

The case for a dynamical subsurface ecosystem

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The introduction and concentration of electron donors and acceptors in the subsurface biosphere is controlled by the mixing of subsurface fluids[1], but the mechanisms and rates at which microbial communities respond to changes induced by fluid mixing and transport are relatively unknown. Subsurface microbial ecosystems whose estimated doubling times range from <1 to >3,000 years[2, 3, 4, 5, 6, 7, 8] are often considered to be relatively static. Despite marked changes in geochemistry over a 1-year period[9, 10], the bacterial community inhabiting a 1339 m below land surface (mbls) fracture (Be326) remained largely unchanged[9] and exhibited PLFA isotopic signatures consistent with the accumulation of ¹³C-DIC impacted by the microbial oxidation of CH₄[10]. These CH₄ oxidizing (MO) bacteria and archaea are an essential link between the Be326 subsurface carbon cycle and microbial community[10] and were hypothesized to contain members of the community that are most sensitive to environmental change. To evaluate this hypothesis, we used a combination of high throughput sequence analysis methods (DNA, RNA, and protein) and geochemical monitoring of Be326's in situ fracture fluids over the course of both longer (2.5 year) and shorter (2-week) timescales and validated our findings through a series of ¹³C-CH₄ laboratory enrichment experiments.

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We show that Be326’s MO organisms responded to changes in electron donor and acceptor availability in their natural subsurface habitat and under laboratory conditions over extended periods of time. These results provide the most definitive evidence to date that, like the marine subsurface[4], CH₄ oxidation occurs and is an integral component of the deep terrestrial subsurface carbon cycle. Further, the responsiveness of this component of the microbial community to changes in geochemistry illustrates a more dynamic subsurface ecosystem than previously understood.

Throughout a 2.5-year sampling period at Be326, $\delta^2\text{H}$ and $\delta^{18}\text{O}$ isotopic signatures of the fracture fluid over the 3 sampling points (Figure 1) indicated significant changes in the nature of the fluids over time. In 2011 (the first sampling point), the fluids were dominated by paleometeoric fluids and were consistent with other fluids located approximately 1,000 to 1,500 mbls in the Witwatersrand Formation (South Africa)[1, 11]. In 2012 and 2013, $\delta^{18}\text{O}$ and $\delta^2\text{H}$ signatures were displaced significantly above the Global Meteoric Water Line[12] (GMWL) in a pattern well-established to be consistent with mixing with older more highly altered fluids[13]. Mixing with contamination fluids such as mine service water would have resulted in a different displacement pattern – ruling out the possibility that the geochemical or microbial shifts in the borehole over time were related to contamination. Instead, this pattern indicates an increased component of ancient *in situ* fracture fluids in the borehole over time (Supplementary Discussion).

Geophysical-chemical measurements were made for both the 2.5-year and 2-week time series (Table 1). For the two time series, the mean and variance of temperature, pH, and CH₄ concentration measurements were relatively similar whereas the degree and magnitude to which E_h , SO_4^{2-} , NO_3^- , and H_2 concentrations changed was much greater over the 2.5-year time series. Within the 2.5-year time series, the fracture fluids shifted from a more reducing environment ($2011_{E_h} = -98$ mV, $2011_{[\text{H}_2]} = 130$ nM) with limited electron acceptor availability ($2011_{[\text{Sulph.}]} = 37$ μM , $2011_{[\text{Nitr.}]} = 0.4$ μM) to a more oxidizing ($2013_{E_h} = 21 \pm 28$ mV, $2013_{[\text{H}_2]} = 25$ nM) environment with much greater electron acceptor availability ($2013_{[\text{Sulph.}]} = 479$ μM , $2013_{[\text{Nitr.}]} = 4.5$ μM). The 2-week time series did not exhibit as large of a shift in electron-acceptor availability as the 2.5-year time series and maintained a positive E_h throughout (Table 1).

