth of ZR1 in the presence of different nitrogen substrate

207 from methane and methanol

Seven categories of nitrogen substrate were tested for the growth of ZR1 208 from both methane and methanol (Figure 1). According to Figure 1, 209 nitrate salts were the best nitrogen sources supporting the growth of ZR1 210 from both methanol and methane among the tested compounds. And the 211 growth of ZR1 from both methanol and methane with ammonium salts 212 was much weaker than from nitrate salts. Meanwhile, the effects of the 213 nitrogen sources on the growth of ZR1 from methane and methanol were 214 somewhat different. With methane as carbon sources the cell density 215 $(OD6_{00})$ of ZR1 achieved 2.5 with potassium nitrate, 1.8 with sodium 216 nitrate, 1.8 with urea, 1.5 with yeast extract, 0.5 with tryptone, and 0.34 217 with ammonia chloride, 0.35 from ammonia sulfate. With methanol as 218 carbon sources the cell density of ZR1 achieved 8.2 from sodium nitrate, 219 7.9 from potassium nitrate, 3.9 from yeast extract, 1.8 from urea, 0.4 from 220 tryptone, 0.35 from ammonia chloride, and 0.33 from ammonia sulfate. It 221 can be seen that nitrate salt was the most suitable nitrogen source and 222 ammonia salt could not effectively support the growth of ZR1 from 223 methane and methanol. It was generally regarded that ammonium radicals 224 were the competitive inhibitor of particular methane monooxygenase 225 (pMMO) (He et al., 2017; Hu and Lu, 2015; Dam et al., 2014; Nyerges et 226 al., 2010; Nyerges and Stein, 2009; Nyerges, 2008; Dunfield and 227 Knowles, 1995; Schnell and King, 1994; Carlsen et al., 1991; Murrellt 228

and Dalton, 1983), which will resulted in the lower growth of strains from
methane. However, this study also proved that ammonium would hinder
the growth of ZR1 from methanol, although its supposed competent
object pMMO is not the key enzyme for the metabolism of methanol. It
means that ammonium might have other inhibition effects for the growth
of ZR1, besides its competitive inhibition effect to pMMO.

235 2. Effect of nitrate concentration on the growth of ZR1 from methane andmethanol

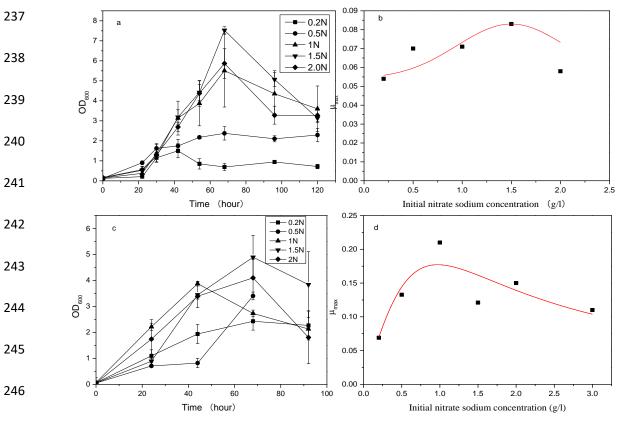
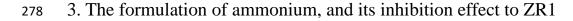


Figure 2. Effect of nitrate concentration on the growth of ZR1 from methane and methanol a, growth of ZR1 with different initial nitrate sodium concentration from methanol; b, specific growth rate of ZR1 with different initial nitrate concentration from methanol; c, growth of ZR1

with different initial nitrate sodium concentration from methane; d,
specific growth rate of ZR1 with different initial nitrate concentration
from methanol;