As the organisms responsible for MO in Be326 were reported to be present at relatively low abundances[10], a targeted assembly pipeline¹ based on the PRICE assembler[14] was implemented to assemble methyl-coenzyme M reductase (*mcrA*) – the protein encoding gene (PEG) responsible for the first step in the anaerobic oxidation of methane (AOM)[15] – and a suite of CH₄ monooxygenases (*mmo*) that are known to play a role in the aerobic oxidation of methane[16]. This pipeline was applied to sets of metagenomic and metatranscriptomic data collected over a 2.5-year period (designated as the 2011, 2012, 2013 samples) and an additional set of metatranscriptomic data that spanned a 2-week

¹<https://github.com/cmagnabosco/OmicPipelines>

2.5-year Time Series	2011	2012	2013
Dates Sampled	Jan 21-27, 2011	July 12-27, 2012	Aug 1-15, 2013
Temperature ($^{\circ}\text{C}$)	36.9	37.3 ± 1.2	35.1 ± 0.8
pH	8.83	8.55	8.17 ± 0.7
E_h (mV)	-98	-27.5 ± 6.4	21 ± 28
SO_4^{2-} (μM)	137	623	479
NO_3^- (μM)	0.4	4.4	4.5
O_2 (Chemet) (μM)	b.d.	9.3	b.d.*
H_2 (nM)	130	9	25
CH_4 (mM)	2.0	0.9	1.1

2-week Time Series	T_0	T_1	T_2
Dates Sampled	Aug 1, 2013	Aug 8, 2013	Aug 15, 2013
Temperature ($^{\circ}\text{C}$)	35.8	35.3	28.6
pH	8.2	8.1	7.9
E_h (mV)	9	53	1
SO_4^{2-} (μM)	531	499	459
NO_3^- (μM)	5.3	5.3	3.7
O_2 (Chemet) (μM)	b.d.*	b.d.*	b.d.*
H_2 (nM)	5.8×10^{-5}	10.4	39.8
CH_4 (mM)	0.8	1.2	1.0

Table 1: Geophysical-chemical measurements from the 2.5-year and 2-week time series. b.d. $\leq 0.31 \mu\text{M}$; b.d.* $\leq 1.6 \mu\text{M}$

period in 2013 (designated as the T_0 , T_1 , T_2 samples). Notably, *mcrA* was selected as an indicator for anaerobic methane oxidizers (ANME) because, phylogenetically, it corresponds with the 16S phylogeny of ANME (and methanogens) and their corresponding AOM (methanogenic) pathways[17]. Metaproteomic data were generated and mapped to the collection of assembled PEGs and a database of other known McrA and MMO peptide sequences (Supplementary Data 1) to confirm the presence of assembled PEGs. The relative abundances of 16S rRNA transcripts were also calculated from the total metatranscriptomic data as described in the Supplementary Methods.

Following targeted assembly and annotation, two complete *mcrA* PEGs related to ANME-1 and “*Ca. Methanoperedens*” (ANME-2d) were assembled while only one complete *mmo* PEG closely related to *Methylococcus capsulatus* was recovered from the metagenomic and metatranscriptomic data (Figure 2, Supplementary Data 2). Partial *mcrA* PEGs related to Methanomicrobia and Methanobacteria were also identified in the high-throughput data (Figure 2, Supplementary Data 2), but partially assembled *mmo*-related PEGs were omitted in downstream analyses due to the difficulty in distinguishing *mmo* from

homologous ammonia monooxygenase PEGs[18]. The predicted proteins (Supplementary Data 3) from the assembled ANME-1, “*Ca. Methanoperedens*”, and *Methylococcus* PEGs were all identified within the metaproteomic data (Supplementary Data 4) and further confirm the presence and activity of these groups of organisms.

The relative abundance (based on coverage) of each PEG was calculated using Bowtie2[19] (-very-sensitive-local; Supplementary Methods) and indicated that the dominant members of the MO community in the metagenomes and metatranscriptomes changed over the 2.5-year time series (Figure 2) but remained constant during the 2-week time series (Table 2). These observations were consistent with the relative changes in geochemistry over both time scales (Table 1). In fact, the relative transcript abundances of MO members throughout the long-term time series were tightly correlated to changes in geochemistry (Tables 3, 4). In particular, ANME-1 transcript abundances were positively correlated to CH₄ ($R^2_{16S} = 0.96$, $R^2_{mcrA} = 0.99$) and H₂ ($R^2_{16S} = 0.98$, $R^2_{mcrA} = 0.98$) while “*Ca. Methanoperedens*” transcript abundances were positively correlated to E_h ($R^2_{16S} = 0.99$, $R^2_{mcrA} = 0.99$) and pH ($R^2_{16S} = 0.96$, $R^2_{mcrA} = 0.99$). O₂ concentrations in 2011 and 2013 were below detection (Table 1) but detected in 2012 (9.3 μM). Similarly, the aerobic *Methylococcus*-related MO exhibited maximum transcript abundance in 2012 (Figure 1, Supplementary Discussion) and may have been positively correlated to O₂ concentrations over time. “*Ca. Methanoperedens*” was the dominant member ($73.3 \pm 2.8\%$) of the MO community during the 2-week time series (Table 2) when fracture fluids contained high concentrations of SO₄²⁻ (496 ± 36 μM) and NO₃⁻ (4.8 ± 0.9 μM) along with a positive E_h (21 ± 28 mV).

	2011	2012	2013	T ₀ [*]	T ₁ [*]	T ₂ [*]
ANME-1	90.21; 98.71	20.55; 17.93	16.94; 40.45	26.82	29.43	23.70
”Ca. M-peredens”	0.79; 0.00	38.20; 25.48	83.06; 52.34	73.18	70.57	76.10
M-microbia	0.00; 0.00	8.71; 8.87	0.00; 0.00	0.00	0.00	0.00
M-bacteria	0.00; 1.14	0.00; 5.04	0.00; 0.00	0.00	0.00	0.00
M. capsulatus	9.00; 0.15	32.54; 42.67	0.00; 7.21	0.00	0.00	0.21

Table 2: Change in the CH₄ community and expression. This table displays the relative abundance (%) of each taxa’s with respect to their CH₄-related protein encoding gene (*mcrA* or *mmo*) abundance relative to the total methane oxidizer (MO) and methanogen community (DNA abundance; RNA abundance). The * next to T₀, T₁, and T₂ signify that only RNA abundances are reported. “*Ca. M-peredens*”=“*Ca. Methanoperedens*”; M-microbia=Methanomicrobia; M-bacteria=Methanobacteria; M. capsulatus=*Methylococcus capsulatus*.

To better understand the response of the MO community to changes in electron donor and acceptor balance, two sets of ¹³C-CH₄ laboratory enrichment experiments were performed on fracture water collected in 2012 and 2013 (Supplementary Methods). As ANME–

		NO ₃ ⁻	SO ₄ ²⁻	CH ₄	E _h	pH	H ₂
2.5-Year Time-Series	“Ca. Methanoperedens”	0.79	0.76	-0.82	0.99	-0.98	-0.86
	ANME-1	-0.97	-0.96	0.98	-0.91	0.81	0.99
	<i>Methylococcales</i>	0.45	0.49	-0.40	-0.20	0.39	-0.33
2-Week Time-Series	“Ca. Methanoperedens”	0.51	0.84	-0.86	-0.36	0.77	-0.71
	ANME-1	0.55	0.87	-0.83	-0.31	0.80	-0.75
	<i>Methylococcales</i>	0.51	0.84	-0.86	-0.36	0.77	-0.71

Table 3: **Correlation (Pearson’s r) between taxa RNA abundance (determined by 16S) and geochemical parameters.** The R_{16S}^2 values reported in the main text are the square of the values reported above.

		NO ₃ ⁻	SO ₄ ²⁻	CH ₄	E _h	pH	H ₂
2.5-Year Time-Series	“Ca. Methanoperedens”	0.70	0.67	-0.75	0.99	-0.99	-0.79
	ANME-1	-0.99	-0.99	0.99	-0.77	0.63	0.99
	<i>Methylococcales</i>	0.80	0.83	-0.76	0.26	-0.06	-0.72
2-Week Time-Series	“Ca. Methanoperedens”	-0.88	-0.58	-0.47	-0.92	-0.68	0.73
	ANME-1	0.89	0.60	0.46	0.91	0.69	-0.75
	<i>Methylococcales</i>	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.

Table 4: **Correlation (Pearson’s r) between taxa RNA abundance (determined by PEG abundance) and geochemical parameters.** The R_{mcrA}^2 values reported in the main text are the square of the values reported above. n.a. = not applicable; *Correlations of *Methylococcales* and geochemical parameters were not calculated as *Methylococcales*-related *mmo* were below detection during T_0 and T_1 .