According to Figure 1, nitrate salts were supposed to be the most suitable 254 nitrate source that supporting the growth of ZR1 from both methane and 255 methanol, and effects of potassium nitrate concentration on the growth of 256 ZR1 were investigated subsequently. According to Fig. 2, the obvious 257 substrate inhibition effects of nitrate salts on the growth of ZR1 were 258 investigated. The highest cell density of ZR1 achieved 7.8 with 1.5 g/l 259 potassium nitrate (Figure 2a). Meanwhile, Figure 2b indicated that the 260 growth rate of ZR1 is first ascend with the increase in the substrate level 261 and approaches a maximum value at 1.5 g/l of potassium nitrate. Then, a 262 subsequent increase in the nitrogen concentration leaded to a decrease in 263 the specific growth rate of ZR1. The situation with methane as carbon 264 sources is similar to that of methanol. The highest cell density of ZR1 265 achieved 4.8 with 1.5 g/l potassium nitrate (Figure 2c), however the 266 specific growth rate of ZR1 achieved 0.23 with 1 g/l of potassium nitrate 267 (Figure 2d). It has been reported that growth of type II methanotrophs 268 OB3b were also inhibited by 40 mM nitrate (4 g/l)(Park et al., 1991a). 269 These phenomenon indicated that nitrate as nitrogen source when its 270 concentration achieved a certain value might inhibit the growth of 271 methanotrophs. According to the substrate inhibition theory (Muchandani 272

and Luong, 1989), an increase in the substrate concentration could cause an alteration in the cell metabolism such as an overproduction of a molecule by one pathway which results in the feedback inhibition of a second related pathway. Thus the nitrate metabolism pathway of ZR1 was further explored.



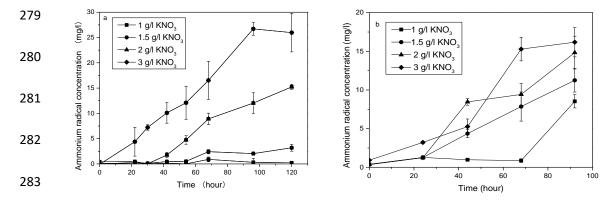


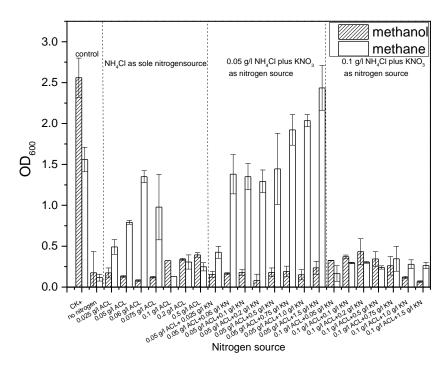
Figure.3 Ammonium accumulation of ZR1 in the fermentation broth with

different initiate nitrate concentration from methanol and methane

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Considering the nitrate metabolism pathway of methanotrophs, nitrogen 286 metabolism intermediate NH₄⁺ was supposed to be accumulated during 287 the process. Thus ammonium concentration in the fermentation broth was 288 analyzed during the fermentation process of ZR1 from methane and 289 methanol. According to Figure 3, the concentration of NH_4^+ in the 290 fermentation broth grew higher and higher during the fermentation 291 process, which indicated that NH₄⁺ might accumulated accompanied with 292 the growth of ZR1 when excess nitrate were supplied. With the increasing 293 concentration of nitrogen source, the starting time of NH₄⁺ accumulation 294

has been moved up. Meanwhile the final concentration of NH₄⁺ increased 295 with the increase of the initial concentration of KNO₃. With 3 g/l of 296 KNO₃, the final accumulated NH₄⁺ achieved 25 mg/l with methanol as 297 carbon sources and 16 mg/l with methane as carbon source. According to 298 Fig. 1, NH_4^+ was supposed to be an inhibitor of the growth of ZR1. The 299 accumulated NH4⁺ during the growth process of ZR1 might inhibit the 300 growth of ZR1. Thus the effect of NH₄⁺ on the growth of ZR1 was further 301 analyzed. 302



303

Figure. 4 Effect of Ammonium on the growth of ZR1 from methane and

- 305 methanol.
- 306 ACL, NH₄Cl; KN, KNO₃;

307 Growths of ZR1 with NH₄Cl as sole nitrogen source or with both NH₄Cl

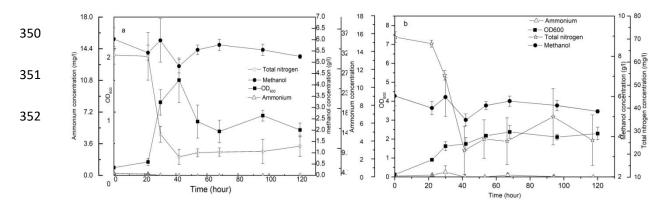
and KNO₃ at different concentrations were investigated. It was found that,

when NH₄Cl were used as the mono-nitrogen sources, cell growth of ZR1 309 first rise up with the going up of the NH₄Cl concentration, the highest cell 310 growth achieved 1.5 when NH₄Cl achieved 0.05, and then growth of ZR1 311 from methane decreased when the initial NH₄Cl concentration growing 312 higher, and OD_{600} of ZR1 with NH₄Cl at 0.1-0.5 g/l only achieved 0.42, 313 much lower than the control (with 1 g/l KNO₃ as nitrogen source). When 314 1 g/l of KNO₃ with addition of 0.05 g/l of NH₄Cl was used as nitrogen 315 sources, the highest cell density of ZR1 from methane achieved 2.5, much 316 higher than control. While the highest cell density of ZR1 from methanol 317 only achieved 0.42 under the same condition. When 1 g/l of KNO₃ with 318 addition of 0.1 g/l of NH₄Cl (NH₄⁺ 34 mg/l, 1.86 mmol/l) was used as 319 nitrogen sources, the highest cell density of ZR1 only achieved 0.7 with 320 methane and 0.8 with methanol. These results showed that, with the 321 existence of KNO₃, growth of ZR1 from methane and methanol was 322 inhibited by NH₄Cl at concentration higher than 0.1 g/l. Growth of ZR1 323 from methane could resist with 0.05 g/l of NH₄Cl (0. 93 mmol/l), and 324 were strictly inhibited with 0.1 g/l of NH₄Cl (1.86 mmol/l), while growth 325 of ZR1 from methanol were totally inhibited by 0.05 g/l of NH₄Cl, which 326 indicated that growth of ZR1 from methanol is more sensitive to NH₄⁺ 327 than that from methane. And according to Figure 3 the final accumulated 328 NH₄⁺ with initial higher concentration of nitrate salts could achieve 25 329 mg/l (0.89 mmol/l), which was high enough to inhibit the growth of ZR1 330

from methanol.

It was generally regarded that ammonium radicals were the competitive 332 inhibitor of pMMO (He et al., 2017). In this study, ammonium was also 333 identified to inhibit the growth of ZR1 from methanol and accumulate 334 during the growing process. Recently, ammonium was reported to have 335 effects on global gene expression of methane and nitrogen 336 metabolism-related gene in methanotrophs (Dam et al., 2014). Being the 337 nitrogen metabolic intermediate, accumulation of ammonium may have 338 multiple effects on the growth of methanotrophs. Revealing its 339 accumulation and assimilation mechanism is meaningful to understand 340 the metabolism mechanism of methanotrophs. 341

4. Carbon and nitrogen metabolic kinetic analysis of ZR1 from methanol 342 According to Figure 4, concentration of the accumulated NH_{4^+} was 343 proportional to the initial nitrogen concentration in the fermentation broth. 344 And the accumulation of NH4⁺ might be result from the high initial 345 nitrogen source concentration (in another words the lacking of the carbon 346 skeleton). Thus, co-metabolism of the carbon and nitrogen and 347 accumulation of ammonium in ZR1 with methanol as carbon source was 348 further analyzed. 349