1, “Ca. Methanoperedens”, and *Methylococcus* are best described as a SO₄²⁻-dependent ANME[15], NO₃⁻-dependent ANME[20], and aerobic methanotrophs[21], respectively, we designed experiments to test whether or not each MO-lifestyle would respond to an increase in the aforementioned electron acceptor. The first experiment (Experiment 1) was a long-term experiment analyzed over 207² days and contained fracture fluid samples from 2012 and 2013 enriched with either ¹³C-CH₄ (Control A), ¹³C-CH₄+SO₄²⁻ (to stimulate ANME-1), or ¹³C-CH₄+NO₃⁻ (to stimulate “Ca. Methanoperedens”). A second set of 2012 and 2013 fracture fluid enrichments (Experiment 2) were analyzed as a short-term experiment for 43 days. Experiment 2 contained ¹³C-CH₄ treatments of ¹³C-CH₄+paraformaldehyde (4%, v/v) (Control B), ¹³C-CH₄+SO₄²⁻, ¹³C-CH₄+O₂, and ¹³C-CH₄+NO₃⁻ as well as an electron acceptor- and donor- free control (Control C). Unlike the correlations observed

²The long term ¹³C-CH₄+NO₃⁻ enrichments were run for 183 days.

between expression data and geochemical parameters, an increase in the proportion of $^{13}\text{C}\text{-CO}_2$ relative to total CO_2 ($\%^{13}\text{C}\text{-CO}_2$) in the $^{13}\text{CH}_4$ enrichments over time is definitive evidence of $^{13}\text{CH}_4$ oxidation under different conditions.

In Experiment 1, the $^{13}\text{C}\text{-CH}_4\text{+NO}_3^-$ enrichments exhibited the greatest rate of $\%^{13}\text{C}\text{-CO}_2$ production and, in 2012, the rate of $\%^{13}\text{C}\text{-CO}_2$ production ($0.017 \pm 0.005 \ \%^{13}\text{C}\text{-CO}_2 \text{ day}^{-1}$) was found to be significantly greater (paired one-tailed Student t-test; $p = 0.02$) than Control A ($0.004 \pm 0.001 \ \%^{13}\text{C}\text{-CO}_2 \text{ day}^{-1}$) (Figure 3). No enrichments from Experiment 2 exhibited an increase in $\%^{13}\text{C}\text{-CO}_2$ production (Supplementary Data 5). Although ANME-1 was present in the metatranscriptomic and metaproteomic data during the 2012 and 2013 sampling points, there was not a significant difference in $\%^{13}\text{C}\text{-CO}_2$ production between Experiment 1's $^{13}\text{C}\text{-CH}_4\text{+SO}_4^{2-}$ enrichments and Control A (Figure 3). This similarity may be best explained by either SO_4^{2-} driven AOM in Control A or trace CH_4 oxidation during methanogenesis[22]. Notably, the concentrations of SO_4^{2-} in Controls A, B, and C ($[\text{SO}_4^{2-}]_{2012} = 623 \ \mu\text{M}$; $[\text{SO}_4^{2-}]_{2013} = 479 \ \mu\text{M}$) are well within the lower range of SO_4^{2-} concentrations (100-1200 μM) that have been reported for SO_4^{2-} coupled AOM[23]. Furthermore, the 2012 Control C samples (Experiment 2) exhibited net CH_4 oxidation ($\Delta[\text{CH}_4] = -0.14 \pm 0.08 \text{ mM}$) suggestive of SO_4^{2-} coupled AOM. On the other hand, the 2013 Control C exhibited net CH_4 production ($\Delta[\text{CH}_4] = 0.11 \pm 0.01 \text{ mM}$); however, the rate of observed methanogenesis is probably too low to be responsible for the elevated rates of $\%^{13}\text{C}\text{-CO}_2$ production that would occur as a result of trace methane oxidation. Therefore, the $\%^{13}\text{C}\text{-CO}_2$ production in the $^{13}\text{C}\text{-CH}_4\text{+SO}_4^{2-}$ enrichments of Experiment 1 are most likely explained by SO_4^{2-} coupled AOM.

The lack of observable changes in the MO community and activity in the 2-week time series and the short-term enrichment experiments (Experiment 2), respectively, suggest that this subsurface system has a slower microbial response to environmental change than other systems[24, 25, 26]. This finding is consistent with the perception that subsurface organisms spend their lives “in the slow lane.” However, the significant increase in the rate of $\%^{13}\text{C}\text{-CO}_2$ production throughout the long-term $^{13}\text{C}\text{-CH}_4\text{+NO}_3^-$ enrichment experiments and the succession of MO organisms in the Be326 time series in coordination with changes in geochemistry illustrate a more dynamic subsurface ecosystem than previously understood.