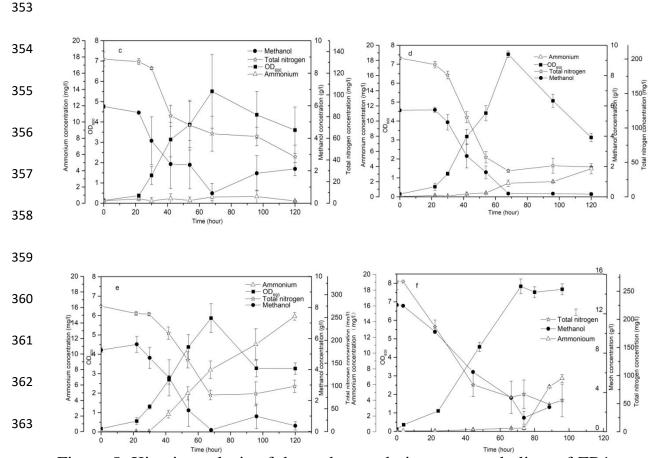


Figure 5. Kinetic analysis of the carbon and nitrogen metabolism of ZR1 from methanol. a, 6 g/l methanol and 0.2 g/l nitrate sodium; b.6g/l methanol and 0.5g/l nitrate sodium; c. 6 g/l methanol and 1 g/l nitrate sodium; d. 6 g/l methanol and 1.5 g/l nitrate sodium; e. 6 g/l methanol and 2 g/l nitrate sodium; f. 12g/l methanol and 2g/l nitrate sodium

According to Figure 5, it can be found that with methanol under the same 369 concentration of 6 g/l, ammonium accumulation increased with the 370 increscent of nitrogen concentration. Nonetheless, increasing the carbon 371 12 g/l (Figure 5e), the accumulated ammonium will sources to 372 subsequently decrease. These results indicated that ammonium 373 accumulation was deduced from the imbalance of the carbon and nitrogen 374

metabolism. And according to the nitrogen test result (Figure 1, Figure 3), the accumulated ammonium could achieve 25-30 g/l which was high enough to inhibit the growth of ZR1 from methane and methanol as carbon sources.

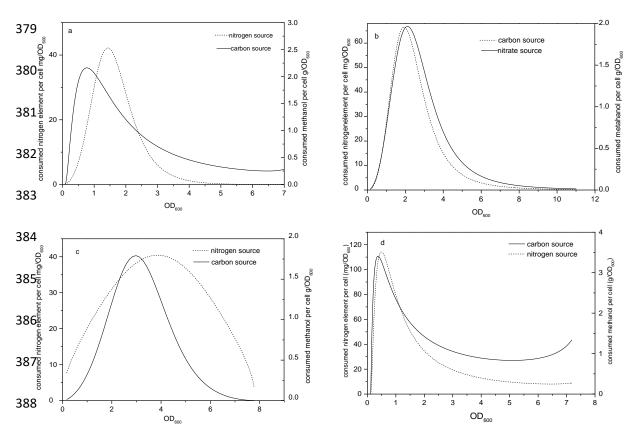


Figure 6. Kinetic analysis of the carbon and nitrogen metabolism to cell growth a. 6 g/l methanol and 1g/l nitrate sodium; b.6g/l methanol and 1.5g/l nitrate sodium; c. 6 g/l methanol and 2 g/l nitrate sodium; d. 12g/l methanol and 2g/l nitrate sodium

To further compare the carbon and nitrogen metabolism pattern, specific consuming rate of the carbon and nitrogen during the growth process of ZR1 were further analyzed. According to Figure 6, it can be seen that the specific uptake rates of nitrogen and carbon dynamically changed with the nitrogen-carbon ratio (NCR). High NCR will result in delayed carbon
consuming, and the carbon and nitrogen uptake will keep in step when
NCR fall into a suitable value.

5. Relieving of ammonium inhibition effect by the carbon metabolites

According to the nitrogen sources metabolic pathway (Murrellt and 401 Dalton, 1983; He et al., 2017; Nyerges, 2008a), methanotrophs mainly 402 assimilate ammonium through the glutamine synthase/ glutamine 403 2-oxo-glutarate amino transferase (GS-GAGOT) and alanine 404 dehydrogenase (ALAD) pathways. To further confirm that accumulation 405 of NH4⁺ was derived from the shortness of the carbon skeleton, several 406 carbon metabolic mediates pyruvate, malic acid, citrate acid, a -KG 407 which are related with the nitrogen metabolism were added to the 408 ammonium accumulation samples (Figure. 7). According to Figure 7, α 409 -KG and glutamate were the most effective carbon metabolic mediates in 410 relieving of the NH₄⁺ inhibition effect with methane as carbon sources. 411 However, with methanol as carbon source only α -KG could effectively 412 relieve the inhibition effect of NH₄⁺ on the growth of ZR1. Carbon 413 metabolic mediates upstream or downstream of α -KG, has small 414 relieving effect on NH₄⁺ inhibition. It was also found that, direct supply of 415 carbon skeleton such as pyruvate, citric acid, α -KG, glutamate or malic 416 acid have somewhat inhibit effect on the growth of ZR1 from methanol. 417 Meanwhile, a -KG and glutamate were found to have stimulate effect on 418

the growth of ZR1 from methane.

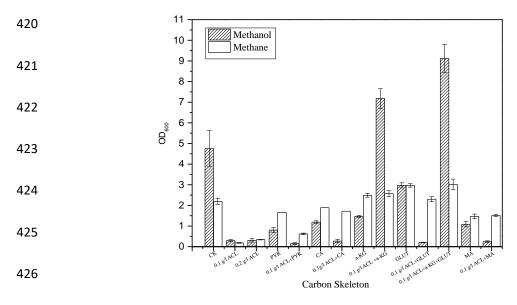


Figure 7. Relieve Effect of the carbon skeleton to ammonium inhibition

428 CK: control; ACL, NH₄Cl; PYR, pyruvate; CA, citrate acid; α -KG, α

-ketoglutaramate ; GLUT, glutamate; MA, malic acid;

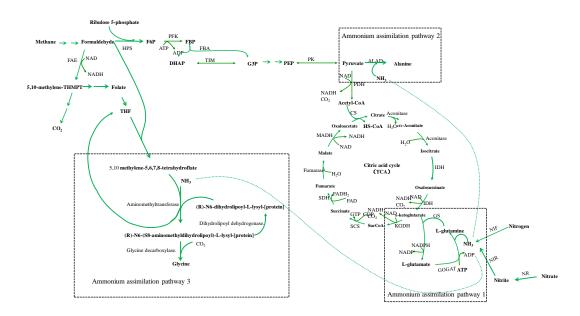
It means that, ZR1 may utilize different carbon and nitrogen metabolic mechanism to balance the carbon and nitrogen metabolism. Nonetheless, α -KG play more important role in assimilation of the nitrogen metabolic intermediate ammonium.

6. Transcript level analysis the carbon and nitrogen metabolism in ZR1with the accumulation of ammonium

According to the nitrogen metabolism pathway of microorganisms, many studies indicated that, ammonia is a competitive inhibitor of pMMO, which deduced the inhibition effect of methanotrophs from methane. However, this study also identified the inhibition effect of NH⁺₄ on ZR1 with methanol as carbon source. So, besides the competitive inhibition effect of MMO, NH₄⁺ as a nitrogen metabolism intermediate may have
multiple complex effects on the metabolic pathway of methanotrophs.
Thus transcript level analysis of the genes relevant with carbon and
nitrogen metabolism in ZR1 was carried out by qPCR.

Ammonium mainly transferred to the C skeletons of pyruvate or 445 intermediates of citric acid cycle (glycogenic) and others to form amino 446 acid. Most amino acid participate in transamination reactions with TCA 447 cycle mediate such as oxaloacetate, or α -ketoglutarate to form aspartate, 448 or glutamate, respectively, and the α -KG corresponding to the original 449 ammonium assimilation. Based on the genomic analysis result of ZR1, 450 besides the ALAD and GS/GOGAT pathway, another special ammonium 451 assimilation mechanism the Glycine synthase system may exist and 452 function in methanotrphs(Figure 8). In this study, the GS-GAGOT, 453 ALAD, and Glycine synthesis system on behalf of three main pathways 454 of the ammonium assimilation were studied by qPCRT, the data quality of 455 qPCR result was presented in supplementary the Table 1. 456 E-supplementary data of this work can be found in online version of the 457 paper. The relative expression fold of the target genes was listed in Table 458 2. 459