METHOD SUMMARY

The Be326 borehole is a 57-m, horizontally drilled borehole. Be326 was drilled in 2007 and is located at the 26 level of the Beatrix Gold Mine (28.232288 S, 26.794365 E; Welkom, South Africa). Samples for this study were collected during field trips in 2011, 2012, and 2013. All other methods are described in the Supplement. Metagenomic and metatranscriptomic data are available at NCBI BioProject PRJNA308990. 16S amplicon data are available under NCBI BioProject PRJNA263371.

Acknowledgements

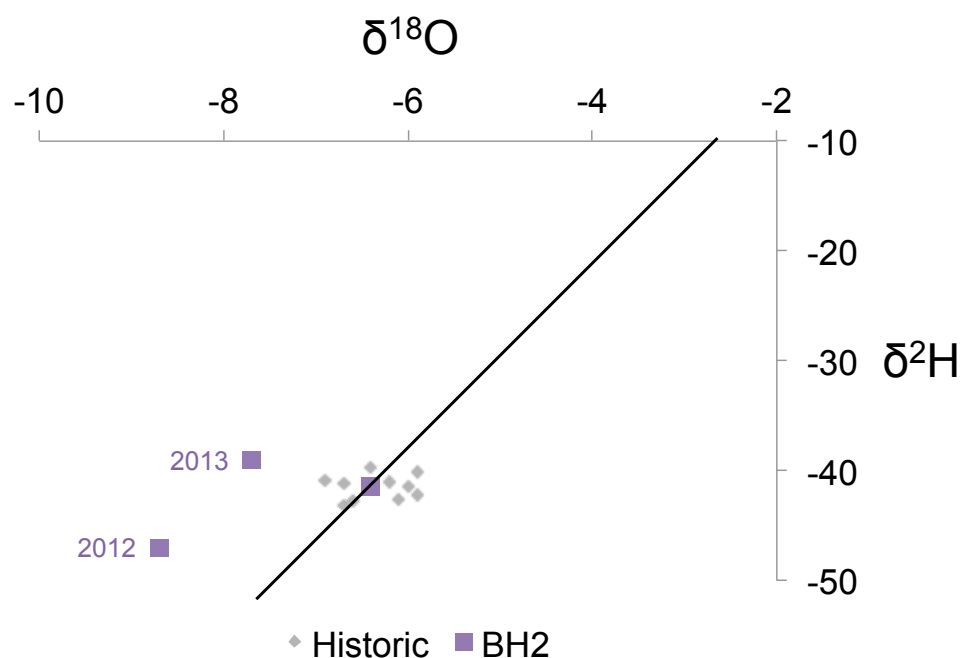
We would like to thank Sibanye Gold Ltd. and the staff of Beatrix Au Mine (Welkom, South Africa) for access and support in collecting subsurface samples. We thank G. Fournier for providing insight and computational support for this project. Subsurface research was funded by the National Geographic Society (BSL), National Science Foundation (NSF) (BSL, TLK, TCO, CM), Center for Dark Energy Biosphere Investigations (CM), and the Natural Sciences and Engineering Research Council of Canada (NSERC) (BSL, GFS). Any opinion, findings, and conclusions or recommendations expressed in this material are those of the authors and do not necessarily reflect the views of the National Science Foundation. Metagenome sequencing was supported by NASA EPSCoR/New Mexico Space Grant Consortium funding to TLK.

Supplementary Material

Supplementary Methods and Discussion along with the listed Supplementary Data are available on the BioArxiv website.

- Supplementary Data 1: Datasets used for protein mapping
- Supplementary Data 2: Assembled *mcrA* and *mmo* PEGs
- Supplementary Data 3: Predicted proteins from assembled *mcrA* and *mmo* PEGs
- Supplementary Data 4: Protein mapping results
- Supplementary Data 5: Enrichment experiment results
- Supplementary Data 6: McrA protein library used to generate phylogenetic tree
- Supplementary Data 7: Results of taxonomic profiling (referred to in Supplementary Discussion)