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460 461

Figure 8. Nitrogen metabolism pathway of ZR1

HPS:3-hexulose-6-phosphate synthase, DHAP, Dihydroxyacetone phosphate; F6P, 462 Fructose-6-phosphate; FBP, Fructose-1, 6 –bisphosphate; G3P, Glyceraldehyde 463 Phosphoenolopyruvate; 3-phosphate; PEP, SucCoA, Succinyl-CoA; TIM. 464 465 Triosephosphate isomerase; GAPD, Glyceraldehyde 3-phosphate dehydrogenase; PK, pyruvate kinase; PFK, Phosphofructose kinase; CS, Citrate synthase; IDH, Isocitrate 466 dehydrogenase; KGDH, Ketoglutarate dehydrogenase; SCS, Succinyl-CoA synthetase; 467 SDH, Succinate dehydrogenase; MDH, Malate dehydrogenase; GS, Glutamine 468 synthetase; GOGAT, Glutamate synthase; cd1 NIRs, Cytochrome cd1 nitrite reductase; 469 ALAD, Alanine dehydrogenase; NIF, Nitrogenase; FAE, Formaldehyde activating enzyme; 470 Nitrogen assimilation pathway 1: 471 (1)NH₄⁺ + ATP + L - glutamate $\xrightarrow{\text{GS}}$ L - glutamine + ADP + Pi + H⁺ 472 (2)NADH + α KG + L - glutamine + H⁺ $\xrightarrow{\text{GOGAT}}$ 2 × L - glutamate + NAD 473 (1)+(2) $NH_4^+ + NADH + \alpha KG + ATP \xrightarrow{GS-GOGAT} L - glutamate + ADP +$ 474 $Pi + NAD^+$

475 Nitrogen assimilation pathway 2:

	ALD
476	$NH_4^+ + pruvate + NADH \xrightarrow{ALD} L - alanine + H_2O + NAD^+$

477 Nitrogen assimilation pathway 3:

- $\begin{array}{l} (1) \ (6R) 5,10 methylene 5,6,7,8 tetrahydrofolate + (R) N^6 dihydrolipoyl L lysyl [protein] + NH_4^+ \longleftrightarrow \\ (6s) 5,6,7,8 tetrahydrofolate + (R) N^6 (S^8 aminomethyldihydrolipoly) L lysyl [protein] \end{array}$
- (2) (R) $N^6 (S^8 aminomethyldihydrolipoyl) L lysyl [glycine cleavage complex H protein]$ + CO₂ \leftrightarrow glycine + H⁺ + N⁶ - lipolyl - L - lysyl - [glycine - cleavage complex H protein]
- (3) (R) $N^6 lipoyl L lysyl [protein] + H^+ + NADH \leftrightarrow (R) N^6 dihydrolipoyl L lysyl [potein] + NAD^+$

$$(1) + (2) + (3) (6R) - 5,10 - methylene - 5,6,7,8 - tetrahydrofolate + NADH + NH_4^+ + CO_2$$
$$\longleftrightarrow glycine + NAD^+ + (6s) - 5,6,7,8 - tetrahydrofolate$$

478

479 Direction 1 is catalyzed by two enzymes, glutamine synthase and

480 glutamine 2-oxo-glutarate amino transferase. Direction 2 is catalyzed by

481 the alanine dehydrogenase. Direction 3 is catalyzed by the Glycine

482 synthesis system.

Table 2. Relative expression level of the genes concerning nitrogen and

484

carbon metabolism

		Mean fold change in gene expression of samples						
genes	protein	N_{control}	NACL	N _{NCL-KG}	$M_{control}$ - $N_{control}$	M_{control}	M_{ACL}	M _{ACL-KG}
glnA	Glutamine synthase	1	1.92	15.73	0.69	1.00	0.89	1.34
gltB	Glutamate synthase large subunit	1	7.48	12.64	42.38	1.00	0.45	0.12
alaD	Alanine dehydroge nase	1	30.84	17.84	25.07	1.00	0.48	0.08
gcvT	Aminome thyltransfe rase	1	19.08	2.96	1.56	1.00	0.95	2.17
nir	Nitrite reductase	1	8.74	1.93	2.01	1.00	0.48	0.92

nif	Nitrogena	1	4.59	4.14	2.15	1.00	0.50	0.92
	se							
fae	Formalde	1	4.32	2.94	2.99	1.00	1.07	1.04
	hyde							
	activating							
	enzyme							
hps	3-hexulos	1	13.30	2.53	5.70	1.00	1.42	0.75
	e-6-phosp							
	hate							
	synthase							
pdh	Pyruvate	1	1.66	1.35	0.05	1.00	1.88	2.06
	Dehydrog							
	enase							
CS	Citrate	1	1.15	1.52	8.32	1.00	1.24	1.21
	synthase							

⁴⁸⁵ N_{control}, Samples cultured with methane; N_{ACL}, Samples cultured with ⁴⁸⁶ methane under 0.1 g/l NH₄Cl condition; N_{ACL-KG}, Samples cultured with ⁴⁸⁷ methane under 0.1 g/l NH₄Cl and 0.3 g/l α -KG; M_{control}, Samples ⁴⁸⁸ cultured with methanol; M_{ACL}, Samples cultured with methanol under 0.1 ⁴⁸⁹ g/l NH₄Cl condition; M_{ACL-KG}, Samples cultured with methanol under 0.1 g/l ⁴⁹⁰ NH₄Cl and 0.3 g/l α -KG;

According to Table 2, with methane as carbon source, when 0.1 g/l 491 ammonium accumulated, the *ala*D and *gcv*T gene expression levels were 492 30.84 and 19.08 times higher than the control, while the genes glnA and 493 *glt*B which responsible for the assimilation of NH₄⁺ with the C₅ substrate 494 expressed 1.92 and 7.48 times higher than that of the control. This 495 indicated that ZR1 may rely more on the C₃ carbon skeleton to assimilate 496 the extra NH₄⁺. Another interesting phenomenon is the high expression 497 level of gcvT gene (19.08) when ZR1 confronted with 0.1g/l of NH₄Cl 498 with methane as carbon source and its expression level returned to 2.96 499

when the C5 carbon skeleton was replenished. According the third 500 direction of nitrogen assimilation, this reaction consumes much 501 5,10-methylene-5,6,7,8-tetrahydrofolate which is a derivative of C1 502 substrate. And the exhaustion of the C1 and C3 substrate may reduce the 503 flux distribution of the downstream carbon metabolic pathway and cause 504 the worse growth of ZR1 when high concentration ammonium appeared. 505 Furthermore, NH₄⁺ accumulation may also induce the expression of the 506 hps gene which is responsible for the assimilation of C1 substrate. 507 Surprisingly, expression level of genes concerns the nitrogen assimilation 508 and formaldehyde oxidization was also up-regulated 4-8 times. On the 509 other hand, with methanol as carbon sources, ZR1 possess a relative high 510 expression background with the *glt*B, *ala*D, *hps* and *cs* genes. And genes 511 changed slightly with NH₄⁺ accumulation in comparison with control. 512 However, if considering the relative gene expression level of the samples 513 with methanol as carbon sources in comparison with methane, the C5 and 514 C3 carbon skeleton assimilation gene gltB and alaD of ZR1 still kept 515 19.07 and 12.03 times higher than the methane control. Thus, when 516 confronted with high ammonium concentration, ZR1 may mainly 517 assimilate NH₄⁺ through pathway 1 and 2. Meanwhile, decreasing the *nir* 518 and *nif* gene expression level to decrease the formulation of ammonium 519 might be the main strategy employed by ZR1. With the replenishment of 520 the C5 skeleton, the expression level of ammonium assimilation gene 521

*glt*B and *ala*D using C5 and C3 carbon skeleton decreased to 0.12 and
0.08, and the nitrate and nitrogen reduction gene return to normal level.
However, the *gcv*T gene in the third ammonium assimilation direction
expressed 2 times higher.