Figures



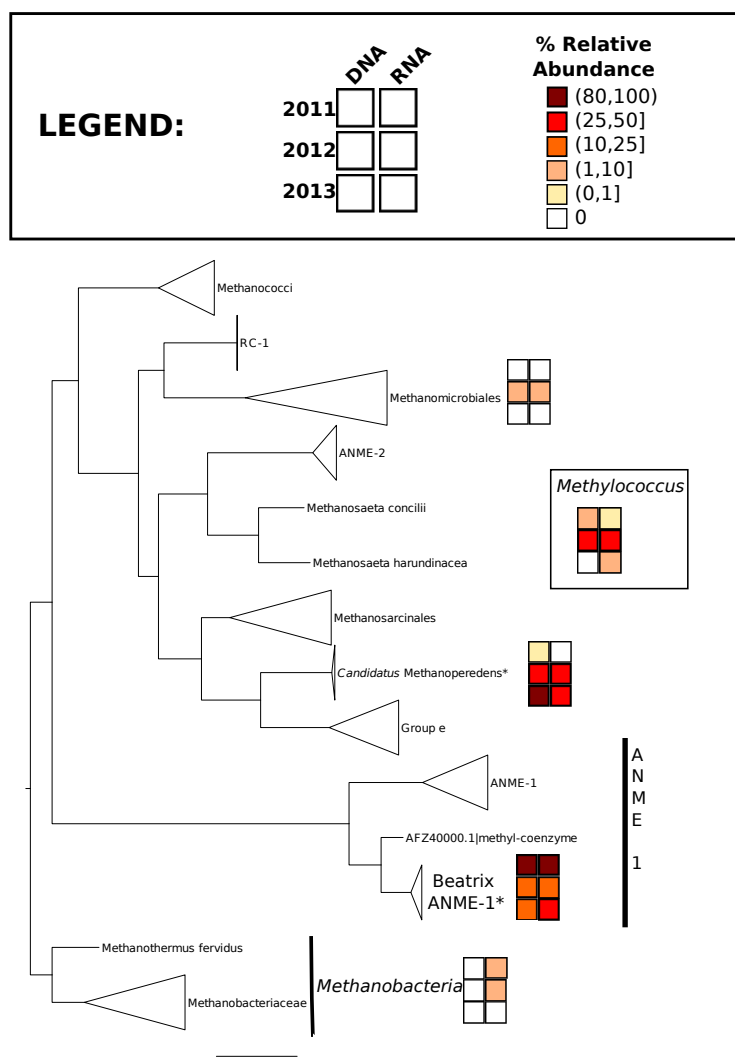


Figure 2: **Change in the CH₄ oxidizing community and activity over time.** A phylogenetic tree was constructed using PhyML[29] from a McrA peptide database (Supplementary Data 6) and the predicted proteins from the mcrA targeted assembly (designated by a *). The predicted MMO from the targeted assembly is also displayed. The 2 × 3 blocks represent the relative abundance of each taxa (DNA, left column; RNA, right column) with respect to their CH₄-related protein of interest (McrA or MMO) over time (row 1, 2011; row 2, 2012; row 3; 2013). Notably, Methanobacteria-related McrA PEGs were selected as the outgroup in this presentation of the McrA phylogenetic tree due to Methanobacteria's placement in species tree of Euryarchaeota.

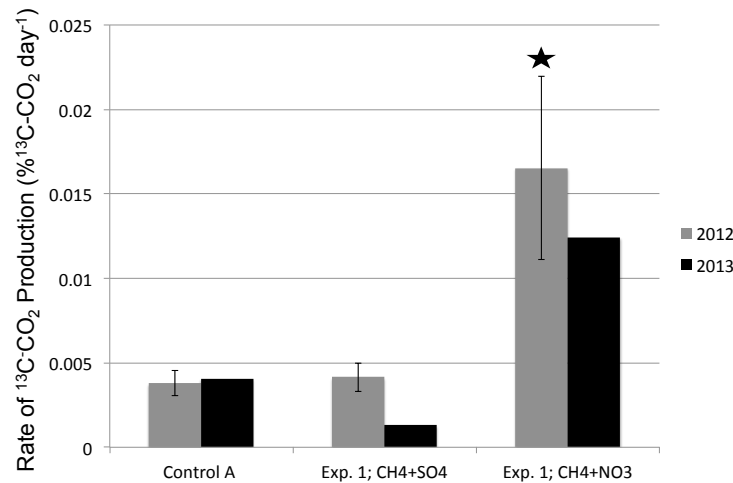


Figure 3: **¹³C-CO₂ production after ¹³C-CH₄ enrichment.** The rates of ¹³C-CO₂ production (%¹³C-CO₂ day⁻¹) from the 2012 (gray) and 2013 (black) fluids of Experiment 1 are displayed. The 2012 samples were run in triplicate and the standard deviations are shown. The star above the 2012 CH₄+NO₃ bar indicates that the rate of ¹³C-CO₂ production in the ¹³C-CH₄+NO₃ enrichment was significantly greater ($p = 0.02$) than the electron acceptor-free control (Control A).

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