Nitrate and its metabolic intermediate were found to have multiple 526 functions for the primary metabolism in biology (Stitt, 1999; Commichau 527 et al., 2006; Cueto et al., 2016). Ammonium was found to be an important 528 signal molecular to affect the methane and nitrogen metabolism in 529 methanotrophs (Dam et al., 2014; Bodelier and Laanbroek, 2004). In this 530 study, it was also found that, owing to the imbalance metabolism of 531 carbon and nitrogen source, ammonium would accumulated to 532 congcentrations high enough to inhibited cell growth. High concentration 533 ammonium will result in high level expression of several genes concerned 534 the carbon and nitrogen metabolism. 535

536 **Conclusions**:

Nitrate and its metabolic intermediate were found to have multiple functions for the primary metabolism in biology (Stitt, 1999; Commichau et al., 2006; Cueto-Rojas et al., 2016). Ammonium was found to be an important signal molecular to affect the methane and nitrogen metabolism in methanotrophs (Dam et al., 2014; Bodelier and Laanbroek, 2004). In this study, it was also found that, owing to the imbalance metabolism of carbon and nitrogen source, ammonium would accumulated to concentrations high enough to inhibited cell growth. High concentration
ammonium will result in high level expression of several genes concerned
the central carbon and nitrogen metabolism, and thus change the
metabolic mode of cells.

First, effect of nitrogen substrate to the growth of ZR1 from methane and 548 methanol were analyzed. Nitrate salts were proved to be the best nitrogen 549 substrate to support the growth of ZR1 from methanol and methane as 550 carbon source. The nitrate intermediate metabolite ammonium was found 551 to inhibit the growth of ZR1 from methanol and methane. High nitrate 552 concentration inhibition phenomenon has long been investigated in 1991 553 by Park et al. (Park et al., 1991), however its inhibition mechanism in 554 methanotrophs still lack studying. According to the substrate inhibition 555 theory (Muchandani and Luong, 1989), the nitrate metabolic intermediate 556 ammonium were found to accumulate, and might inhibit the growth of 557 ZR. Kinetic study revealed that the concentration of accumulated 558 ammonium was proportional to the original nitrate concentration in the 559 medium. With high initiate nitrate salt, the metabolic mediates NH_{4^+} 560 would accumulate to more than 10 mM which was high enough to inhibit 561 the growth of ZR1. These results indicated that worse growth of ZR1 562 might be owing to the disequilibrium of carbon and nitrogen metabolism. 563 Supplying carbon skeleton to assimilate extra ammonium was proven to 564 be a suitable strategy to relieve the inhibition effect of NH_4^+ . The carbon 565

skeleton replenish effect of several carbon substrate to ammonium 566 inhibition was found to be different with methanol and methane as carbon 567 source. However α -KG was found to be the best carbon skeleton to 568 relieve the ammonium inhibition effect. qPCR analysis indicated that, 569 gcvT, glnA, and alaD genes were expressed relatively higher when ZR1 570 confronted with the high ammonium concentration with methane as 571 carbon source. With addition of the C5 carbon skeleton α -KG, glnA gene 572 expressed higher, and the gcvT gene decreased 8 times lower to 573 ammonium accumulation condition, which indicated that, ZR1 may rely 574 more on the third direction to assimilate extra ammonium. With methanol 575 as carbon source, the *glt*B and *ala*D gene expressed at a relative high 576 level at normal condition in comparison with methane as carbon source. 577 When confronted the ammonium inhibition condition, ZR1 may decrease 578 its nir and nif gene expression level and up-regulate the hps gene to 579 further prevent the accumulation of ammonium. 580

In this study it was found that the nitrogen metabolic intermediate ammonium might be a signature of the nitrogen metabolism in *Methylomonas*, and induce related genes to balance the carbon and nitrogen metabolism. Besides the normal ALAD and GS/GAGOT ammonium assimilation pathway, the third Glycine synthase using the C1 carbon skeleton may also actively expressed in ZR1. Methanotrophs may utilize a complex strategy to balance the carbon and nitrogen metabolism

according to the available carbon source. These findings are meaningful to reveal the complex coordination metabolic mechanism of nitrogen and carbon in methanotrophs. However, the ammonium signal transduction pathway of methanotrophs needs further study to reveal its induction mechanism.

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598 **Conflicts of interest**

599 The authors have declared that no conflicts of interest exist.

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